High-resolution pulsed field ionization photoelectron-photoion coincidence study of C_2H_2 : Accurate 0 K dissociation threshold for C_2H^+



G. K. Jarvis, Karl-Michael Weitzel, Marcus Malow, Tomas Baer, Y. Song, and C. Y. Ng*d

- ^a Lawrence Berkeley National Laboratory, Chemical Science Division, Berkeley, CA 94720, USA
- ^b Freie Universität Berlin, Institut für Physikalische und Theoretische Chemie, Takustr. 3, 14195 Berlin, Germany
- ^c The University of North Carolina at Chapel Hill, Department of Chemistry, Chapel Hill, NC 27599-3290, USA.
- ^d Ames Laboratory, US Department of Energy and Department of Chemistry, Iowa State University, Ames IA 50011, USA

Received 13th September 1999, Accepted 18th October 1999

By employing the newly developed pulsed field ionization photoelectron-photoion coincidence (PFI-PEPICO) apparatus of the Chemical Dynamics Beamline at the Advanced Light Source, we have examined the formation of ethynyl ion (C_2H^+) from acetylene (C_2H_2) at high resolution. The PFI-PEPICO time-of-flight spectra reveal that fragmentation of C_2H_2 in high-n Rydberg states occurs at energies above the dissociation threshold prior to pulsed field ionization. This study shows that for a prompt dissociation process, the disappearance energy of the parent molecule determined in PFI-PEPICO measurements provides an unambiguous measure of the 0 K ion dissociation threshold. For the formation of C_2H^+ from C_2H_2 this is found to be $17.3576 \pm 0.0010 \ eV$.

Vacuum ultraviolet (VUV) photoionization mass spectrometry and photoelectron spectroscopy are major techniques for providing energetic, 1 spectroscopic, 2,3 and dynamical 4,5 information for cations. The VUV photoelectron-photoion coincidence (PEPICO) method^{4,6} is the combination of these techniques, which involves the detection of correlated photoelectron-photoion pairs. The ion-internal-energy selection achieved in a PEPICO experiment depends critically on the photoelectron detection scheme used. Due to the efficient detection of threshold photoelectrons (TPEs) formed slightly above the ionization threshold,^{2,7} the threshold-PEPICO (TPEPICO) technique has been widely used for the study of state- or energy-selected unimolecular⁴ and bimolecular⁵ reaction dynamics. Despite many successes in the application of the TPEPICO method, its full potential has not been realized due partly to the difficulty in eliminating the contamination of hot electrons. Hot electrons often manifest as a high-energy tail and are the main source of degradation in the TPE resolution. This problem can be partially alleviated by using the electron time-of-flight (TOF) discrimination⁸ and/or the penetration field technique9 in a synchrotron experiment operating in a single or a two-bunch mode. However, due to the nature of the penetration field approach, it is not appropriate for PEPICO studies, especially for ion kinetic energy release measurements. We note that high-resolution TPEPICO studies of rare gas dimers have been reported recently using the penetration field technique. 10 Nevertheless, because of the very short ion flight path used in these studies, the ion TOF resolution was seriously compromised. Pulsed field ionization (PFI)-photoelectron (PFI-PE) spectroscopy,11,12 which relies on the detection of electrons formed at

slightly below the true ionization threshold by PFI of high-n $(n \ge 100)$ Rydberg species, is shown to be free from the hottail problem in previous laser-based studies. Weitzel and Güthe⁸ were the first to demonstrate the potential for the PFI-PE and PFI-PEPICO measurements using synchrotron radiation in a single-bunch operation. Taking advantage of the high-resolution capability of the Chemical Dynamics Beamline at the Advanced Light Source (ALS), we have developed PFI-PE detection schemes for multibunch measurements, ^{13,14} routinely achieving resolutions in the range 1-5 cm⁻¹ (full-width-at-half-maximum, FWHM), as compared with 0.3-4 cm⁻¹ (FWHM) reported in VUV laser studies.^{3,15} Most recently, we have further developed the PFI-PEPICO method for use with both two-bunch and multibunch synchrotron radiation at the ALS, attaining a resolution of 0.6 meV (5 cm⁻¹, FWHM), limited only by the PFI-PE measurement.16

