



Performance, materials and coating technologies of thermochromic thin films on smart windows



M. Kamalisarvestani^{a,*}, R. Saidur^a, S. Mekhilef^b, F.S. Javadi^a

^a Department of Mechanical Engineering, University of Malaya, 50603 Kuala Lumpur, Malaysia

^b Department of Electrical Engineering, University of Malaya, 50603 Kuala Lumpur, Malaysia

ARTICLE INFO

Article history:

Received 6 December 2011

Received in revised form

5 May 2013

Accepted 20 May 2013

Available online 25 June 2013

Keywords:

Smart windows

Thermochromic glazing

Thin film

Nano particle

Glass coating

ABSTRACT

A significant amount of energy is consumed to maintain thermal comfort in buildings, a huge portion of which is lost through windows. Smart coating, thin films with spectrally selective properties on the surface of glass, is the innovative solution to the problem. Thermochromic smart windows change their color and optical properties in response to temperature variations. The performance, materials, coating technologies and energy modeling of thermochromic windows are reviewed in the present study. The effect of doping vanadium dioxide (VO₂) coatings with different dopants such as tungsten, fluorine, gold nanoparticles and etc. is elaborated. Various deposition techniques, specifically hybrid chemical vapor deposition (AA/APCVD) and physical vapor deposition (PVD) methods are elucidated. Different dopants and techniques show different results on metal to semiconductor transition (MST) and the critical temperature. The “change in visible and infra-red transmission and reflectance” is the touchstone of performance for the different afforded chromogenic intelligent windows.

© 2013 Elsevier Ltd. All rights reserved.

Contents

1. Introduction.....	353
2. Smart windows.....	354
3. Thermochromic windows (TCW).....	356
4. Thermochromic materials and dopants.....	358
5. Thermochromic coating technologies.....	360
6. Energy modeling of thermochromic windows.....	361
7. Conclusions.....	361
Acknowledgment.....	361
References.....	361

1. Introduction

A significant amount of energy is consumed for maintaining thermal comfort in buildings. The energy used to maintain thermal comfort in buildings is mostly exploited to keep HVAC devices running. The building energy consumption in developed countries accounts for 20–40% of the total energy use. About 41% of primary energy the U.S. (as the second largest consumer of world energy representing 19% of global consumption), consumed in 2010 was for buildings sector. Consequently, this amount

accounted for 7% of global energy use in 2010. Approximately 60% of all used energy in building sector was consumed for space heating, space cooling, lighting and ventilation in 2010 [1]. Buildings in China, as the largest consumer of world energy, consumed 26% of total primary energy in 2006; the figure is anticipated to rise to more than 30% by 2020 [2]. The building energy consumption is even more dominant in hot and humid regions, using one-third to half of the electricity produced in some countries [3–5].

In addition, building sector was the culprit of around 40%, 18% and 8% of energy-related carbon dioxide emissions in 2010 for the US, China and worldwide, respectively [1,6]. Therefore, energy saving measures should be taken in order to reduce buildings energy losses and CO₂ emissions.

There are two approaches in building energy saving strategies, the active strategies and the passive ones. Improving HVAC

* Corresponding author: Tel: +603 79677611; mobile: +60176617504.

E-mail addresses: masoudkamalis@gmail.com, mkamali@siswa.um.edu.my (M. Kamalisarvestani).

Nomenclature

TC	Thermochromic
EC	Electrochromic
TCW	Thermochromic window
ECW	Electrochromic window
SPD	Suspended particle device
MST	Metal to semiconductor transition

ΔT	Change in transmittance
T_t	Transition temperature (critical temperature)
PVD	Physical vapor deposition
CVD	Chemical vapor deposition
APCVD	Atmospheric pressure chemical vapor deposition
AACVD	Aerosol assisted chemical vapor deposition
ΔR	Change in reflectance

systems and building lighting can actively increase the building's energy efficiency, whereas measures amending the properties and thermal performance of building envelopes such as adding thermal insulation to wall, using cool coatings on roofs and coated window glazing are among the passive methods. Any building element, such as wall, roof and fenestration which separates the indoor from outdoor is called building envelope [7–10].

