

Lycopodine-Type Alkaloids from *Lycopodium japonicum*



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Abstract Three new lycopodine-type alkaloids, 4 α -hydroxyanhydrolycodoline (**1**), 4 α ,6 α -dihydroxyanhydrolycodoline (**2**), and 6-*epi*-8 β -acetoxylycoclavine (**3**), and an artifact, lycoposerramine G nitrate (**4**), along with seventeen related known compounds, were isolated from the club moss *Lycopodium japonicum* Thunb. ex Murray (Lycopodiaceae). Their structures were elucidated by extensive spectroscopic methods as well as X-ray analysis. Compounds **1–4** were evaluated for their acetylcholine esterase inhibitory activity.

Keywords *Lycopodium japonicum* · Lycopodine-type alkaloids · 4 α -Hydroxyanhydrolycodoline · 4 α ,6 α -Dihydroxyanhydrolycodoline · 6-*epi*-8 β -Acetoxylycoclavine · Lycoposerramine G nitrate

1 Introduction

Lycopodium alkaloids have attracted great interests of phytochemists and synthetic chemists for a long time due to their complicated structures as well as potent biological activities [1–5]. Till now, more than 300 *Lycopodium* alkaloids have been obtained [6, 7], which were classified into four structural types by chemist Ayer [8], namely, lycopodine-type, lycodine-type, fawcettimine-type, and miscellaneous-type.

Lycopodium japonicum Thunb. ex Murray, abundant in Guangdong, Guangxi, Yunnan, and Guizhou provinces of China, was historically used as a traditional folk medicine for the treatment of contusion, strains, and myasthenia. Its

chemical constituents have been widely investigated and a large number of compounds have been isolated [9–14]. Our previous study on this plant reported a novel *Lycopodium* alkaloid, lycojapodine A [15, 16]. A continuous study on the same plant led to the isolation of three new lycopodine-type alkaloids, 4 α -hydroxyanhydrolycodoline (**1**), 4 α ,6 α -dihydroxyanhydrolycodoline (**2**), and 6-*epi*-8 β -acetoxylycoclavine (**3**), and an artifact, lycoposerramine G nitrate (**4**) (Fig. 1), together with seventeen related known compounds. Compounds **1–4** were tested for their acetylcholine esterase (AChE) inhibitory activity, yet no positive results were observed. Herein, we report the isolation and structural elucidation of these compounds.

2 Results and Discussion

The crude base extract of *L. japonicum* was separated by normal-phase silica gel, RP-18 silica gel, and Sephadex LH-20 chromatography to afford twenty-one lycopodine-type alkaloids, seventeen of which were known ones. The structures of known compounds, compared with literature data, were identified as lycopodine [17], clavolonine [17], alkaloid L-23 [17], lucidioline [18], alkaloid L-20 [18],

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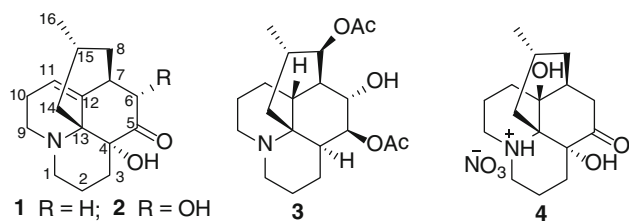


Fig. 1 Compounds **1–4** isolated from *L. japonicum*

lycoposerramine M [18], lycoposerramine G [18], lycoposerramine K [18], lycoposerramine F [18], anhydrolycodoline [19], lycoclavine [20], serratezomine C [21], huperzine E [22], 12-epilycodoline-*N*-oxide [23], diphaldine A [24], 12-deoxyhuperzine O [25], and 8 β -acetoxy-11 α -hydroxylycopodine [26].

Compound **1** was obtained as a colorless crystal. Its molecular formula, C₁₆H₂₃NO₂, was established by

HRESIMS ([M+H]⁺ at *m/z* 262.1806). The ¹H NMR (Table 1) spectrum of **1** displayed one methyl (δ_{H} 0.84, *d*, *J* = 6.2 Hz) and one olefinic proton (δ_{H} 5.65, *d*, *J* = 5.1 Hz). The ¹³C NMR and DEPT spectra of **1** (Table 2) exhibited 16 carbon resonances due to four quaternary carbons (one oxygenated at δ_{C} 75.2, one olefinic at δ_{C} 139.9, and one carbonyl at δ_{C} 210.5), three methines (one olefinic at δ_{C} 118.8), eight methylenes, and one methyl group at δ_{C} 22.1. The ¹H–¹H COSY and HSQC data revealed three partial structures: **a** CH₂CH₂CH₂, **b** CH₂CH₂CH, and **c** CH₂CHCH₂CH(CH₃)CH₂ (Fig. 2). Further detailed 2D NMR analysis indicated compound **1** was closely related to anhydrolycodoline [19]. The only difference was that **1** possessed an OH additional group, which was suggested to be connected to C-4 as inferred from the HMBCs of δ_{H} 2.25 (1H, *d*, *J* = 15.2 Hz, H-6b), 1.58 (1H, *br. d*, *J* = 12.1 Hz, H-2b), and 1.25 (1H, *m*, H-14b) with C-4.