In this communication, we report on the application of this PFI-PEPICO method for the study of the formation of ethynyl ion (C₂H⁺) from acetylene (C₂H₂), C₂H₂ $+ hv \rightarrow C_2H^+ + H^- + e^-$, which is known to be a prompt dissociation process.¹⁷ By analyzing the dissociation due to C₂H₂ in the cold beam, we have obtained the dissociation threshold for C_2H^+ to an accuracy of $\leq 1.0 \text{ meV}$ (8 cm⁻¹). The elimination of the hot-tail in PFI-PE detection has made possible the use of the disappearance energy of the parent C₂H₂ to determine the 0 K dissociation threshold of C₂H⁺. Furthermore, this experiment provides evidence that C₂H₂* fragments into C₂H* + H prior to field ionization, where C₂H₂* (C₂H*) represents C₂H₂ (C₂H) in high-n Rydberg states. The C₂H radical is believed to play a significant role in soot formation in flames¹⁸ and the hydrocarbon balance in planetary atmospheres. 19 For this reason, the accurate establishment of the thermochemical cycle for the C₂H/C₂H⁺ system is important.

All measurements were made using the newly developed PFI-PEPICO apparatus¹⁶ and dispersed multibunch synchrotron radiation from the ALS. In the multibunch operation, the ALS period (656 ns) consists of 256 micro-VUV bunch (bunch width = 50 ps, separation of adjacent bunches = 2 ns) followed by a dark gap (light off period) of 144 ns. In this study, the PFI pulse (height = 7.0 V cm⁻¹, width = 200 ns) was applied approximately 10 ns after the start of the dark gap. Since the dark gap was only 144 ns in duration, some overlap with the light occurs. However, only a small decrease in PFI-PEPICO signal is observed for this overlap and the long pulse

aids in ion extraction, which occurs in a pseudo-continuous fashion. A dc field of 0.2 V cm⁻¹ is maintained across the interaction region to sweep prompt electrons from the ionization region prior to the application of the electric field pulse for Stark ionization. The PFI-PE selection was achieved by employing the electron TOF scheme.¹⁴ The ion PFI-PEPICO TOF spectra were recorded using a multichannel scaler triggered by the detection of an electron. The current setup is sensitive to the ion kinetic energy.¹⁶ Thus, the FWHM of the PFI-PEPICO TOF peak for Ar⁺(²P_{3/2}) observed using an Ar supersonic beam (translational temperature = 20 K) is significantly narrower than that using an Ar effusive beam (temperature = 298 K). The analysis of the Ar TOF peak obtained using a supersonic Ar beam reveals that the thermal background of Ar in the photoionization chamber contributes ≈15% to the experimental Ar sample. 16 This thermal background contribution is roughly consistent with the estimated densities for the molecular beam and thermal background gas sample at the photoionization/photoexcitation region. The photon energy (hv) calibration was achieved using the PFI-PE bands for $Ne^{+(^{2}P_{3/2})}$ and $Ar^{+(^{2}P_{3/2})}$ recorded under the same experimental conditions before and after each scan. 13,14 This calibration procedure assumes that the Stark shift for ionization thresholds of C₂H₂ and the rare gases are identical. On the basis of the measured PFI-PE band for Ar+(2P3/2), we estimate that the ion-energy selection achieved here is ≈ 1.0 meV (FWHM).

We have obtained PFI-PEPICO TOF spectra for C_2H^+ and $C_2H_2^+$ in the $h\nu$ region of 17.2688–17.3959 eV (see Fig. 1), which is near the C_2H^+ dissociation threshold. At $h\nu=17.2688-17.3373$ eV, the parent $C_2H_2^+$ peak dominates

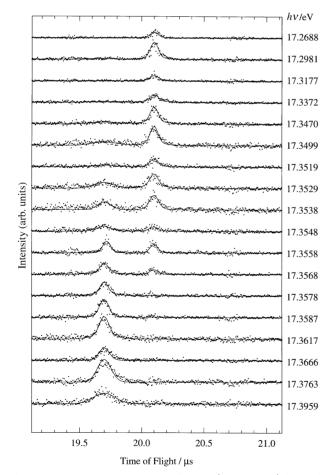


Fig. 1 PFI-PEPICO TOF spectra for C_2H^+ and $C_2H_2^+$ in the $h\nu$ range 17.2688–17.3959 eV. The TOF peaks centered at 19.7 and 20.1 μs are due to C_2H^+ and $C_2H_2^+$, respectively. The $h\nu$ values associated with individual spectra are indicated on the right hand side of the spectra.