Windows are known as one of the most energy inefficient components of buildings [11]. Preventing these losses by improving the windows thermal performance will result in reduced electricity costs and less greenhouse gas emissions. While controlling transmitted Infrared (IR) radiation, an ideal window should be capable of sufficient transmission of visible light [12]. The most significant parameters influencing the heat transfer through windows include outdoor conditions, shading, building orientation, type and area of window, glass properties and glazing characteristics [13]. Improving glazing characteristics of windows such as thermal transmittance and solar parameters is the most important criterion to be considered in building windows standards [14]. Several international standards have been published to evaluate the performance of windows and glasses in building in order to achieve minimum requirement by considering energy performance improvement of building. ISO 10291:1994 [15], ISO 12567:2005 [16,17], ISO 9050:2003 [18], and ISO 14438:2002 [19] are the examples of such standards.

Based on international standard (ISO 9050) [18], light transmittance, solar direct transmittance, total solar energy transmittance, ultraviolet transmittance and related glazing factors are the essential parameters for determining the light and energy performance of glazing in building. Some studies have been done based on this standard (or its European equivalent EN 410) [20] in order to determine the optical properties of coated glass products [21], modeling the solar energy transmittance of windows by considering to the angular behavior, and calculate the solar control parameters [22].

Generally, international standards specify the criteria and the essential characteristics to be considered worldwide. International standards can apply directly or modify based on the local conditions. There are many international and local standards related to the energy and lighting performance of windows, some of which are tabulated in Table 1.

The thermal dynamics and lighting potential of glazing should be considered in building energy calculations. Energy efficiency in building envelopes is generally calculated based on the ratio of the temperature difference across a building compartment and the heat flux (R -value) or the rate of heat transfer through a building element at certain conditions (U -factor). In cold climates, low U -factors or high R -values prevent heat from escaping from buildings, and in hot climates, they prevent heat from entering buildings [23,24].

2. Smart windows

Smart windows, defined as the type of windows that partially block the unwanted solar radiation, can help building to maintain higher energy performance levels. The energy performance can be

improved by increasing heat gain in cold weather and decreasing it in hot weather by adopting windows' radiative and thermal properties dynamically [25]. Adding controllable absorbing layer on the surface of the glass can change the optical properties of the glass by controlling the incident solar heat flux [26]. Therefore, smart windows lead to reduced HVAC energy consumption and size and electric demand of the building [11,27,28].

There is a wide range of modern intelligent glazing options for energy saving purposes including Low- e coatings [29,30], micro blinds, dielectric/metal/dielectric (D/M/D) films [31,32], and switchable reflective devices including electrochromic windows (ECW) [33–35], gasochromic windows [36], liquid crystal glazing [11], Suspended-Particle Devices (SPD) [37] and thermochromic windows (TCW).

Low emissivity (low- E) coatings are spectrally selective films that are aimed to let the visible light pass through and block the IR and UV wavelengths which generally create heat [10]. Because of its high IR-reflectance, this type of glazing has been developed greatly, and many have studied their different properties during the last two decades [30,38–40]. Typically, there are two types of low- e coatings, the tin oxide based hard coating and the silver based soft coating with higher IR reflectance and lower transmittance than the other one. However, the visible transmittance of hard coatings can boost up with anti-reflecting property of silicon dioxide [29].

D/M/D films on glass exhibit great energy saving effects by reflecting the IR radiation by their reflective metal film and transmitting visible and near IR radiation through the two antireflective dielectric coatings [41]. Design, fabrication and properties of D/M/D films have been studied thoroughly by many researchers focusing on the optimization [31,32,42–45]. Beside the optimized performance, cost of these films in terms of their material and the fabrication technique is also important [41].

The switchable reflective devices (also called dynamic tintable windows) are categorized in to passive and active systems. In passive devices, the switching process is activated automatically in accordance with the environmental conditions. This environmental factor can be light in case of photochromic windows; or temperature and heat in thermochromic windows (TCW). Alternatively, the active systems require an external triggering mechanism to perform the modulation. For instance, electricity is the actuating signal in electrochromic windows (ECW). The active switchable glazing systems offer supplementary options compared to the passive systems whereas their dependency on power supply and wiring should be reckoned with as a drawback. Chromic materials, liquid crystals, and suspended particle windows are the three most common active-controlled intelligent windows [11]. The latter two share the disadvantage of their dependency on an electric field to be maintained when a transparent mode is desired; resulting in excessive electricity consumption. This is not the case in EC glazing that wants electricity only for transition [46]. However, chromic materials are classified into four types: electrochromic (EC), gasochromic, photochromic and thermochromic (TC). The first two belong to active glazing, responding to electricity and hydrogen

gas, respectively, as a function of solar irradiation [11,47]. Smart windows are apt to glazing the cooling load demanding buildings with large solar gain [48], though providing a see-through mode is a must in any application.

