

Protective effect of selected calcium channel blockers and prednisolone, a phospholipase- A_2 inhibitor, against gentamicin and carbon tetrachlorideinduced nephrotoxicity Human and Experimental Toxicology 2014, Vol. 33(8) 831–846 © The Author(s) 2014 Reprints and permission: sagepub.co.uk/journalsPermissions.nav DOI: 10.1177/0960327113509660 het.sagepub.com



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Abstract

The ameliorative effect of calcium channel blockers (CCBs) and a phospholipase-A₂ inhibitor in drug-/ chemical-induced nephrotoxicity was investigated. Rats were divided into 7 groups of 5 rats in each group. In the gentamicin model, group I rats were pretreated with normal saline (10 ml kg⁻¹), while groups II–VII rats were pretreated with normal saline (10 ml kg⁻¹), ascorbic acid (10 mg kg⁻¹), nifedipine (0.86 mg kg⁻¹), verapamil (4.3 mg kg⁻¹), diltiazem (3.43 mg kg⁻¹), and prednisolone (0.57 mg kg⁻¹), respectively, perorally I h before intraperitoneal (i.p.) injection of gentamicin (40 mg kg⁻¹) for 14 days. In the carbon tetrachloride (CCl₄) model, rats were pretreated with CCBs and prednisolone for 7 days before inducing nephrotoxicity with 20% CCl₄ (1.5 ml kg⁻¹). Rats were thereafter killed and blood and tissue samples were collected for assessments. I.p. injections of gentamicin and CCl₄ caused significant hypernatremia, hypokalemia, hypocalcemia, hypophosphatemia, and hyperchloremic alkalosis and reduced renal tissue levels of antioxidants. Also, significant reductions in the hemoglobin, packed cell volume, red blood cells, and platelet indices were observed. Pretreatments with nifedipine (0.86 mg kg⁻¹), verapamil (4.3 mg kg⁻¹), diltiazem (3.43 mg kg⁻¹), and prednisolone (0.57 mg kg⁻¹) significantly ameliorated the deleterious effects of gentamicin and CCl₄ possibly via antioxidant and antilipoperoxidation mechanisms. The results obtained in this study suggest potential clinical usefulness of tested CCBs and prednisolone in drug-/chemical-induced nephrotoxicity.

Keywords

Nephroprotective, calcium channel blockers, phospholipase-A2 inhibitor, renal toxicity

Introduction

Drugs, including aminoglycoside antibiotics, nonsteroidal anti-inflammatory drugs, amphotericin B, and angiotensin-converting enzyme inhibitors, have been implicated as causes of nephrotoxicity.¹ Factors such as age, presence of comorbidities, and exposure to scores of drugs and other therapeutic agents could trigger and culminate into drug-induced nephrotoxicity.² The primary function of the renal system is the elimination of waste products derived either from endogenous metabolism or from the metabolism of xenobiotics. The kidneys also play a major role in the process of erythropoiesis.³ The kidney is a common site of chemical toxicity as a result of its constant exposure to chemicals, solutes, and other detrimental materials, toxicants, which it needs to eliminate from the host body. Although the kidneys comprise <1% of the body mass, they receive around 25% of the cardiac

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output, thus significant amounts of exogenous chemicals and their metabolites are delivered to the kidney.⁴ Biotransformation of chemicals to reactive and toxic metabolites is a key feature of nephrotoxicity and many toxicants, including acetaminophen, bromobenzene, chloroform, and carbon tetrachloride (CCl₄), can be activated. Drug-induced nephrotoxicity can be seen in both inpatient and outpatient settings with variable presentations ranging from mild to reversible injury to advanced kidney disease. Manifestations of drug-induced nephrotoxicity include acid-base abnormalities, electrolyte imbalances, urine sediment abnormalities, proteinuria, pyuria, hematuria, and decline in the glomerular filtration rate.¹ The effects used clinically to diagnose and assess kidney damage include increases in urinary glucose, amino acids, or protein, changes in urine volume, osmolarity or pH, blood urea nitrogen (BUN), plasma creatinine, and serum enzymes.⁵

Drug-induced nephrotoxicity is a relatively common complication, and its incidence has been on the increase with the ever-increasing plethora of drugs and the ease of availability of over-the-counter drugs with detrimental effect on the renal function. Drugs cause approximately 20% of community- and hospital-acquired episodes of acute renal failure.^{6,7} Although renal impairment is often reversible if the offending drug is discontinued, the condition can be costly and may require multiple interventions including hospitalization.⁸ In the acute care hospital setting, drug-induced nephrotoxicity has been implicated in 8-60% of all cases of inhospital acute kidney injury and is a recognized source of significant morbidity and mortality.⁹ Inhospital drug use may contribute to 35% of all cases of acute tubular necrosis, most cases of allergic interstitial nephritis, as well as nephrotoxicity due to alterations in renal hemodynamics and postrenal obstruction.¹⁰ The incidence of antibiotic-induced nephrotoxicity alone may be as high as 36%.¹¹ Thus, there is no gainsaying that drug-induced nephrotoxicity in a clinical setting is prominent and has profound negative impact on the patient, the health-care service provider, and the community at large. One important impact of this is an increase in the hospitalization rate in the hospitals due to drug-induced nephrotoxicity and renal damage in both inpatients and the community.

Despite the high prevalence of drug-induced and drug-related renal diseases among the human population, there still appears not to be effective clinical drugs available to curtail this public human nightmare. This

has necessitated the urgent need to find or develop new drugs in this respect. Various strategies have been known and used to ameliorate or counter the effects of nephrotoxicity whether drug induced or otherwise. Also, studies have been conducted to investigate the nephroprotective properties of different medicinal plants, including Orthosiphon stamineus and Pimpinella tirupatiensis, and pharmaceutical drugs with nephroprotective potentials. The overall objective of these studies is to alleviate nephrotoxic effects by preventing the onset of acute renal failure in the host patient by inhibiting one or more of the several biochemical mechanisms through which renal cell injury occurs. These mechanisms include cell death through apoptosis and oncosis, disruption of cell volume/ion homeostasis, changes in cell membrane integrity/polarity, mitochondrial adenosine triphosphate dysfunction, and disruption in cellular calcium (Ca^{2+}) homeostasis.⁵ Other biochemical mechanisms of renal cell injury are mediated through the enzyme systems that include phospholipases, endonucleases, proteinases, and signaling kinases.⁵

 Ca^{2+} is a second messenger that plays a critical role in a variety of cellular functions. Its distribution within renal cells is known to be complex and involves binding to anionic sites on macromolecules and compartmentalization within subcellular organelles. However, because the proximal renal tubular cells reabsorb approximately 50–60% of the filtered load of Ca^{2+} , they must maintain low cytosolic Ca²⁺ concentrations during a large Ca²⁺ flux. Sustained elevations or abnormally large increase in free cytosolic Ca^{2+} can exert a number of detrimental effects on the renal cells. An increase in free cytosolic Ca²⁺ can activate a number of degradative Ca²⁺-dependent enzymes such as phospholipases and proteinases, which can produce an aberration in the structure and function of cytoskeletal elements.⁵ While the precise role of Ca^{2+} in toxicantinduced injury remains unclear, release of endoplasmic reticulum (ER) Ca²⁺ stores may be a key step in initiating the injury process and increasing cytosolic-free Ca^{2+} concentrations.¹² However, prior depletion of ER Ca²⁺ stores protect renal proximal tubules from extracellular Ca^{2+} influx and cell death produced by mitochondrial inhibition and hypoxia.¹³ Ca²⁺ channel blockers (CCBs) are known pharmacologically to inhibit the influx of Ca^{2+} into cells thereby depleting the Ca^{2+} stores in the renal cells, thus potentially exerting nephroprotective effect.