in the TOF spectra, whereas at $hv \ge 17.3578$ eV only the daughter C_2H^+ peaks are discernible. Since the internal energy distribution for $C_2H_2^+$ ions is expected to be similar to that of their neutral C₂H₂ precursors, the broad C₂H⁺ peaks observed at $hv \le 17.3548$ eV can be attributed to fragmentation of C₂H₂⁺ formed in the photoionization of thermal (298 K) background C_2H_2 . At hv = 17.3558-17.3617 eV, the C_2H^+ TOF peaks also possess a narrower component, indicating that the main contribution is due to fragmentation of C₂H₂⁺ produced by photoionization of internally cold C2H2 in the molecular beam. We have simulated the individual TOF spectra using two Gaussian functions with FWHMs of 75 and 240 ns to account for the respective contributions due to the cold beam and thermal background C2H2 sample. The kinetic energy release for C₂H⁺ due to fragmentation is expected to be small around the dissociation threshold region because the H atom should carry away most of the energy release. Therefore, to a good approximation, we have treated both daughter C₂H⁺ and parent C₂H₂⁺ similarly in the analysis. However, for TOF spectra at 17.3666, 17.3763, and 17.3959 eV (Fig. 1), it is necessary to increase the FWHMs of the 'cold' Gaussian fits to 96, 130, and 200 ns, respectively, to account for kinetic energy released in fragmentation. The least-squared Gaussian fits (solid lines) to the TOF spectra are shown in Fig. 1.

Since the PFI-PEPICO TOF spectra resolve the dissociation due to cold C₂H₂ from that of thermal C₂H₂, we have analyzed the TOF spectra by taking into account only the 'cold' C₂H⁺ ion signal based on the narrow TOF component. Due to the efficient rotational cooling of C₂H₂ achieved by the supersonic expansion, the 'cold' breakdown curves thus obtained are very sharp as indicated by open and solid squares for the respective fractional abundances of C₂H⁺ and C₂H₂⁺ shown in Fig. 2. These breakdown curves show that the complete dissociation of $C_2H_2^+$ to $C_2H^+ + H$ occurs in an internal energy interval of <5 meV with the cross-over point located at 17.3562 eV. We have also constructed the breakdown curves for C₂H⁺ and C₂H₂⁺ [see open and solid circles, respectively, in Fig. 2] by including the total 'hot' and 'cold' C₂H⁺ signals. As expected, the variation of these breakdown curves is more gradual with a lower cross-over energy at 17.3550 eV, reflecting the nominally higher temperature of the C₂H₂ sample. We have performed simulation of the breakdown curves using similar procedures described previously.²⁰ All calculations were based on rovibronic densities of states taking into account one two-dimensional rotor. The vibrational densities of states were calculated by the Beyer-

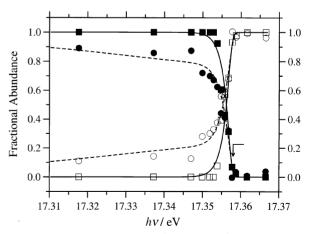


Fig. 2 Breakdown curves of C_2H^+ and $C_2H_2^+$ in the hv range 17.31–17.37 eV. The experimental fractional abundances for C_2H^+ and $C_2H_2^+$ obtained based on the entire daughter ion signal are (\bigcirc) and (\bigcirc), whereas those obtained using only the cold daughter ion signal are indicated as (\square) and (\square), respectively. The lines are simulations curves: (\longrightarrow) 20 K ensemble and (\longrightarrow) 20 K ensemble with 15% thermal background and with an energy dependent loss. See text.