Table 1 ¹H NMR spectroscopic data for **1–4** in CDCl₃; *J* in Hz and δ in ppm

No.	1 ^a	2 ^a	3 ^b	4 ^a
1	2.83 (<i>t</i> , 13.2)	2.83 (<i>m</i>)	3.28 (<i>td</i> , 14.0, 3.2)	3.48 (<i>td</i> , 16.2, 4.6)
	2.42 ^c	2.43 ^c	2.49 (<i>m</i>)	2.89 (<i>dd</i> , 13.0, 4.6)
2	1.89 (<i>m</i>)	1.87 ^c	1.91 (<i>m</i>)	2.22 (<i>m</i>)
	1.58 (<i>br. d</i> , 12.1)	1.57 (<i>m</i>)	1.31 (<i>m</i>)	1.49 (<i>m</i>)
3	1.87 (<i>m</i>)	1.87 ^c	1.68 ^c	1.89 ^c
	1.69 (<i>m</i>)	1.69 (<i>m</i>)	1.37 (<i>m</i>)	1.89 ^c
4			2.72 (<i>m</i>)	
5			4.84 (<i>d</i> , 6.7)	
6	3.12 (<i>dd</i> , 15.2, 6.6)	3.86 (<i>br. s</i>)	3.77 (<i>br. s</i>)	3.26 (<i>dd</i> , 16.1, 5.6)
	2.25 (<i>d</i> , 15.2)			2.19 (<i>d</i> , 16.1)
7	2.76 ^c	2.77 ^c	2.33 (<i>d</i> , 6.5))	2.11 (<i>br. d</i>)
8	1.77 (<i>d</i> , 13.0)	1.81 (<i>m</i>)	4.55 (<i>dd</i> , 11.0, 5.2)	1.96 (<i>m</i>)
	1.23 (<i>m</i>)	1.26 (<i>m</i>)		1.32 (<i>dd</i> , 13.5, 6.1)
9	2.76 ^c	2.77 ^c	3.12 (<i>td</i> , 12.4, 2.5)	4.31 (<i>td</i> , 12.0, 4.6)
	2.56 (<i>br. s</i>)	2.58 (<i>dd</i> , 10.9, 6.1)	2.47 (<i>m</i>)	3.02 ^c
10	2.42 ^c	2.43 ^c	1.68 ^c	2.46 (<i>m</i>)
	1.90 (<i>m</i>)	1.97 (<i>br. d</i> , 17.1)	1.52 (<i>m</i>)	1.32 (<i>m</i>)
11	5.65 (<i>d</i> , 5.1)	5.79 (<i>d</i> , 5.3)	1.88 (<i>m</i>)	3.02 ^c
			1.43 (<i>m</i>)	1.47 ^c
12			2.11 (<i>m</i>)	
14	2.11 (<i>dd</i> , 12.8, 3.5)	2.10 (<i>dd</i> , 12.9, 4.0)	2.64 (<i>dd</i> , 13.2, 6.5)	1.81 ^c
	1.25 (<i>m</i>)	1.29 (<i>m</i>)	0.97 (<i>t</i> , 13.2)	1.81 ^c
15	1.66 (<i>m</i>)	1.52 (<i>m</i>)	2.32 (<i>m</i>)	1.47 ^c
16	0.84 (3H, <i>d</i> , 6.2)	0.83 (3H, <i>d</i> , 6.2)	0.87 (3H, <i>d</i> , 6.2)	0.81 (3H, <i>d</i> , 6.1)
OAc-5			2.04 (3H, <i>s</i>)	
OAc-8			2.00 (3H, <i>s</i>)	

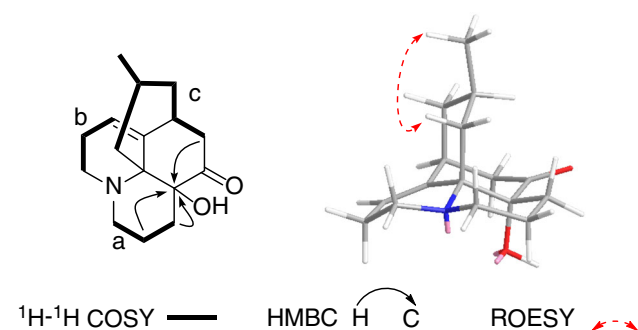
^a Measured on a Bruker DRX-500 MHz

^b Measured on a Bruker AV-400 MHz

^c Overlapped

Table 2 ^{13}C NMR spectroscopic data for **1–4** in CDCl_3 ; J in Hz and δ in ppm