Phopholipase A_2 (PLA₂) consists of a family of enzymes that hydrolyze the acyl bond of phospholipids

resulting in the release of arachidonic acid and lysophospholipid upon activation during inflammation. PLA₂ activation has been suggested to play a role in various forms of cell injury through a variety of mechanisms.¹⁴ The enzyme phospholipase is involved in the biochemical mechanism through which renal cell injury is manifested. A supraphysiologic increase in PLA₂ activity could cause an increase in the loss of membrane phospholipids and consequently impair membrane function. It is known that during inflammatory processes, the activity of PLA₂ is increased and thus plays a fundamental role in causing damage to cells during inflammation or cell injury. Inhibition of PLA₂ activity is thought to alleviate or ameliorate renal cell injury in nephrotoxicity.

This study was designed to investigate the ameliorative effect of CCBs (nifedipine, verapamil, and diltiazem), and PLA₂ inhibitor (prednisolone) in gentamicin- and CCl₄-induced renal injury in rats.

Materials and methods

Chemicals and drugs

Chemicals used in the study were of analytical grade and include ascorbic acid, potassium chloride (KCl; 1.15%), 10% formalin, 20% CCl₄ solution in olive oil, and normal saline (Unique Pharmaceuticals, Sango-Ota, Nigeria).

Drugs used include gentamicin (Lek Pharmaceutical and Chemical Company, Veroskova, Slovenia), nifedipine (Salutas Pharma GmBH for Lek Pharmaceutical and Chemical Company, Veroskova, Slovenia), verapamil (Teva UK Limited, West Yorkshire, UK), diltiazem (Generics UK Limited, UK), and prednisolone (Hovid Berhad, Malaysia).

Animals

A total of 84 young adult male Wistar rats aged 4–6 weeks (weighing 150–200 g) were obtained from the Laboratory Animal Center, College of Medicine, University of Lagos, Lagos, Nigeria. The animals were housed in standard metal cages, maintained under standard laboratory conditions (23–25°C, 12 h/12 h light/ dark cycle), and fed with standard rat pellets (Livestock Feeds PLC, Ikeja, Lagos, Nigeria) and water ad libitum. The protocol for this study was in compliance with the requirements of the Experimentation Ethics Committee on Animal Use of the College of Medicine, University of Lagos, Lagos, Nigeria, and the United States National

Academy of Sciences Guide for the Care and Use of Laboratory Animals.¹⁵ The rats were acclimatized for 2 weeks before the commencement of the experimental procedures. Seventy-five rats from the stock procured were used for the experiment after acclimatization.

Experimental design

Gentamicin-induced nephrotoxicity. Rats were randomly divided into 7 groups with 5 rats in each group such that the weight differences within and between groups did not exceed $\pm 20\%$. Group I rats were pretreated with 10 ml kg⁻¹ of normal saline perorally (p.o.) 1 h before intraperitoneal (i.p.) injection of 1 ml kg^{-1} of 0.9% normal saline. The rats in groups II-VII were pretreated with 10 ml kg⁻¹ of normal saline, 10 mg kg⁻¹ of ascorbic acid, 0.86 mg kg⁻¹ of nifedipine, 4.3 mg kg⁻¹ of verapamil, 3.43 mg kg⁻¹ of diltiazem, and 0.57 mg kg^{-1} of prednisolone p.o., respectively, 1 h before i.p. injection of 40 mg kg⁻¹ gentamicin.¹⁶ Treatments were carried out once daily between 07:00 h and 09:00 h for a period of 14 days. Nifedipine, verapamil, diltiazem, and prednisolone were administered at their respective therapeutic doses.

CCl₄-induced nephrotoxicity. Animals were randomly grouped and treated as was the case in the gentamicininduced nephrotoxicity model.¹⁷ Treatments were conducted once daily between 07:00 h and 09:00 h for a period of 6 days. After 1 h of peroral administration on day 6, the rats in group I were injected with 1 ml kg⁻¹ of normal saline, while groups II–VII rats were injected i.p. with 1.5 ml kg⁻¹ of 20% CCl₄ dissolved in olive oil. After 24 h of post-CCl₄ injection, the rats were killed under inhaled diethyl ether.

Collection of blood samples and kidneys. Feed was withdrawn from the evening of the 14th day, but rats still had access to drinking water in respect of the gentamicin-induced nephrotoxicity model. On the 15th day, the rats were weighed and then anesthetized using inhalational diethyl ether. In the CCl_4 model, animals were anesthetized 24 h post- CCl_4 injection. Blood samples for biochemical assays (renal function tests) and hematological assays (full blood counts) were obtained by cardiac puncture and collected into plain sample bottles and ethylenediaminetetraacetic acid-treated bottles, respectively.

After blood collection, the kidneys were harvested from each animal and weighed. One of the kidneys was rinsed in 1.15% KCl solution in order to preserve the oxidative enzyme activities of the kidney before being placed in a clean sample bottle which itself was in an ice-pack filled cooler. This is to prevent the breakdown of the enzymes for kidney function enzyme biomarkers. Thereafter, the oxidative stress markers activity was determined.

Kidney enzymatic and nonenzymatic antioxidant parameters. Superoxide dismutase (SOD) activity was determined by its ability to inhibit the auto-oxidation of epinephrine by the increase in absorbance at 480 nm as described by Sun and Zigman.¹⁸ Catalase (CAT) activity was determined by measuring the decrease in absorbance at 240 nm due to the decomposition of hydrogen peroxide (H₂O₂) in ultraviolet-recording spectrophotometer.¹⁹ Reduced glutathione (GSH) content was estimated according to the method described by Sedlak and Lindsay,²⁰ using 10% trichloroacetic acid, 0.5 ml of Ellman's reagent (5,5'-dithiobis-(2-nitrobenzoic acid)), and 3.0 ml phosphate buffer (0.2 M, pH 8.0), and the absorbance was read at 412 nm.

Serum biochemical parameters. Creatinine was determined by adding an aliquot of 0.5 ml of serum sample to 3.5 ml of picric acid. The mixture was centrifuged for 5 min. Then, 3 ml of the supernatant was taken, and to this, 0.2 ml of 4 N sodium hydroxide was added. The mixture was incubated for 1 min, and the absorbance was read at 520 nm. The concentration of creatinine was determined.²¹ In respect of urea determination, 0.1 ml of serum sample was added into a universal bottle containing 19.9 ml of distilled water and the suspension was shaken very well. Then, 1 ml of the suspension was transferred into a test tube and 1 ml of color reagent was added followed by 1 ml of acid reagent. The mixture was heated in boiling water for 20 min. It was then cooled and the absorbance was read at 520 nm against blank.²² Serum levels of sodium (Na⁺), potassium (K⁺), chloride (Cl⁻), Ca^{2+} , bicarbonate (HCO₃⁻), and phosphate (PO₄²⁻) were determined using the ISE 6000 BYY SFRI spectrophotometer (SFRI, Lieu dit Berganton 33127 Saint Jean d'Illac, France).

Hematological assessment. Full blood count analysis of blood samples was carried out using the SYSMEX KX-21N auto analyzer (Sysmex Corporation, Kobe, Japan).