Swinehardt alogorithm²¹ based on known harmonic frequencies.²² By assuming a temperature of 20 K for C₂H₂ in the molecular beam, we have obtained an excellent simulation (solid lines) of the 'cold' breakdown curves, yielding a value of 17.3576 ± 0.0010 eV for the 0 K dissociation threshold of C₂H⁺ from C₂H₂. We note that this value is identical to the disappearance energy for the parent C₂H₂⁺ at 17.3578 eV as marked by the arrow in Fig. 2. The dashed lines in Fig. 2 are calculated breakdown curves assuming 15% thermal background in addition to the 20 K cold beam discussed above. As in the case of the methane dissociation²³ to form CH₃+, in order to obtain good fit to the breakdown curves, which include the dissociation of thermal C2H2, we need to take into account the loss of parent C2H2* due to spontaneous autoionization. Here, we assume that the percentage lost depends on energy, which decreased from 77% at 17.320 eV to 50% at 17.356 eV. Compared to the methane dissociation reaction,²³ the total fraction of parent C₂H₂* molecules lost is smaller and its energy dependence is less pronounced.

As expected, although the shapes and cross-over points of the two sets of break down curves obtained by including and excluding the dissociation of 'hot' C2H2 sample are quite different, the disappearance energies for the parent C₂H₂⁺ are identical. With increasing hv, an increasing part of the energy distribution of the neutral molecules is shifted above the dissociation threshold. The disappearance energy of the parent molecule is the energy at which even the coldest part of the neutral energy distribution reaches above the dissociation threshold. Thus, the disappearance energy of the parent molecule is an intrinsic feature in a PFI breakdown diagram and can be used to give an unambiguous determination of the 0 K fragmentation threshold independent of the internal energy distribution of parent molecules. We emphasize that in order for this feature to serve as a true measure of the ion dissociation threshold, the dissociation reaction must be prompt, i.e., the dissociation lifetime of the excited parent species is much shorter then the time scale $\approx 10^{-7}$ s) of the experiment. Furthermore, it is worth noting that in TPEPICO experiments, where the TPE measurement suffers from the hot-tail problem, the fractional abundance for the parent molecule is not zero at the 'true' dissociation threshold. As a result, the disappearance energy for the parent molecule cannot be used for the determination of the ion dissociation threshold in such

A most interesting observation is that the fraction of fragmentation due to the 'hot' C_2H_2 sample is considerably higher than the estimated 15-25% for the thermal background contribution to the experimental C2H2 sample. This estimation is based on the fractional abundance of 'hot' signal at 17.2688 eV, which is entirely associated with the parent $C_2H_2^+$ and has a value of $\approx 19\%$. The TOF spectra at 17.3529-17.3548 eV, which are slightly lower than the ion dissociation threshold for 'cold' C₂H⁺, reveal a broad TOF peak for C₂H⁺, indicating a 'hot' C₂H⁺ fractional abundance of 40-50%. At photon energies (>17.3578 eV) above the dissociation threshold for 'cold' C₂H⁺, the fractional abundance for the 'hot' signal has a value of $\approx 10-30\%$. In summary, the 'hot' C₂H⁺ signal increases from zero at well below the dissociation threshold to a maximum of $\approx 50\%$ near the dissociation threshold and drops sharply to $\approx 10-30\%$ once the threshold has passed. The observation of this effect can be rationalized by a competition between autoionization and fragmentation of C₂H₂* at energies below and above the dissociation threshold. If the ion core C₂H₂⁺ of C₂H₂* is dissociative with a lifetime shorter than the delay time between the excitation micro-VUV pulse for the formation of C₂H₂* and the electric field pulse for PFI, a viable mechanism for the formation of C_2H^{-} from $C_2H_2^*$ is

$$C_2H_2 + hv \rightarrow C_2H_2^* \rightarrow C_2H^* + H \rightarrow C_2H^+ + e^- + H$$
 (1)