No.	1 ^a	2 ^a	3 ^b	4 ^b
1	46.5 CH ₂	46.3 CH ₂	47.4 CH ₂	46.8 CH ₂
2	20.0 CH ₂	19.7 CH ₂	19.6 CH ₂	14.4 CH ₂
3	23.9 CH ₂	23.9 CH ₂	23.3 CH ₂	24.3 CH ₂
4	75.2 C	77.0 C	27.3 CH	75.8 C
5	210.5 C	207.3 C	76.4 CH	205.9 C
6	43.3 CH ₂	79.5 CH	69.2 CH	39.8 CH ₂
7	40.4 CH	49.3 CH	45.2 CH	38.5 CH
8	44.0 CH ₂	39.4 CH ₂	79.6 CH	41.0 CH ₂
9	45.1 CH ₂	45.0 CH ₂	46.8 CH ₂	50.3 CH ₂
10	25.6 CH ₂	25.9 CH ₂	36.4 CH ₂	17.8 CH ₂
11	118.8 CH	123.2 CH	25.9 CH ₂	29.2 CH ₂
12	139.9 C	144.6 C	42.6 CH	69.6 C
13	63.5 C	64.2 C	54.4 C	65.9 C
14	34.3 CH ₂	33.7 CH ₂	40.8 CH ₂	35.4 CH ₂
15	24.7 CH	24.9 CH	29.7 CH	24.2 CH
16	22.1 CH ₃	22.8 CH ₃	19.6 CH ₃	22.1 CH ₃
OAc-5			170.6 C	
			21.1 CH ₃	
OAc-8			170.6 C	
			21.1 CH ₃	

^a Measured on a Bruker DRX-500 MHz^b Measured on a Bruker AV-400 MHz**Fig. 2** Key 2D NMR correlations of compound **1**

In the ROESY spectrum of **1**, the correlation of H-14a with Me-16 was observed (Fig. 2). However, due to overlapped signals of H-1b with H-10a and H-9a with H-7, the ROESY spectrum could not provide more sufficient information to elucidate the stereochemistry of **1**. The relative configuration of **1** was established by X-ray analysis (Fig. 3), which validated the α -orientation of OH-4, H-7, and Me-16. Therefore, the structure of compound **1** was established as 4 α -hydroxyanhydrolycodoline.

Compound **2** was isolated a colorless crystal. The HRESIMS displayed an $[\text{M}+\text{H}]^+$ peak at m/z 278.1748

(corresponding to a molecular formula $\text{C}_{16}\text{H}_{23}\text{NO}_3$), 16 mass unit higher than that of **1**. Comparison of the 1D NMR data (Tables 1 and 2) with those of **1**, compound **2** was readily identified as 6-hydroxy derivative of **1** as deduced from the HMBCs of δ_{H} 3.86 (1H, *br. s*, H-6) with δ_{C} 39.4 (*t*, C-8), 77.0 (*s*, C-4), and 207.3 (*s*, C-5). The relative configuration of **2** was also established by X-ray analysis, which validated the α -orientation of OH-4, OH-6, H-7, and Me-16 (Fig. 3). Thus, the structure of **2** was elucidated as 4 α ,6 α -dihydroxyanhydrolycodoline.

The molecular formula of compound **3** was determined as $\text{C}_{20}\text{H}_{31}\text{NO}_5$ on the basis of its HRESIMS ($[\text{M}+\text{H}]^+$ at m/z 366.2270), indicating 6° of unsaturation. IR absorption bands implied the presence of ketone (1738 cm^{-1}) and OH (3472 cm^{-1}) groups. The ^1H and ^{13}C NMR (Tables 1 and 2) spectra revealed the existence of two OAc groups, seven sp^3 methylenes, seven sp^3 methines (three oxygenated at δ_{C} 69.2, 76.4, and 79.6), one sp^3 quaternary carbon, and one methyl group. The above data indicated that **3** had a similar structure to that of lycoclavine [20], except for the existence of an additional OAc group which was located at C-8 according to the HMBCs of δ_{H} 2.11 (1H, *m*, H-12), 2.64 (1H, *dd*, $J = 13.2, 6.5$ Hz, H-14a), and 0.87 (3H, *d*, $J = 6.2$ Hz, Me-16) with δ_{C} 79.6 (*d*, C-8) as well as δ_{H} 4.55 (1H, *dd*, $J = 11.0, 5.2$ Hz, H-8) with δ_{C} 170.6. To establish the relative configuration, an X-ray experiment was evolved, which suggested the relative configuration of H-4, H-5, H-6, H-8, H-12, and H-15 to be $\alpha, \alpha, \beta, \alpha, \beta,$ and β , respectively (Fig. 4). Thus, the structure of compound **3** was elucidated and named as 6-*epi*-8 β -acetoxylycoclavine.