Histopathology of the kidneys. The remaining pair of kidneys harvested is gently but briskly rinsed in

0.9% normal saline and fixed in 10% formal saline. The method described by Baker and Silverton²³ was employed for processing the kidney histology. The tissues were fixed in 10% formalin for a minimum of 10 days to prevent microbial invasion and keep the tissue in original form. After fixation, the tissues were dehydrated by passage through different ascending concentrations of absolute alcohol (50, 70, 90, and 95%) for 30 min each time. The tissues were left in xylene overnight to allow for wax impregnation. After impregnation, the tissues were embedded in melted wax and allowed to harden. Sections of the tissues were cut from the blocks using a microtome. The cut sections were floated on warm water bath to smoothen out creases. The sections were then lifted onto a clean glass microscope slide and allowed to dry on a hot plate with temperature set below melting point. The tissues were stained with hematoxylin and eosin technique. The stained slides were mounted and examined using a photomicroscope (Model N-400ME, CEL-TECH Diagnostics, Hamburg, Germany). Photographs were taken and examined.

Statistical analysis

Statistical analysis was performed using GraphPad Prism (Graph Pad Software; version 5.0, Graph Pad Software Inc., La Jolla, California, USA.). Data were expressed as mean \pm SD for body weight and relative kidney weight and as mean \pm SEM for biochemical and hematological assays. Data were analyzed using one-way analysis of variance for comparison between the control and treated groups, and post hoc test was conducted using Newman– Keuls–Student's *t* test. Level of statistical significance was determined at p < 0.05.

Results

Effect of nifedipine, verapamil, diltiazem, and prednisolone on the average body and kidney weight of rats

As shown in Table 1, significant increases (p < 0.05) were observed in the body weight of rats across the treatment groups except the prednisolone-treated set of rats. However, there was no significant change in weight of kidneys in comparison of all other treatment groups with the normal and gentamicin control groups.

In the CCl₄ model, significant increases (p < 0.05) were observed in the body weight of rats in the normal

Group	Animal weight before (g)	Animal weight after (g)	Weight of kidneys (per 100 g body weight)
	126.0 <u>+</u> 26.88	158.2 ± 51.12 (25.56%) ^a	I.2 ± 0.45
11	116.2 <u>+</u> 16.78	$152.6 \pm 9.61 (31.33\%)^{a}$	I.0 ± 0
	109.4 <u>+</u> 36.04	$160.2 \pm 33.85 (46.44\%)^{a}$	1.0 <u>+</u> 0
V	119.6 ± 7.47	$156.6 \pm 9.48 (30.94\%)^{a}$	1.2 <u>+</u> 0.45
/	131.8 ± 26.69	$180.6 \pm 33.24 (37.03\%)^{a}$	1.4 <u>+</u> 0.55
/ I	32.6 ± 25.11	$163.6 \pm 27.69 (23.38\%)^{a}$	1.0 ± 0
/ II	110.0 ± 13.78	156.6 ± 12.79 (42.36%)	1.0 <u>+</u> 0

Table 1. Effect of nifedipine, verapamil, diltiazem, and prednisolone on body and kidney weights of gentamicin-treated rats.

I: normal saline + normal saline; II: normal saline + gentamicin; III: ascorbic acid + gentamicin; IV: nifedipine + gentamicin; V: verapamil + gentamicin; VI: diltiazem + gentamicin; VII: prednisolone + gentamicin.

^aRepresents significant increase at p < 0.05 between each group weight before and after the start of treatment with test drugs.

Table 2. Effect of nifedipine, verapamil, diltiazem and prednisolone on body and kidney weight of CCl₄-treated rats.

Group	Animal weight before (g)	Animal weight after (g)	Weight of kidneys (per 100 g body weight)
I	169.00 <u>+</u> 9.78	$192.80 \pm 17.55 (14.08\%)^{a}$	I.4 ± 0.55
II	115.80 <u>+</u> 17.21	$122.00 \pm 12.34 (5.35\%)^{a}$	I.2 ± 0.45
III	98.20 <u>+</u> 10.73	97.00 ± 9.64 (-1.22%)	I.0 ± 0
IV	115.20 <u>+</u> 13.92	116.40 ± 7.89 (1.04%)	I.0 ± 0
V	117.40 <u>+</u> 17.64	122.20 ± 25.13 (4.09%)	I.0 ± 0
VI	56.80 <u>+</u> 4.09	$72.60 \pm 12.51 (27.82\%)^{a}$	I.0 ± 0
VII	127.60 ± 8.34	$138.40 \pm 8.62 (8.46\%)^{a}$	I.2 ± 0.45

 CCI_4 : carbon tetrachloride; I: normal saline + normal saline; II: normal saline + CCI_4 ; III: ascorbic acid + CCI_4 ; IV: nifedipine + CCI_4 ; V: verapamil + CCI_4 ; VI: diltiazem + CCI_4 ; VII: prednisolone + CCI_4 .

^aRepresents significant increase at p < 0.05 between each group weight before and after the start of treatment with test drugs.

control, CCl₄ control, diltiazem- and prednisolonetreated groups only. In respect of weight of kidneys, no significant changes were observed in comparison of all other treatment groups with the normal and CCl₄ control groups (Table 2).

Effect of nifedipine, verapamil, diltiazem, and prednisolone on the serum electrolytes, urea, and creatinine

Repeated single daily i.p. injection with 40 mg kg⁻¹ of gentamicin was associated with significant (p < 0.05) increases in the serum level of Na⁺, Cl⁻, HCO₃⁻, urea, and creatinine when compared with untreated control group (group I). Repeated gentamicin treatment was also associated with significant (p < 0.05) reductions in the serum levels of K⁺, Ca²⁺, and PO₄²⁻ when compared with untreated control group (group I). However, pretreatments with nifedipine, verapamil, diltiazem, and prednisolone significantly (p < 0.05) ameliorated

increase in the serum levels of Na⁺, Cl⁻, HCO₃⁻, urea, and creatinine when compared with gentamicin-onlytreated group (group II) returning the values to levels which are comparable to those of untreated control group (group I). In the same vein, nifedipine, verapamil, diltiazem, and prednisolone pretreatments significantly (p < 0.05) ameliorated reductions in the serum levels of K⁺ and Ca²⁺ when compared with gentamicin-onlytreated group (group II) returning the value to those of untreated control group (group I). While ascorbic acid, verapamil, and diltiazem significantly (p < 0.05) ameliorated reductions in the serum levels of PO₄²⁻, nifedipine and prednisolone did not produce significant (p > 0.05) ameliorative effect on the serum PO₄²⁻ levels (Table 3).