The EC effect which was first explained in 1969 is a characteristic of a device which varies its optical properties when an external voltage triggers the EC material. The EC device modulates its transmittance in visible and near IR when a low DC potential is applied [34,49].

It is usually consisted of several layers deposited on glass. The glass substrates are usually coated with transparent conducting films with natural colors—mostly tin oxide doped with either indium (ITO) or fluorine (FTO). The three major deposited layers cover the coated glass substrate as follows: *The Electrochromic film* (cathodic electro-active layer with reversible transmittance modulation characteristic) which gets a darker color when the external circuit transfers electrons into the EC lattice to compensate for the positive ions injected from the adjacent ion storage layer; *Ion conductor* (ion conducting electrolyte); and *Ion storage layer* (anodic electro-active layer) that becomes darker while releasing positive ions [33,34,50–52].

The electro-active layers (also named electrochromics) switch between their oxidized and reduced forms causing variations in their optical properties and colors as well. Ideally, it is desired that electrochromics act more reflective rather than absorptive in their colored state compared with their bleached mode [49].

EC windows should provide daylight while acting as a barrier to heat. Obviously, this type of window is not capable of providing

both effects simultaneously [35]. The EC function can be controlled by thermal load, temperature and sunlight. The latter is stated to be the best governing parameter, especially from the comfort point of view [53–56]. All the more, self-powered EC windows are also developed using semitransparent PV cells, which provide the required activating electricity [57–65].

The function of gasochromic devices is also based on electrochromism in EC windows. The main difference is that instead of DC voltage, a hydrogen gas (H_2) is applied to switch between colored and bleached states. Compared to their counterpart, gasochromic devices are cheaper and simpler because only one EC layer is enough and the ion conductor and storage layers are not needed anymore. Although, gasochromic devices exhibit some merits such as better transmittance modulation, lower required voltage, staying lucid in the swap period, and adjustability of any middle state between transparent and entirely opaque; only a few numbers of EC materials can be darkened by hydrogen. Furthermore, strict control of the gas exchange process is another issue [66].

Commonly used in wrist watches, LC technology is getting more popular as a means of protecting privacy in some interior applications such as bathrooms, conference halls and fitting rooms in stores. Two transparent conductor layers, on plastic films squeeze a thin liquid crystal layer, and the whole set is pressed between two layers of glass. Normally, the liquid crystal molecules are situated in random and unaligned orientations scattering light and cloaking the view to provide the interior space with privacy. When the power is switched on the two conductive layers provide

Table 1
Energy and lighting performance standards of windows.

Name	Description	Country
ASTM E-2141-06	Standard test methods for assessing the durability of absorptive electrochromic coatings on sealed insulating glass units	International [11]
ISO 9050:2003	Glass in building—determination of light transmittance, solar direct transmittance, total solar energy transmittance, ultraviolet transmittance and related glazing factors	International [22]
ISO 10291:1994	Glass in building—determination of steady-state U values (thermal transmittance) of multiple glazing—Guarded hot plate method	International
ISO 10292:1994	Glass in building—calculation of steady-state U values (thermal transmittance) of multiple glazing	International
ISO 10293:1997	Glass in building—determination of steady-state U values (thermal transmittance) of multiple glazing- Heat flow meter method	International
EN 1096-1:2012	Glass in building—coated glass—Part 1: definitions and classification	European
EN 1096-2:2012	Glass in building—coated glass—Part 2: requirements and test methods for class A, B and S coatings	European
EN 1096-3:2012	Glass in building—coated glass—Part 3: requirements and test methods for class C and D coatings	European
EN 1096-4:2004	Glass in building—coated glass—Part 4: evaluation of conformity/Product standard	European
EN ISO 14438:2002	Glass in building—determination of energy balance value—calculation method (ISO 14438:2002)	European
BS EN 14449-1:2005	Glass in building. Laminated glass and laminated safety glass. Evaluation of conformity/product standard	British
BS EN 14179-2:2005	Glass in building. Heat-soaked thermally-toughened soda lime silicate safety glass. Evaluation of conformity/product standard	British
BS EN 1748-1-2:2004	Glass in building. Special basic products. Borosilicate glasses. Evaluation of conformity. Product standard	British
BS EN 12337-2:2004	Glass in building—chemically strengthened soda lime silicate glass—evaluation of conformity/Product standard	British
BS EN 14178-2:2004	Glass in building—basic alkaline earth silicate glass products—evaluation of conformity/Product standard	British
BS EN 1096	Coated glass	British
BS EN 12898:2001	Glass in building, determination of the emissivity	British
BS EN 410: 1998	Glass in building—determination of luminous and solar characteristics of glazing	British
BS EN 673: 1998	Glass in building: determination of thermal transmittance (value). Calculation method	British
BS EN ISO 12567	Thermal performance of windows and doors	British
BS EN ISO 14438:2002	Glass in building. Determination of energy balance value. Calculation method	British
ES ISO 12567-1:2012	Thermal performance of windows and door—determination of thermal transmittance by the hot box method—complete windows and doors	Ethiopian
ES ISO 12567-2:2012	Thermal performance of windows and door—determination of thermal transmittance by the hot box method—roof windows and other projecting windows	Ethiopian
ES ISO 14438:2012	Glass in building. Determination of energy balance value. Calculation method—calculation method	Ethiopian