Compound **4** had a molecular formula of $\text{C}_{16}\text{H}_{25}\text{NO}_3$, the same as that of lycoposerramine G [18], a known compound also isolated this time. The NMR data and detailed 2D analysis indicated the two compounds had the same planar structure. However, according to the ROESY spectrum, the two compounds also possessed the same relative configuration which indicated **4** should be a salt form of lycoposerramine G. Therefore, a X-ray experiment was implemented that confirmed **4** was lycoposerramine G nitrate (Fig. 4), which was produced during the isolation as verified by the TLC (Al_2O_3).

The compounds **1–4** were tested for acetylcholine esterase (AChE) inhibitory activity, yet no positive results were observed.

3 Experimental Section

3.1 General Experimental Procedures

Melting points were obtained on an WRX-4 micro melting point apparatus (Shanghai Yice Instrument Co., Ltd., Shanghai, China). Optical rotations were measured with

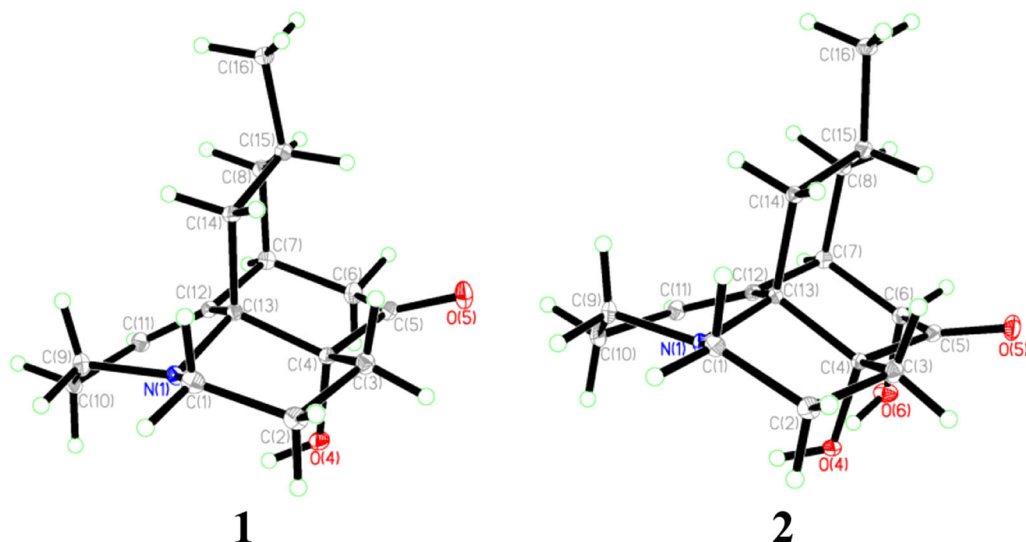
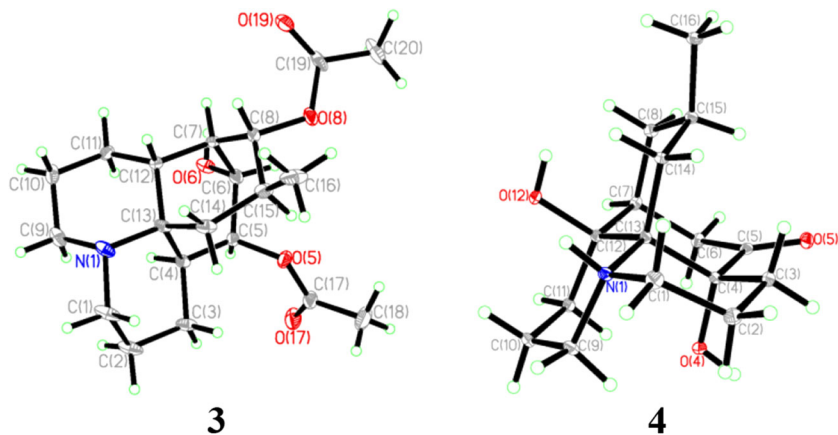