Single i.p. injection with 1.5 ml kg⁻¹ of CCl₄ was associated with significant (p < 0.05) increases in the serum levels of Na⁺, Cl⁻, HCO₃⁻, urea, and creatinine, while causing a significant reduction in the serum levels of K⁺ when compared with untreated control group (group I; Table 4). However, pretreatments with ascorbic

Group	Na ⁺ (mmol I ⁻¹)	Group Na ⁺ (mmol I^{-1}) Cl ⁻ (mmol I^{-1}) HCO ⁻ ₃		Urea (mmol I ⁻¹)	$(mmol \ l^{-1})$ Urea $(mmol \ l^{-1})$ Creatinine $(\mu mol \ l^{-1})$ K ⁺ $(mmol \ l^{-1})$ Ca ²⁺ $(mmol \ l^{-1})$ PO ²⁻ ₄ $(mmol \ l^{-1})$	K ⁺ (mmol I ⁻¹)	Ca^{2+} (mmol I^{-1})	PO_4^{2-} (mmol I ⁻¹)
_	I 39.70 ± 4.I3	98.26 ± 2.20	12.48 ± 1.40	07.60 ± 1.26	65.62 ± 6.77	+1	1.66 ± 0.07	01.32 ± 0.15
=	179.00 ± 4.052^{a}	174.26 ± 4.03^{a}	27.74 ± 1.41^{a}	16.72 ± 0.49^{a}	92.68 ± 2.36^{a}	03.28 ± 0.47 ^c	0.78 ± 0.02^{c}	$00.58 \pm 0.05^{\circ}$
≡	143.00 ± 0.93^{b}	103.60 ± 0.91 ^b	14.58 ± 2.60^{b}	06.24 ± 0.19^{b}	66.12 ± 1.15 ^b	+1	1.78 ± 0.06^{d}	01.28 ± 0.16^d
≥	143.00 ± 0.72 ^b	104.60 ± 1.06 ^b	14.34 ± 1.31^{b}	05.58 ± 0.32^{b}	65.48 ± 2.51^{b}	08.56 ± 0.65^{d}	1.84 ± 0.12^{d}	0.98 ± 0.14
>	143.40	102.40 ± 1.12 ^b	16.57 ± 7.03^{b}	06.10 ± 0.71 ^b	63.52 ± 1.19 ^b	+	1.78 ± 0.06^{d}	1.06 ± 0.07^d
>	142.70	105.90 ± 0.36^{b}	15.44 ± 1.25^{b}	05.86 ± 0.49^{b}	65.62 ± 1.82 ^b	+	1.70 ± 0.00^{d}	+1
I	143.80 ± 1.11 ^b	103.40 ± 0.67^{b}	13.68 ± 1.43 ^b	05.24 ± 0.50^{b}	62.22 ± 2.38^{b}	08.70 ± 0.83^{d}	1.70 ± 0.00^d	00.90 ± 0.10
l: norma	l saline + normal sali	ne; II: normal saline +	- gentamicin; III: ascorbi	ic acid + gentamicin;	: normal saline + normal saline; II: normal saline + gentamicin; III: ascorbic acid + gentamicin; IV: nifedipine + gentamicin; V: verapamil + gentamicin; VI: diltiazem + gentamicin; VII:	cin; V: verapamil + g	entamicin; VI: diltiaze	m + gentamicin; VII:

Table 3. Effect of nifedipine, verapamil, diltiazem, and prednisolone on serum electrolytes, urea, and creatinine in gentamicin-treated rats.

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mal sal	isolon
l: nori	predn

^aRepresents significant increases at p < 0.05 when compared with group I. ^bRepresents significant decreases at p < 0.05 when compared with group II. ^cRepresents significant decreases at p < 0.05 when compared with group I. ^dRepresents significant increases at p < 0.05 when compared with group II.

Group	Na ⁺ (mmol I ⁻¹)	Cl^- (mmol l^{-1})	HCO_3^- (mmol I^{-1})	Urea (mmol I ⁻¹)	$Creatinine (\mu mol I^{-1})$	K^+ (mmol I^{-1})
II III IV V	$\begin{array}{r} 131.00 \ \pm \ 5.85 \\ 177.20 \ \pm \ 8.32^a \\ 106.00 \ \pm \ 9.63^b \\ 121.20 \ \pm \ 3.12^b \\ 120.00 \ \pm \ 3.56^b \\ 129.00 \ \pm \ 18.06^b \end{array}$	$\begin{array}{r} 94.00\ \pm\ 8.25\\ 123.00\ \pm\ 5.54^a\\ 76.20\ \pm\ 3.44^b\\ 84.60\ \pm\ 0.98^b\\ 85.20\ \pm\ 1.66^b\\ 89.60\ \pm\ 4.98^b\\ \end{array}$	$\begin{array}{r} 17.20 \ \pm \ 1.99 \\ 42.40 \ \pm \ 1.03^{a} \\ 13.80 \ \pm \ 1.20^{b} \\ 13.00 \ \pm \ 1.41^{b} \\ 13.80 \ \pm \ 1.46^{b} \\ 15.00 \ \pm \ 0.95^{b} \end{array}$	$\begin{array}{c} 4.56 \ \pm \ 0.15 \\ 15.36 \ \pm \ 0.30^a \\ 05.72 \ \pm \ 0.29^b \\ 05.68 \ \pm \ 0.20^b \\ 05.02 \ \pm \ 0.27^b \\ 07.76 \ \pm \ 0.34^b \end{array}$	$\begin{array}{r} 59.66 \ \pm \ 2.38 \\ 85.54 \ \pm \ 1.55^a \\ 39.46 \ \pm \ 2.46^b \\ 48.20 \ \pm \ 4.26^b \\ 34.14 \ \pm \ 10.67^b \\ 30.26 \ \pm \ 5.71^b \end{array}$	$\begin{array}{c} 12.50 \pm 0.82 \\ 08.34 \pm 0.61^{c} \\ 12.42 \pm 1.19^{d} \\ 11.60 \pm 0.58^{d} \\ 11.52 \pm 0.39^{d} \\ 10.82 \pm 1.54^{d} \end{array}$

Table 4. Effect of nifedipine, verapamil, diltiazem, and prednisolone on serum electrolytes, urea, and creatinine in CCl₄-treated rats.

 CCl_4 : carbon tetrachloride; I: normal saline + normal saline; II: normal saline + CCl_4 ; III: ascorbic acid + CCl_4 ; IV: nifedipine + CCl_4 ; V: verapamil + CCl_4 ; VI: diltiazem + CCl_4 ; VII: prednisolone + CCl_4 .

^aRepresents significant increases at p < 0.05 when compared with group I.

^bRepresents significant decreases at p < 0.05 when compared with group II.

^cRepresents significant decreases at p < 0.05 when compared with group I.

^dRepresents significant increases at p < 0.05 when compared with group II.

acid, nifedipine, verapamil, diltiazem, and prednisolone significantly (p < 0.05) ameliorated increase in the serum levels of Na⁺, Cl⁻, HCO₃⁻, urea, and creatinine when compared with CCl₄-only-treated group (group II) returning the levels that are comparable to those of untreated control group (group I). In the same vein, ascorbic acid, nifedipine, and verapamil pretreatments significantly (p < 0.05) ameliorated reductions in the serum levels of K⁺, while diltiazem and prednisolone caused nonsignificant alterations in serum level of K⁺ when compared with CCl₄-only-treated group (group II) returning it to the level that is comparable with that of untreated control group (group I).

Effect of nifedipine, verapamil, diltiazem, and prednisolone on the renal tissue antioxidant indices

Repeated single daily injection of 40 mg kg⁻¹ of gentamicin to treated rats was associated with significant (p < 0.05) decreases in the tissue concentration of GSH and activities of SOD and CAT when compared with untreated control. However, SOD and CAT activities were significantly (p < 0.05) restored in kidneys of rats pretreated with ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone when compared with gentamicin-only-treated rat kidneys, while GSH levels were significantly (p < 0.05) increased in diltiazemand prednisolone-treated kidneys when compared with gentamicin-only-treated rat kidneys (Table 5).

Single i.p. injection of 1.5 ml kg⁻¹ of CCl₄ to treated rats was associated with significant (p < 0.05) decreases in the tissue concentration of GSH and activities of SOD and CAT when compared with untreated control. However, GSH concentration and SOD and CAT activities were significantly (p < 0.05) restored in kidneys of rats pretreated with ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone with diltiazem producing the most significant restorative effect when compared with CCl₄-only-treated rat kidneys (Table 6).