According to this mechanism, C₂H₂* first undergoes prompt dissociation to form $C_2H^* + H$. The subsequent PFI of C_2H^* results in the formation of C_2H^+ and a PFI-PE. The recent lifetime measurements²⁴ of O_2 in high-*n* Rydberg states (O_2^*) converging to dissociative O_2^+ states provide strong support for process (1) as the major process for C₂H⁺ formed in the PFI of C₂H₂*. We note that any high-n Rydberg species, which have spontaneously autoionized before the application of the pulsed electric field, are lost to the PFI detection. The C₂H* radicals formed by fragmentation of C₂H₂* at the dissociation threshold are expected to form below the ionization energy (IE) of C₂H. Thus, autoionization is not readily accessible for C₂H* and is predominantly operative for C₂H₂*, which lies at energies far above the IE of C₂H₂. For C₂H* formed slightly above the dissociation threshold, it can also autoionize. However, the autoionization of this C₂H* may still be less probable than that of C2H2*. The higher than expected PFI-PEPICO intensity for C₂H⁺ from thermal C₂H₂ observed below the dissociation threshold can thus be accounted for by a longer autoionization lifetime for C₂H* than that for C₂H₂*. Since the (15-25%) thermal sample has a large distribution of energies, as the dissociation threshold is approached, the formation of C₂H⁺ from C₂H* will be favored for the thermal C_2H_2 sample over $C_2\bar{H}_2^+$ from the supersonically cooled C₂H₂ sample. Due to the magnification of the C_2H^+ intensity from thermal C_2H_2 , the breakdown diagram will reveal a lower cross-over point if C2H+ ions from both thermal and cold C2H2 are included in the data analysis. This effect of increasing the fragment ion signal relative to the parent ion signal is equivalent to a nominally higher temperature for the parent molecules.

Taking into account the experimental uncertainities, the dissociation threshold (17.3576 \pm 0.0010 eV) for C_2H^+ from C₂H₂ determined here is consistent with values obtained in previous photoion²⁵ and PEPICO studies,^{17,26-28} which fall in the range 17.300-17.360 eV. However, the value obtained here is characterized by a significantly higher accuracy. The IE(C₂H₂) has been determined in a recent PFI-PE study to be 11.4006 ± 0.0006 eV.²⁹ The combination of the latter value and the 0 K dissociation threshold for C_2H^+ from C_2H_2 determined here yields $D_0(H-C_2H^+) = 5.9570 \pm 0.0012$ eV. Although an accurate value for the H-C₂H bond dissociation energy at 0 K $[D_0(H-C_2H) = 5.7125 \pm 0.0010 \text{ eV}]$ and the heat of formation at 0 K ($\Delta_{f_0}H^{\circ}$) for C₂H have been determined in recent high-resolution photodissociation experiments,30 to our knowledge the IE for C2H has not been measured. By combining the 0 K dissociation threshold for process (1) and this D₀(H-C₂H) value, we calculate a value of 11.6451 ± 0.0014 eV for IE(C₂H). Using the well-established $\Delta_{\rm f_0} H^{\circ}$ values¹ for $\rm C_2 H_2$ (235.76 \pm 0.79 kJ mol⁻¹) and H (216.035 \pm 0.006 kJ mol⁻¹), we also obtain $\Delta_{\rm f_0} H^{\circ} (\rm C_2 H^{+}) =$ 1694.44 ± 0.80 kJ mol⁻¹. We note that the accuracy of the latter value is now limited by the $\Delta_{f_0}H^{\circ}(C_2H_2)$. A highresolution PFI-PE study of C₂H is timely to check the consistency of the D₀(H-C₂H) and 0 K dissociation threshold for C₂H⁺. The self-consistency of these values forming the thermochemical cycle will provide confirmation for the accuracy of the experimental $\Delta_{f_0}H^{\circ}(C_2H)$ value.

Acknowledgements

This work was supported by the Director, Office of Energy Research, Office of Basic Energy Sciences, Chemical Science Division of the US Department of Energy under Contract No. W-7405-Eng-82 for the Ames Laboratory and Contract No. DE-AC03-76SF00098 for the Lawrence Berkeley National Laboratory. K.M.W. and M.M. acknowledge financial support by the Deutsche Forschungsgemeinschaft. C.Y.N. acknowledges the support of the Alexander von Humboldt

Senior Scientist Award. Y.S. is the recipient of the 1999 Wall Fellowship at the Iowa State University.