Fig. 3 The X-ray structures of compounds 1–2

Fig. 4 The X-ray structures of compounds 3 and 4



JASCO P-1020 digital polarimeter (JASCO, Tokyo, Japan). UV spectra were obtained using a Shimadzu UV-2401A spectrophotometer (Shimadzu, Kyoto, Japan). A Tenor-27 FT infrared spectrophotometer (Bruker Optics, Ettlingen, Germany) was used for scanning IR spectrum using KBr pellets. ESIMS were recorded on an Agilent 6530 Q-ToF spectrometer (Agilent, Palo Alto, CA, USA). HREIMS were measured using a Waters Auto Premier P776 spectrometer (Waters, Milford, MA, USA). 1D and 2D spectra were run on Bruker AV-400 and DRX-500 spectrometers (Bruker Optics, Ettlingen, Germany). Chemical shifts (δ) were expressed in ppm with reference to the solvent signals. Column chromatography (CC) was performed on Silica gel (200–300 mesh, Qingdao Marine Chemical Ltd., Qingdao, China), RP-18 gel (20–45 μ m, Fuji Silysia Chemical Ltd., Japan), and Sephadex LH-20 (GE healthcare Bio-sciences AB, Sala, Sweden).

Semipreparative HPLC was performed on an Agilent 1100 liquid chromatograph (Agilent, Palo Alto, CA, USA). Liquid chromatograph was equipped with a UV detector (190–400 nm) and a Zorbax SB-C₁₈ (9.4 mm \times 25 cm column, particle size 5 μ m, 1–3 mL/min). Fractions were monitored by Thin-layer chromatography (TLC) (GF₂₅₄, Qingdao Haiyang Chemical Co. Ltd., Qingdao, China), and spots were visualized by heating silica gel plates sprayed with 10 % H₂SO₄ in EtOH or by Dragendorff's reagent.

3.2 Plant Material

The whole plants of *L. japonicum* were collected in Simao of Yunnan Province, People's Republic of China, in August 2006. The sample was identified by Prof. Xiao Cheng at Kunming Institute of Botany, Chinese Academy of Sciences (voucher no. 2006-8-17).

3.3 Extraction and Isolation

Air-dried, powdered sample (50 kg) of *L. japonicum* was dealt as the method reported before to obtain an base extract (67 g) [15]. This extract was subjected to a silica gel column chromatography (CC) with a gradient elution system of petroleum ether–acetone (1:0–0:1) to give 7 fractions (I–VI). Fraction I (7 g) was chromatographed over several silica gel CC eluted with petroleum ether–EtOAc (9:1–1:1) to give three sub-fractions, I-a, I-b, and I-c. I-b was purified by (CHCl₃–MeOH, 1:1) to yield **1** (8 mg), 12-deoxyhuperzine O (14 mg), and huperzine E (7 mg). I-c was repeatedly purified by silica gel CC (petroleum ether–acetone) to afford anhydrolycodoline (40 mg) and lycoposerramine K (27 mg). Fraction II (14 g) was eluted by silica gel CC (petroleum ether–acetone) to afford four sub-fractions, II-a, II-b, II-c, and II-d. II-a repeatedly purified by silica gel CC (petroleum ether–acetone, 8:2) to obtained **2** (7 mg) and lycopodine (14 mg). Alkaloid L-20 was obtained from II-b by recrystallisation. After repeatedly purified by silica gel CC (CHCl₃–acetone) and Sephadex LH-20 (CHCl₃–MeOH, 1:1), II-c give compound clavolonine (11 mg). Fraction III (6 g) was purified by silica gel CC (CHCl₃–MeOH, 9.8:0.2), Sephadex LH-20 CC (CHCl₃–MeOH, 1:1), and semipreparative HPLC (MeOH–H₂O, 85:15) to afford **3** (10 mg) and 8β-acetoxy-11α-hydroxylycopodine (3 mg). Fraction IV (9 g) was subjected to silica gel CC (CHCl₃–MeOH, 9.5:0.5) to give three sub-fractions: IV-a, IV-b, and IV-c. Fraction IV-a was purified by recrystallization and silica gel CC (CHCl₃–MeOH, 9.8:0.2) to afford lycoposerramine M (16 mg), lycoclavine (5 mg), and 12-epilycodoline-*N*-oxide (11 mg). IV-b was subjected to silica gel CC (CHCl₃–MeOH, 9:1) to yield alkaloid L-23 (21 mg) and lycodoline (54 mg). Fraction IV-c was subjected to silica gel CC (CHCl₃–MeOH, 9.5:0.5) and then purified by recrystallization to afford **4** (9 mg) and serratezomine C (8 mg). Fraction-V (11 g) was subjected to silica gel CC (CHCl₃–MeOH, 9:1) and further purified by RP-18 CC (MeOH–H₂O, 4:6–7:3) to yield lucidioline (25 mg), lycoposerramine G (11 mg), and lycoposerramine F (10 mg). Fraction-VI (8 g) was purified by RP-18 CC (MeOH–H₂O, 5:5) to afford diphaldine A (13 mg).