Effect of nifedipine, verapamil, diltiazem, and prednisolone on the full blood count

Repeated i.p. injection of gentamicin at a single daily dose of 40 mg kg⁻¹ body weight of rats for 14 days was associated with significant (p < 0.05) decreases in hemoglobin (Hb), packed cell volume (PCV), and red blood cell (RBC), while causing significant (p < 0.05) increases in the mean corpuscular volume (MCV) and mean corpuscular Hb (MCH) when compared with gentamicin-only-treated values (group II). However, this repeated treatment caused nonsignificant (p > p)0.05) increases in total white blood cell count (TWBC) and mixed (monocytes, eosinophils, and basophils) percentage (%MXD) and nonsignificant (p > 0.05)decreases in platelet (PLT), lymphocyte percentage (%LYM), and neutrophils percentage (%NEU) when compared with gentamicin-only-treated (group II) values. Pretreatments with ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone significantly (p < 0.05) reversed the deleterious effect of gentamicin treatment on the Hb, PCV, and RBC returning their values to those of untreated control values (group I). However, pretreatment with the aforementioned drugs significantly (p < 0.05) decreased the MCV and MCH values when compared with gentamicin-only-treated (group II) values. Neither gentamicin treatment nor

Group	GSH ($\mu g m g^{-1}$ protein)	SOD (U mg ⁻¹ protein)	CAT (U mg ⁻¹ protein)
 V V V V	$\begin{array}{c} 0.37 \ \pm \ 0.05 \\ 0.07 \ \pm \ 0.02^{a} \\ 0.38 \ \pm \ 0.06 \\ 0.50 \ \pm \ 0.07 \\ 0.54 \ \pm \ 0.21 \\ 0.65 \ \pm \ 0.20^{b} \\ 0.76 \ \pm \ 0.25^{b} \end{array}$	$\begin{array}{r} 5.51 \ \pm \ 0.62 \\ 1.70 \ \pm \ 0.16^{a} \\ 6.25 \ \pm \ 0.35^{b} \\ 11.11 \ \pm \ 1.35^{b} \\ 10.19 \ \pm \ 2.05^{b} \\ 9.47 \ \pm \ 2.70^{b} \\ 8.85 \ + \ 2.21^{b} \end{array}$	$\begin{array}{r} 21.88 \ \pm \ 2.37 \\ 6.42 \ \pm \ 0.81^{a} \\ 55.58 \ \pm \ 2.60^{b} \\ 45.48 \ \pm \ 6.39^{b} \\ 36.57 \ \pm \ 7.03^{b} \\ 33.64 \ \pm \ 8.92^{b} \\ 29.39 \ \pm \ 7.03^{b} \end{array}$

Table 5. Effect of nifedipine, verapamil, diltiazem, and prednisolone on renal tissue antioxidant indices in gentamicintreated rats.

GSH: glutathione; SOD: superoxide dismutase; CAT: catalase; l: normal saline + normal saline; ll: normal saline + gentamicin; ll: ascorbic acid + gentamicin; lV: nifedipine + gentamicin; V: verapamil + gentamicin; VI: diltiazem + gentamicin; VII: prednisolone + gentamicin.

^aRepresents significant decreases at p < 0.05 when compared with group I.

^bRepresents significant increases at p < 0.05 when compared with group II.

Table 6. Effect of nifedipine, verapamil, diltiazem, and prednisolone on renal tissue antioxidant indices in CCl₄-treated rats.

GSH ($\mu g m g^{-1}$ protein)	SOD (U mg ⁻¹ protein)	CAT (U mg ⁻¹ protein)
1.32 ± 0.24	2.89 ± 0.27	${\begin{array}{r}11.66 \pm 1.12 \\ 6.12 \pm 1.03^{\rm a}\end{array}}$
$2.80 \pm 0.04^{\mathrm{b}}$	4.44 \pm 0.55 ^b	$23.02 \pm 3.00^{ ext{b}}$
2.35 ± 0.21 ⁵ 1.91 ± 0.34 ⁵	4.31 ± 0.32 ⁵ 3.55 ± 0.41 ⁶	9. 0 ± .48 ^b 5. 3 ± .88 ^b
3.66 ± 0.51 ^b 2.27 ± 0.28 ^b	7.30 ± 0.91 ^b 4.57 ± 0.46 ^b	27.82 ± 3.09 ^b 16.16 ± 1.77 ^b
	$\begin{array}{r} 1.32 \ \pm \ 0.24 \\ 0.86 \ \pm \ 0.04^{a} \\ 2.80 \ \pm \ 0.04^{b} \\ 2.35 \ \pm \ 0.21^{b} \\ 1.91 \ \pm \ 0.34^{b} \\ 3.66 \ \pm \ 0.51^{b} \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

GSH: glutathione; SOD: superoxide dismutase; CAT: catalase; CCl₄: carbon tetrachloride; I: normal saline + normal saline; II: normal saline + CCl_4 ; III: ascorbic acid + CCl_4 ; IV: nifedipine + CCl_4 ; V: verapamil + CCl_4 ; VI: diltiazem + CCl_4 ; VII: prednisolone + CCl_4 . ^aRepresents significant decreases and increases at p < 0.05 when compared with group I.

^bRepresents significant increases and decreases at p < 0.05 when compared with group II.

oral pretreatment with ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone produced significant (p > 0.05) alterations in the TWBC, %LYM, %NEU, and %MXD values (Table 7).

Effect of nifedipine, verapamil, diltiazem, and prednisolone on the renal histopathology

Figures 1 and 2 depict the renal histopathological changes associated with repeated high dose of gentamicin treatment and the effect of oral pretreatment with nifedipine, verapamil, diltiazem, and prednisolone over 14 days. Repeated i.p. treatment with high-dose gentamicin was associated with patchy glomerular atrophy and mild tubulonephritis (Figure 1(b)) when compared with the control kidney (Figure 1(a)). However, in kidneys of rats pretreated with ascorbic acid, nifedipine,

verapamil, diltiazem, and prednisolone, there were no remarkable histological changes in the renal architecture except for scanty tubular necrosis (Figures 1(c) and (d) and 2(b) to (d)).

Induction of drug-induced nephrotoxicity with 1.5 ml kg^{-1} of 20% CCl₄ in olive oil manifested as hemorrhagic areas with congested vascular channels in renal stroma and glomeruli (Figure 3(b)) when compared with normal renal architecture found in untreated control kidney (Figure 3(a)). Oral pretreatments with ascorbic acid, nifedipine, and verapamil manifested as improvements of some of the histopathological lesions induced by subsequent CCl₄ treatment (Figures 3(c) and (d) and 4(b)). However, pretreatments with diltiazem and prednisolone did not offer protection against the deleterious effect of CCl₄ as there were both renal cortical and medullary