References

- S. G. Lias, J. E. Bartmess, J. L. Holmes, R. D. Levin and W. G. Mallard, J. Phys. Ref. Data, 1988, 17, suppl. 1.
- 2 P.-M. Guyon and T. Baer, in High Resolution Laser Photoionization and Photoelectron Studies, ed. I. Powis, T. Baer and C. Y. Ng, Wiley Series in Ion Chemistry and Physics, Wiley, Chichester, 1995. ch. 1.
- 3 J. W. Hepburn, in Vacuum Ultraviolet Photoionization and Photodissociation of Molecules and Clusters, ed. C. Y. Ng, World Scientific, Singapore, 1991, p. 435.
- 4 T. Baer, J. Booze and K.-M. Weitzel, in *Vacuum Ultraviolet Photoionization and Photodissociation of Molecules and Clusters*, ed. C. Y. Ng, World Scientific, Singapore, 1991, p. 259.
- 5 C. Y. Ng, in Techniques for the Study of Ion-Molecule Reactions, ed. J. M. Farrar and W. H. Saunders, Wiley, New York, 1988, p. 417.
- 6 E. v. Puttkammer, Z. Naturforsch. A, 1970, 25, 1062.
- 7 W. B. Peatman, T. B. Borne and E. W. Schlag, Chem. Phys. Lett., 1969, 3, 492.
- 8 K.-M. Weitzel and F. Güthe, Chem. Phys. Lett., 1996, 251, 295.
- R. I. Hall, A. McConkey, K. Ellis, G. Dawber, L. Avaldi, M. A. MacDonald and G. C. King, Meas. Sci. Technol., 1992, 3, 316.
- Y. Morioka, T. Tanaka, H. Yoshii and T. Hayaishi, J. Chem. Phys., 1998, 109, 1324.
- 11 K. Müller-Dethlefs, M. Sander and E. W. Schlag, Z. Naturforsch. A, 1984, 39, 1089.
- 12 High Resilution Laser Photoionization and Photoelectron Studies, ed. I. Powis, T. Baer and C. Y. Ng, Wiley Series in Ion Chemistry and Physics, Wiley, Chichester, 1995.

- 13 C.-W. Hsu, P. Heimann, M. Evans and C. Y. Ng, Rev. Sci. Instrum., 1997, 68, 1694.
- 14 G. K. Jarvis, Y. Song and C. Y. Ng, Rev. Sci. Instrum., 1999, 70, 2615.
- 15 H. Palm and F. Merkt, Phys. Rev. Lett., 1998, 81, 1385.
- G. K. Jarvis, K.-M. Weitzel, M. Malow, T. Baer, Y. Song and C. Y. Ng, Rev. Sci. Instrum., 1999, 70, 3892.
- 17 K.-M. Weitzel, J. Mähnert and M. Penno, Chem. Phys. Lett., 1994, 224, 371.
- 18 K. H. Homann and H. G. Wagner, Proc. R. Soc. London, Ser. A, 1968, 307, 141.
- 19 R. P. Wayne, J. Geophys. Res., 1993, 98, 13119.
- 20 K.-M. Weitzel, Trends Chem. Phys., Res. Trends, 1997, 6, 143.
- 21 T. Beyer and D. F. Swinehart, Assoc. Comput. Mach., Commun., 1973, 16, 379.
- 22 G. Herzberg, Infrared and Raman Spectra of Polyatomic Molecules, Van Nostrand Reinhold, New York, 1945, p. 290.
- 23 K.-M. Weitzel, M. Malow, G. K. Jarvis, T. Baer, Y. Song and C. Y. Ng, J. Chem. Phys., 1999, 111, in the press.
- 24 M. Evans, S. Stimson, C. Y. Ng and C.-W. Hsu, *J. Chem. Phys.*, 1999, **110**, 315.
- V. H. Diebeler, J. A. Walker and K. E. McCulloh, *J. Chem. Phys.*, 1973, **59**, 2264.
- 26 J. H. D. Eland, Int. J. Mass Spectrom. Ion Phys., 1979, 31, 161.
- 27 K. Norwood and C. Y. Ng, J. Chem. Phys., 1989, 91, 2898.
- 28 Ch. Servais and R. Locht, Chem. Phys. Lett., 1995, 236, 96.
- 29 S. T. Pratt, P. M. Dehmer and J. L. Dehmer, J. Chem. Phys., 1993, 99, 6233.
- D. H. Moraunt and M. N. R. Ashfold, J. Chem. Phys., 1994, 101, 2630; and references therein.

Paper 9/07383C