3.4 Acetylcholinesterase Inhibition

Acetylcholinesterase (AChE) inhibitory activity of the compounds **1–4** isolated was assayed by the spectrophotometric method developed by Ellman et al. [27] with slightly modification. *S*-Acetylthiocholine iodide, *S*-butyrylthiocholine iodide, 5,5'-dithio-bis-(2-nitrobenzoic) acid

(DTNB, Ellman's reagent), acetylcholinesterase derived from human erythrocytes were purchased from Sigma Chemical. Compounds were dissolved in DMSO. The reaction mixture (totally 200 μL) containing phosphate buffer (pH 8.0), test compound (50 μM), and acetyl cholinesterase (0.02 U/mL), was incubated for 20 min (30 °C). Then, the reaction was initiated by the addition of 40 μL of solution containing DTNB (0.625 mM) and acetylthiocholine iodide (0.625 mM) for AChE inhibitory activity assay, respectively. The hydrolysis of acetylthiocholine was monitored at 405 nm every 30 s for 1 h. Tacrine was used as positive control with final concentration of 0.333 μM. All the reactions were performed in triplicate. The percentage inhibition was calculated as follows: % inhibition = (E – S)/E × 100 (E is the activity of the enzyme without test compound and S is the activity of enzyme with test compounds).

3.5 4α-Hydroxyanhydrolycodoline (**1**)

Colorless crystal (MeOH); mp 129–130 °C; $[\alpha]_D^{26}$ –163.50 (*c* 0.01, CHCl₃); UV (MeOH) λ_{\max} (log ϵ) 202 (2.85) nm; IR (KBr) ν_{\max} 3362, 2919, 1711 cm⁻¹; ¹H and ¹³C NMR data, see Tables 1 and 2; positive HRESIMS *m/z* 262.1806 (calcd for C₁₆H₂₄NO₂ [M+H]⁺, 262.1807).

3.6 Crystal Data for 4α-Hydroxyanhydrolycodoline (**1**)

C₁₆H₂₃NO₂, *M* = 261.35; orthorhombic, space group *P*₂₁₂₁; *a* = 7.4471 (7) Å, *b* = 9.7363 (9) Å, *c* = 9.2184 (9) Å, α = 90.00, β = 90.6970, γ = 90.00, *V* = 668.35 (11) Å³, *Z* = 2, μ (MoK α) = 0.085 mm⁻¹, crystal dimensions 0.14 × 0.23 × 0.45 mm was used for measurement on a Bruker APEX DUO diffractometer using graphite-monochromated MoK α radiation. The total number of reflections measured was 7187, of which 3461, were observed, *I* > 2 σ (*I*). Final indices: *R*₁ = 0.0326, *wR*₂ = 0.0912. Crystallographic data for the structure of **1** have been deposited in the Cambridge Crystallographic Data Centre (deposition number CCDC 870095). Copies of the data can be obtained free of charge from the CCDC via www.ccdc.cam.ac.uk.

3.7 4α,6α-Dihydroxyanhydrolycodoline (**2**)

Colorless crystal (MeOH); mp 157–158 °C; $[\alpha]_D^{26}$ –91.27 (*c* 0.01, CHCl₃); UV (MeOH) λ_{\max} (log ϵ) 203 (3.05), 264 (3.09) nm; IR (KBr) ν_{\max} 3463, 3431, 2921, 1722 cm⁻¹; ¹H and ¹³C NMR data, see Tables 1 and 2; positive HRESIMS *m/z* 278.1748 (calcd for C₁₆H₂₄NO₃ [M+H]⁺, 278.1756).

3.8 Crystal Data for 4 α ,6 α -Dihydroxyanhydrolycodoline (2)

$C_{16}H_{23}NO_3$, $M = 277.35$; orthorhombic, space group $P2_12_12_1$; $a = 7.7550$ (8) Å, $b = 8.8037$ (9) Å, $c = 10.1722$ (9) Å, $\alpha = 90.00$, $\beta = 98.2380$, $\gamma = 90.00$, $V = 687.32$ (12) Å³, $Z = 2$, $\mu(\text{MoK}\alpha) = 0.092 \text{ mm}^{-1}$, crystal dimensions $0.33 \times 0.40 \times 0.40 \text{ mm}$ was used for measurement on a Bruker APEX APEX DUO diffractometer using graphite-monochromated MoK α radiation. The total number of reflections measured was 7457, of which 3525, were observed, $I > 2\sigma(I)$. Final indices: $R_1 = 0.0335$, $wR_2 = 0.0865$. Crystallographic data for the structure of **2** have been deposited in the Cambridge Crystallographic Data Centre (deposition number CCDC 870093). Copies of the data can be obtained free of charge from the CCDC via www.ccdc.cam.ac.uk.