Group	Group Hb (g dl ⁻¹)	PCV (%)	RBC (I ⁻¹)	MCV (fl)	PLT (I ⁻¹)	MCH (pg)	ТWBC (I ⁻¹) LYM (%)	LYM (%)	NEU (%)	MXD (%)
_==2>55	$\begin{array}{c} 10.92 \pm 0.76 \\ 07.32 \pm 0.44^{c} \\ 10.42 \pm 0.28^{d} \\ 11.32 \pm 0.53^{d} \\ 11.60 \pm 0.19^{d} \\ 11.32 \pm 0.26^{d} \\ 10.84 \pm 0.32^{d} \end{array}$	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 06.66 \pm 0.50 & 61.6 \\ 04.08 \pm 0.26^{c} & 71.4 \\ 06.82 \pm 0.23^{d} & 58.6 \\ 07.38 \pm 0.49^{d} & 56.4 \\ 07.28 \pm 0.14^{d} & 57.9 \\ 07.50 \pm 0.03^{d} & 54.4 \\ 06.02 \pm 0.27^{d} & 61.4 \end{array}$	$\begin{array}{c} \textbf{61.62} \pm \textbf{1.12} \\ \textbf{71.41} \pm \textbf{1.80}^{a} \\ \textbf{58.62} \pm \textbf{1.01}^{b} \\ \textbf{56.44} \pm \textbf{1.38}^{b} \\ \textbf{56.44} \pm \textbf{1.38}^{b} \\ \textbf{57.94} \pm \textbf{0.45}^{b} \\ \textbf{54.46} \pm \textbf{1.05}^{b} \\ \textbf{61.48} \pm \textbf{1.21}^{b} \end{array}$	$\begin{array}{c} 328.00 \pm 63.60 \\ 269.20 \pm 17.84^{c} \\ 757.80 \pm 20.60^{d} \\ 664.20 \pm 35.21^{d} \\ 717.62 \pm 86.73^{d} \\ 588.00 \pm 57.35^{d} \\ 720.00 \pm 79.72^{d} \end{array}$	$\begin{array}{l} \textbf{16.40} \pm \textbf{0.76} \\ \textbf{17.96} \pm \textbf{0.26}^{a} \\ \textbf{15.28} \pm \textbf{0.33}^{b} \\ \textbf{15.32} \pm \textbf{0.65}^{b} \\ \textbf{15.32} \pm \textbf{0.65}^{b} \\ \textbf{15.96} \pm \textbf{0.29}^{b} \\ \textbf{15.10} \pm \textbf{0.35}^{b} \\ \textbf{18.12} \pm \textbf{0.96}^{d} \end{array}$	$\begin{array}{c} 20.36 \pm 4.01 \\ 29.92 \pm 2.58 \\ 21.30 \pm 2.42 \\ 20.10 \pm 1.72 \\ 23.56 \pm 2.84 \\ 22.34 \pm 5.94 \\ 16.34 \pm 1.70 \end{array}$	$\begin{array}{c} 85.76 \pm 0.76 \ \\ 82.62 \pm 3.74 \ \\ 81.82 \pm 3.03 \ \\ 81.82 \pm 3.03 \ \\ 85.92 \pm 1.44 \ \\ 79.22 \pm 2.57 \ \\ 80.62 \pm 2.34 \ \end{array}$	$\begin{array}{c} \textbf{9.64} \pm 0.77\\ \textbf{0.892} \pm 1.37\\ \textbf{0.08} \pm 2.16\\ \textbf{1.04} \pm 2.13\\ \textbf{9.50} \pm 1.22\\ \textbf{2.74} \pm 1.80\\ \textbf{2.74} \pm 1.80\\ \textbf{0.26} \pm 2.34\\ \end{array}$	$\begin{array}{c} 04.60 \pm 1.26 \\ 08.46 \pm 2.44 \\ 08.10 \pm 1.41 \\ 06.70 \pm 1.64 \\ 04.58 \pm 0.31 \\ 08.04 \pm 0.86 \\ 09.12 \pm 2.29 \end{array}$
Hb: hem lumphoc + gentan ^a Represe ^b Represe ^c Represe	oglobin; PCV: pacl te; NEU: neutrop nicin; V: verapami nts significant incl nts significant dec nts significant dec nts significant incl nts significant incl	ked cell volume; R1 bhils; MXD: mixed 1 1 + gentamicin; V1: reases at $p < 0.05$:reases at $p < 0.05$:reases at $p < 0.05$	Hb: hemoglobin; PCV: packed cell volume; RBC: red blood cell; MCV: mean lumphocyte; NEU: neutrophils; MXD: mixed (monocytes, eosinophils, and bz $+$ gentamicin; V: verapamil $+$ gentamicin; VI: diltiazem $+$ gentamicin; VII: p ^a Represents significant increases at $\rho < 0.05$ when compared with group I. ^b Represents significant decreases at $\rho < 0.05$ when compared with group II. ^c Represents significant decreases at $\rho < 0.05$ when compared with group II. ^d Represents significant decreases at $\rho < 0.05$ when compared with group II.	MCV: mean corpu: ophils, and basophil micin; VII: predniss <i>i</i> th group I. <i>m</i> ith group I. <i>i</i> th group I. <i>i</i> th group I.	Hb: hemoglobin; PCV: packed cell volume; RBC: red blood cell; MCV: mean corpuscular volume; PLT: platelet; MCH: mean corpuscular hemoglobin; TWBC: total white blood cell count; LYM: lumphocyte; NEU: neutrophils; MXD: mixed (monocytes, eosinophils, and basophils); l: normal saline + normal saline; l1: normal saline + gentamicin; l1: ascorbic acid + gentamicin; lV: nifedipine + gentamicin; V: verapamil + gentamicin; VI: diltitazem + gentamicin; VII: prednisolone + gentamicin. Represents significant increases at <i>p</i> < 0.05 when compared with group I. ^P Represents significant decreases at <i>p</i> < 0.05 when compared with group I. ^R epresents significant decreases at <i>p</i> < 0.05 when compared with group I. ^R epresents significant decreases at <i>p</i> < 0.05 when compared with group I.	latelet; MCH: meal normal saline; II: no	n corpuscular hem ormal saline + gent	oglobin; TWBC: amicin; III: ascorb	total white blood ic acid + gentamic	cell count; LYM: in; IV: nifedipine

Table 7. Effect of nifedipine, verapamil, diltiazem, and prednisolone on hematological parameters in gentamicin-treated rats.

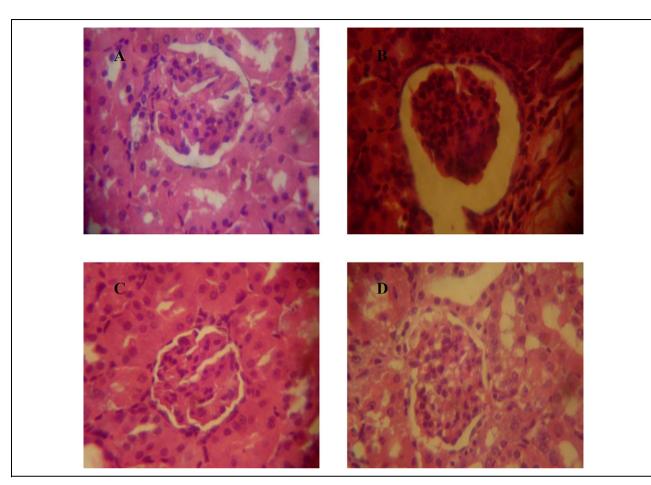


Figure 1. Representative sections: (a) normal kidney showing normal tubular brush borders and intact glomerulus surrounding Bowman's capsule; (b) 40 mg kg⁻¹ day⁻¹ of intraperitoneal gentamicin-treated rat kidney showing glomerular atrophy and significant patchy tubular necrosis indicating mild tubulonephritis; (c) 10 mg kg⁻¹ day⁻¹ of oral ascorbic acid-treated rat kidney showing an intact glomerulus and tubules; (d) 0.86 mg kg⁻¹ day⁻¹ of oral nifedipine-treated rat kidney showing an intact glomerulus and mild patchy tubular necrosis (×400 magnification, hematoxylin and eosin staining).

hemorrhages in the treated kidneys (Figure 4(c) and (d)).