3.9 6-*epi*-8 β -Acetoxylycoclavine (3)

Colorless crystal (MeOH); mp 170–171 °C; $[\alpha]_D^{26} + 40.63$ (c 0.01, CHCl_3); UV (MeOH) λ_{max} (log ϵ) 203 (2.89) nm; IR (KBr) ν_{max} 3472, 2937, 1738, 1236, 1030 cm^{-1} ; ^1H and ^{13}C NMR data, see Tables 1 and 2; positive HRESIMS m/z 366.2270 (calcd for $C_{20}H_{32}NO_5$ $[\text{M}+\text{H}]^+$, 366.2280).

3.10 Crystal Data for 6-*epi*-8 β -acetoxylycoclavine (3)

$C_{20}H_{33}NO_6$ ($C_{20}H_{31}NO_5 + \text{H}_2\text{O}$), $M = 383.47$; orthorhombic, space group $P2_12_12_1$; $a = 9.364$ (3) Å, $b = 12.755$ (4) Å, $c = 9.528$ (3) Å, $\alpha = 90.00$, $\beta = 119.105$, $\gamma = 90.00$, $V = 994.3$ (5) Å³, $Z = 2$, $\mu(\text{MoK}\alpha) = 0.094 \text{ mm}^{-1}$, crystal dimensions $0.05 \times 0.05 \times 0.60 \text{ mm}$ was used for measurement on a Bruker APEX APEX DUO diffractometer using graphite-monochromated MoK α radiation. The total number of reflections measured was 13705, of which 3167, were observed, $I > 2\sigma(I)$. Final indices: $R_1 = 0.0580$, $wR_2 = 0.1206$. Crystallographic data for the structure of **3** have been deposited in the Cambridge Crystallographic Data Centre (deposition number CCDC 970097). Copies of the data can be obtained free of charge from the CCDC via www.ccdc.cam.ac.uk.

3.11 Lycoserramine G Nitrate (4)

Colorless crystal (MeOH); $[\alpha]_D^{26} - 47.21$ (c 0.01, CHCl_3); UV (MeOH) λ_{max} (log ϵ) 202 (1.93) nm; IR (KBr) ν_{max} 3421, 2924, 1720, 1439 cm^{-1} ; ^1H and ^{13}C NMR see Tables 1 and 2; positive HRESIMS m/z 280.1913 (calcd for $C_{16}H_{26}NO_3$ $[\text{M}+\text{H}]^+$, 280.1912).

3.12 Crystal Data for Lycoserramine G Nitrate (4)

$C_{16}H_{28}N_2O_7$ ($C_{16}H_{26}NO_3^+ + \text{NO}_3^- + \text{H}_2\text{O}$), $M = 360.40$; orthorhombic, space group $P2_12_12_1$; $a = 8.5470$ (8) Å, $b = 8.8731$ (8) Å, $c = 21.963$ (2) Å, $\alpha = 90.00$, $\beta = 90.00$, $\gamma = 90.00$, $V = 1665.6$ (3) Å³, $Z = 4$, $\mu(\text{MoK}\alpha) = 0.112 \text{ mm}^{-1}$, crystal dimensions $0.30 \times 0.33 \times 0.90 \text{ mm}$ was used for measurement on a Bruker APEX APEX DUO diffractometer using graphite-monochromated MoK α radiation. The total number of reflections measured was 17689, of which 4580, were observed, $I > 2\sigma(I)$. Final indices: $R_1 = 0.0277$, $wR_2 = 0.0756$. Crystallographic data for the structure of **4** have been deposited in the Cambridge Crystallographic Data Centre (deposition number CCDC 1001519). Copies of the data can be obtained free of charge from the CCDC via www.ccdc.cam.ac.uk.

3.13 Crystal Data for Lycoserramine G

$C_{32}H_{50}N_2O_6$ ($2 \times C_{16}H_{25}NO_3$), $M = 558.74$; orthorhombic, space group $P2_12_12_1$; $a = 8.2885$ (7) Å, $b = 15.8779$ (13) Å, $c = 21.1225$ (17) Å, $\alpha = 90.00$, $\beta = 90.00$, $\gamma = 90.00$, $V = 2813.3$ (4) Å³, $Z = 4$, $\mu(\text{MoK}\alpha) = 0.090 \text{ mm}^{-1}$, crystal dimensions $0.29 \times 0.43 \times 0.43 \text{ mm}$ was used for measurement on a Bruker APEX APEX DUO diffractometer using graphite-monochromated MoK α radiation. The total number of reflections measured was 30120, of which 7570, were observed, $I > 2\sigma(I)$. Final indices: $R_1 = 0.0320$, $wR_2 = 0.0837$. Crystallographic data for the structure of lycoserramine G have been deposited in the Cambridge Crystallographic Data Centre (deposition number CCDC 1001520). Copies of the data can be obtained free of charge from the CCDC via www.ccdc.cam.ac.uk.