Discussion

Gentamicin is a prototype of a class of antimicrobials known as aminoglycosides, and it is extensively used in the clinical management of both gram positive and gram negative bacterial infections, with preference for the latter.²⁴ However, its use either at high dose or prolonged use, even at a clinical dose, has been reported to be associated with organ toxicities such as ototoxicity and nephrotoxicity.^{25–27} Thus, the use of gentamicin as a standard nephrotoxicant in nephrotoxicity studies is well established.^{16,28} Also, therapeutic doses of gentamicin and other aminoglycoside antibiotics are known as the most common causes of acute renal failure,²⁹ possibly due to increased renal uptake of the antibiotic, mainly by the proximal tubules. The effect of gentamicin on biological membranes appears to be important in its toxicity. It has been proposed that the accumulation of aminoglycosides in proximal tubular epithelial cells leads to membrane structural disturbance and cell death by reactive oxygen species (ROS) involvement.³⁰ ROS produce cellular injury and necrosis via several mechanisms including peroxidation of membrane lipids, protein denaturation, and DNA damage.²⁹ Hydroxyl radical scavengers such as dimethyl thiourea, SOD, and dietary antioxidants (vitamin E, selenium, vitamin C, taurine, and the carotenoids (β -carotene, lutein, and lycopene)) can decrease the gentamicin-induced reduction in the glomerular filtration rate and the severity of the tubular damage.^{31–34}

Nephrotoxicity has been related to a gentamicin trough concentration above 2 μ g ml⁻¹, Ca²⁺ deficiency, Ca²⁺ channel activation, prostaglandins pathways, free radical generation, pyridoxal

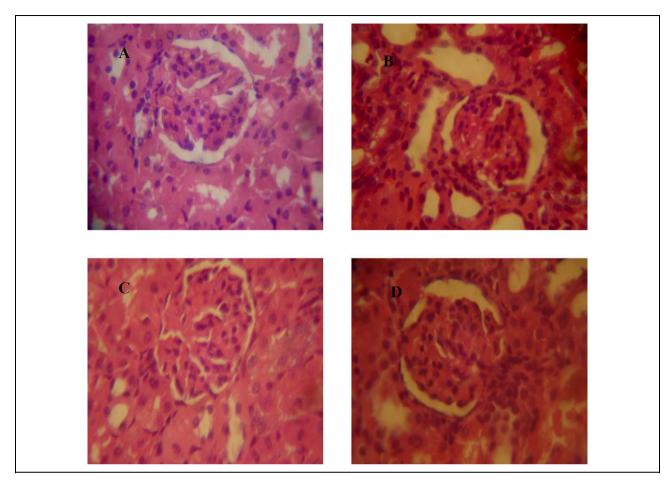


Figure 2. Representative sections: (a) normal kidney showing normal tubular brush borders and intact glomerulus surrounding Bowman's capsule; (b) 4.3 mg kg⁻¹ day⁻¹ of oral verapamil-treated rat kidney showing an intact glomerulus and mild few early tubular necrosis that may not be significant; (c) 3.43 mg kg⁻¹ day⁻¹ of oral diltiazem-treated rat kidney showing an intact and normal glomerulus and significant patchy necrosis of the proximal tubules; and (d) 0.57 mg kg⁻¹ day⁻¹ of oral prednisolone-treated rat kidney showing an intact and normal glomerulus and normal glomerulus and significant patchy necrosis of the proximal tubules; and (d) 0.57 mg kg⁻¹ day⁻¹ of oral prednisolone-treated rat kidney showing an intact and normal glomerulus and very scanty necrosis of the proximal tubules (×400 magnification, hematoxylin and eosin staining).

phosphate deficiency, and ascorbic acid depletion.²⁶ Thus, any strategy or intervention that ameliorates these above-mentioned pathways and deficiencies may ameliorate nephrotoxicity. In this present study, nephrotoxicity was successfully induced through repeated daily i.p. injection of 40 mg kg⁻¹ of gentamicin for 14 days, and this was characterized by hypernatremia, hypokalemia, hypocalcemia, hypophosphatemia, and hyperchloremic alkalosis, ^{35,36} hyperuremia, and hypercreatininemia, which appears to be in agreement with those reported previously.^{37–40} Increased concentration of serum urea and creatinine are considered reliable parameters for assessing the renal functions, particularly, in druginduced nephrotoxicity in animals and human.41 BUN is found in liver protein that is derived from diet or tissue sources and is normally excreted in the

urine. In renal disease, the serum accumulates urea because the rate of serum urea production exceeds the rate of clearance.⁴² Creatinine is mostly derived from endogenous sources by tissue creatinine breakdown.⁴³ Thus, elevation of urea and creatinine levels the serum can be taken as the index of nephrotoxicity.^{41,43,44} In this study, pretreatments with ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone significantly (p < 0.05) improved the serum levels of Na⁺, Cl⁻, K⁺, HCO₃⁻, Ca²⁺, PO₄²⁻, urea, and creatinine returning their levels to about normal values.

Cuzzocrea et al.²⁹ investigated the potential role of the superoxide anion in gentamicin-induced renal toxicity by M40403, a low-molecular-weight synthetic manganese containing a SOD-mimetic that selectively removes superoxide. The authors observed

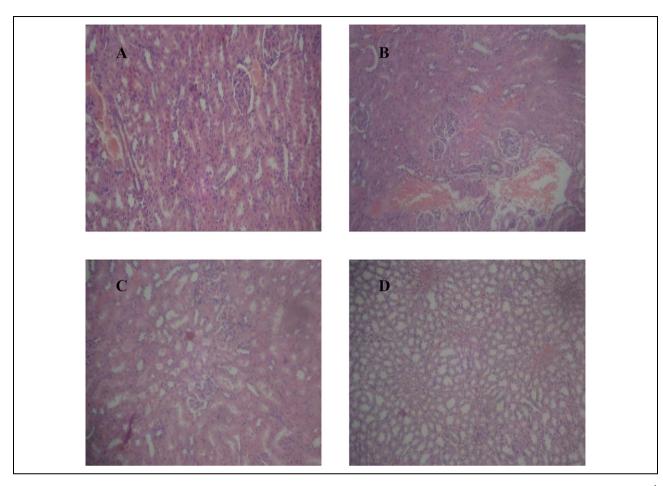


Figure 3. Representative sections: (a) normal rat kidney with regular glomerular and tubular architecture; (b) 1.5 ml kg^{-1} of 20% CCl₄ in olive oil-treated rat kidney showing diffused cortical and medullary hemorrhages with congested vascular channels and tubular necrosis; (c) ascorbic acid-pretreated, CCl₄-treated rat kidney showing areas of focal hemorrhages and mild glomerular capillary congestion; and (d) nifedipine-pretreated, CCl₄-treated rat kidney showing scattered hemorrhages at the medulla and sparing the cortical areas (×100 magnification, hematoxylin and eosin staining). CCl₄: carbon tetrachloride.

that gentamicin treatment was associated with a significant (p < 0.05) increase in kidney myeloperoxidase activity and lipid peroxidation in gentamicintreated rats suggesting a pivotal role of oxidative stress in the etiopathogenesis of gentamicin-induced nephrotoxicity. During kidney injury, superoxide radicals are generated at the site of damage and modulate SOD and CAT resulting in the loss of activity and accumulation of superoxide radical that damages the kidney. SOD and CAT are the most important enzymes involved in ameliorating the effects of oxygen metabolism.^{45,46} The generation of oxidative ROS has been proposed as a mechanism by which many chemicals can induce nephrotoxicity.⁴⁷ Other in vivo antioxidants involved in amelioration of the damaging effects of ROS generation include reduced GSH and GSH peroxidase. In this study, repeated

single daily injection of 40 mg kg^{-1} of gentamicin to treated Wistar rats for 14 days was associated with significant decreases in the renal tissue concentration of GSH and activities of SOD and CAT when compared with untreated control values. With oral pretreatments with nifedipine, verapamil, diltiazem, and prednisolone also for 14 days, SOD and CAT activities were significantly restored to about the values recorded for the standard nephroprotective agent (ascorbic acid) used in this study which mediates its protective effect via antioxidant mechanisms. This strongly suggests that nifedipine, verapamil, diltiazem, and prednisolone were able to significantly prevent decrease in tissue concentration of GSH and activities of SOD and CAT suggesting that protective mechanism of these drugs could possibly be mediated via antioxidant mechanism.⁴⁸⁻⁵⁰ It is important to

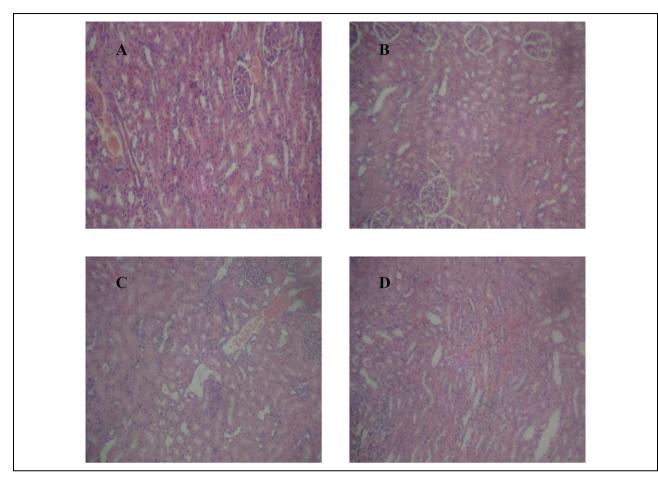


Figure 4. Representative sections: (a) normal rat kidney with regular glomerular and tubular architecture; (b) verapamilpretreated, CCl_4 -treated rat kidney showing few areas of glomerular hemorrhages; (c) diltiazem-pretreated, CCl_4 -treated rat kidney showing areas of cortical and medullary hemorrhages; and (d) prednisolone-pretreated, CCl_4 -treated rat kidney showing diffused areas of cortical and medullary hemorrhages ($\times 100$ magnification, hematoxylin and eosin staining). CCl_4 : carbon tetrachloride.

note that the prooxidant effect of CCBs has been reported on the reproductive system.⁵¹ This suggests that depending on the target cells, CCBs may produce anti-oxidative effect^{49,50} as the results from this study suggests or a prooxidant effect as on the reproductive system.⁵¹

Various in vitro and in vivo studies have shown CCl₄ to enhance lipid peroxidation, reduce renal microsomal nicotinamide adenine dinucleotide phosphate cytochrome P450, and renal reduced/oxidized glutathione ratio (GSH/GSSG) in kidney cortex as well as renal microsomes and mitochondria.^{52,53} CCl₄ is such a widely used environmental toxicant, which is employed to experimentally induce animal models of acute nephrotoxicity and hepatotoxicity.⁵³ CCl₄ is cytochrome P450 metabolized by 2E1 to trichloromethyl radical (CCl_3^-) and its highly reactive trichloromethylproxyl derivative. radical the

(Cl₃COO⁻), which are thought to initiate free radicalmediated lipid peroxidation leading to the accumulation of lipid peroxidation products that cause renal and hepatic toxicities.⁵⁴ These highly reactive radicals are capable of initiating a chain of lipid peroxidation reactions by abstracting hydrogen from polyunsaturated fatty acids (PUFA). Peroxidation of lipids, particularly those containing PUFA, can dramatically change the properties of biological membranes, resulting in severe cellular injury and play a significant role in the pathogenesis of diseases.⁵⁴ This phenomenon results in the generation of ROS (like superoxide anion O_2^- , H_2O_2 , and hydroxyl radical (OH⁻)).

In the present study, single i.p. injection of 1.5 ml kg^{-1} of CCl₄ to treated Wistar rats was associated with significant decreases in the tissue activities of SOD and CAT and concentration of GSH. These results are in complete agreement with observations previously

reported by Adewole et al.⁵³ and Bandhopadhy et al.⁵⁵ Thus, the decreased activity of SOD in kidney in CCl₄treated rats may be due to the enhanced lipid peroxidation or inactivation of the antioxidative enzymes. This would cause an increased accumulation of superoxide radicals, which could further stimulate lipid peroxidation. However, administration of ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone prior to CCl₄ intoxication protected the antioxidant machineries of the kidneys as revealed from increased levels of SOD, CAT, and GSH content with nifedipine producing the most significant restorative effect of all the CCBs tested.

The vital function that blood cells perform, together with the susceptibility of the highly proliferative tissue to intoxication by xenobiotics, makes the hematopoietic system unique as a target organ.⁵⁶ Certain drugs including alkylating cytotoxic agents could affect blood formation rate and normal range of hematological parameters.⁵⁶ In this study, repeated i.p. injection of gentamicin at a single daily dose of 40 mg kg⁻¹ body weight of rats for 14 days was associated with significant decreases in Hb, PCV, and RBC, while causing significant increases in the MCV and MCH. Previous studies have shown gentamicin to induce anemia via vitamins B₆,^{37,57} C, and E deficiencies.^{30,32,38} However, pretreatments with ascorbic acid, nifedipine, verapamil, diltiazem, and prednisolone significantly prevented the deleterious effect of gentamicin treatment on the Hb, PCV, and RBC. Also, the decrease in Hb, PCV, and RBC in the gentamicin-only-treated group may be due to the damage to the kidney cells that are involved in the process of erythropoiesis.³ A reduction in the erythropoiesis process produces less erythropoietin, a polypeptide hormone produced by the macula densa cells of the kidneys, which stimulates the red bone marrow, thereby increasing the rate of RBC production. With more RBCs in circulation, the oxygen-carrying capacity of the blood is greater.³ Lower RBC means lower Hb in the blood which could eventually lead to hypoxia and then death. All four drugs (nifedipine, verapamil, diltiazem, and prednisolone) were able to prevent the deleterious effect on RBC and Hb lowering associated with kidney damage.

Another interesting finding of this study is the histopathological lesions induced by gentamicin treatment on the treated kidneys. Repeated i.p. treatment with high-dose gentamicin was associated with patchy glomerular atrophy and mild tubulonephritis when compared with the control kidney. Again, rat kidneys pretreated with nifedipine, verapamil, diltiazem, and prednisolone showed no remarkable histological changes in the renal architecture except for scanty tubular necrosis. These findings corroborated the biochemical and hematological findings of this study.

In conclusion, both CCBs (nifedipine, verapamil, and diltiazem) and PLA₂ inhibitor (prednisolone) showed significant nephroprotection against drug-/ chemical-induced renal toxicities, which was mediated via antioxidant and anti-lipoperoxidation mechanisms. Thus, this study establishes the chemopreventive potentials of CCBs and prednisolone against drug-/ chemical-induced nephrotoxicities. However, further research would be required in understanding the molecular mechanisms involved in the protective effect of these CCBs and prednisolone against drug-induced nephrotoxicity.

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Conflict of interest

The authors declared no conflict of interest.

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