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Conflict of Interest The authors declare no conflict of interest.

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References

1. N. Sizemore, S.D. Rychnovsky, Org. Lett. **16**, 688–691 (2014)
2. L. Zhao, C. Tsukano, E. Kwon, Y. Takemoto, M. Hirama, Angew. Chem. **52**, 1722–1725 (2013)
3. K. Bishayee, D. Chakraborty, S. Ghosh, N. Boujedaini, A.R. Khuda-Bukhsh, Eur. J. Pharm. **698**, 110–121 (2013)

4. L.B. Dong, X. Gao, F. Liu, J. He, X.D. Wu, Y. Li, Q.S. Zhao, *Org. Lett.* **15**, 3570–3573 (2013)
5. M.G. Vallejo, M.G. Ortega, J.L. Cabrera, A.M. Agnese, *Tetrahedron Lett.* **54**, 5197–5200 (2013)
6. Y. Hirasawa, J. Kobayashi, H. Morita, *Heterocycles* **77**, 679–729 (2009)
7. X.Q. Ma, D.R. Gang, *R. Nat. Prod. Rep.* **21**, 752–772 (2004)
8. W.A. Ayer, *Nat. Prod. Rep.* **8**, 455–663 (1991)
9. X.J. Wang, L. Li, S.S. Yu, S.G. Ma, J. Qu, Y.B. Liu, Y. Li, Y. Wang, W. Tang, *Fitoterapia* **91**, 74–81 (2013)
10. X.J. Wang, L. Li, Y.K. Si, S.S. Yu, S.G. Ma, X.Q. Bao, D. Zhang, J. Qu, Y.B. Liu, Y. Li, *Tetrahedron* **69**, 6234–6240 (2013)
11. X.J. Wang, Y.B. Liu, L. Li, S.S. Yu, H.N. Lv, S.G. Ma, X.Q. Bao, D. Zhang, J. Qu, Y. Li, *Org. Lett.* **14**, 5688–5691 (2012)
12. X.L. Li, Y. Zhao, X. Cheng, L. Tu, L.Y. Peng, G. Xu, Q.S. Zhao, *Helv. Chim. Acta* **89**, 1467–1473 (2006)
13. J. Yan, L. Sun, X. Zhang, M. Qiu, *Heterocycles* **65**, 661–666 (2005)
14. X. Cai, D. Pan, Y. Chen, W. Wu, X. Liu, *Shanghai Yike Daxue Xuebao* **18**, 383–385 (1991)
15. J. He, X.Q. Chen, M.M. Li, Y. Zhao, G. Xu, X. Cheng, L.Y. Peng, M.J. Xie, Y.T. Zheng, Y.P. Wang, Q.S. Zhao, *Org. Lett.* **11**, 1397–1400 (2009)
16. Y.R. Yang, L. Shen, K. Wei, Q.S. Zhao, *J. Org. Chem.* **75**, 1317–1320 (2010)
17. T.T. Nakashima, P.P. Singer, L.M. Browne, W.A. Ayer, *Can. J. Chem.* **53**, 1936–1942 (1975)
18. H. Takayama, K. Katakawa, M. Kitajima, K. Yamaguchi, N. Aimi, *Chem. Pharm. Bull.* **51**, 1163–1169 (2003)
19. W.A. Ayer, B. Altenkirk, S. Valverde-Lopez, B. Douglas, R.F. Raffauf, J.A. Weisbach, *Can. J. Chem.* **46**, 15–20 (1968)
20. D.B. MacLean, *Can. J. Chem.* **41**, 2654–2670 (1963)
21. H. Morita, M. Arisaka, N. Yoshida, J.I. Kobayashi, *J. Org. Chem.* **65**, 6241–6245 (2000)
22. B.D. Wang, J. Wang, H.F. Sun, D.Y. Zhu, *Chin. J. Org. Chem.* **21**, 606–610 (2001)
23. C.H. Tan, D.Y. Zhu, *Helv. Chim. Acta* **87**, 1963–1967 (2004)
24. X.D. Wu, J. He, G. Xu, L.Y. Peng, L.D. Song, Q.S. Zhao, *Acta Bot. Yunnan.* **31**, 93–96 (2009)
25. Y.F. Yang, S.J. Qu, K. Xiao, S.H. Jiang, J.J. Tan, C.H. Tan, D.Y. Zhu, *J. Asian Nat. Prod. Res.* **12**, 1005–1009 (2010)
26. B. Li, W.D. Zhang, Y.R. He, L. Lu, D.Y. Kong, Y.H. Shen, *Chem. Pharm. Bull.* **60**, 1448–1452 (2012)
27. G.L. Ellman, K.D. Courtney, V.J. Andres, R.M. Featherstone, *Biochem. Pharmacol.* **7**, 88–95 (1961)