Table 2
Comparison between different smart windows.

Smart windows	Activation	Colour	Advantages	Disadvantages	Thermal performance	Optical performance	Application
Photochromic	Light	Bleach	No activation electricity, No foggy effect, Automatic	Manufacturing difficulties for large sizes, limited application, Cannot reduce heat gain, More darker in the winter than the summer		UV protector, darken and lighten at sunlight and dusk	Sunglasses, Supramolecular chemistry, data storage
TC	Heat	Colored	High energy saving, No activation electricity	No outside visibility, manufacturing difficulties for large sizes	Reflecting heat gain	Reflecting infrared light	Duracell battery state indicators, Thermochromic paints, Thermochromic papers are used for thermal printers, thermochromism polymer
EC	Voltage/current	Colored	Energy saving	Activation electricity is needed	Control the amount of heat	Reversibly changing color when a burst of charge is applied, control the amount of light	Electrochromic devices such as windows and smart glasses, Automobile industry
Liquid Crystal	Voltage	Bleach	Control privacy	No energy saving, No outside visibility, Activation electricity is needed	High heat gain	Transmit incident light	Electrooptical devices, hyperspectral imaging
Suspended Particle	Current	Bleach	Instantaneous control of light, Outside visibility, Have wide range of transmittance	Limited in size, Activation electricity is needed	Energy saving	UV protector, Reduction of infrared light,	Polaroid camera, Windows

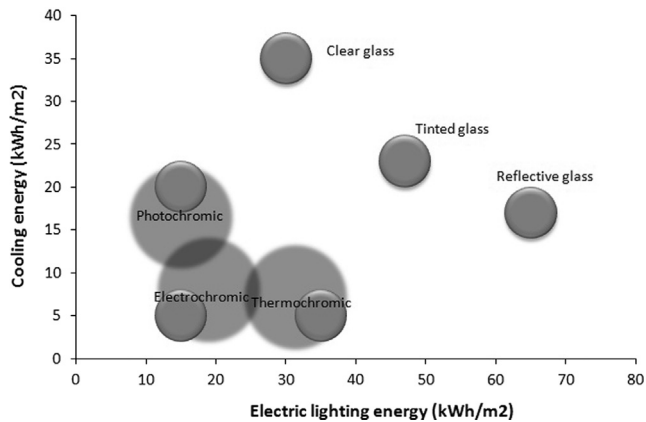


Fig. 1. Comparison of electric lighting energy and cooling energy between different glazing types [24].

an electric field via their electrodes. The field causes the crystal molecules to be positioned in an aligned direction causing a change in transmittance [67,68]. The LC technology suffers from the disadvantage of high power demand in transparent mode, resulting in an electricity usage of 5–20 W/m². These devices have problems, in long term UV stability and high cost disadvantages as well [11]. The technology using liquid crystals in intelligent windows is called Polymer dispersed liquid crystals (PDLC).

SPDs have many things in common with LC devices: they are both fast in switching between phases, high electricity consumptive and dependent on an electric field. They consist of the liquid like active layer formed by adsorbing dipole needle-shaped or spherical particles (molecular particles), i.e. mostly polyhalide, suspended in an organic fluid or gel sandwiched between two sheets of glass coated with transparent conductive films. Normally, the device is in the dark reflective state because of the random

pattern of the active layer's light absorbing particles. When the electric field is applied, the particles will align resulting in the clear transmissive state. As soon as the power turns off the device switches to its dark state. Typically, the transmission of SPDs varies between 0.79–0.49 and 0.50–0.04, with 100 to 200 ms switching time and 65–220 V AC requirements [11]. The critical comparison between different smart windows is summarized in Table 2.

Shown in Fig. 1, clear glass, tinted glass, reflective glass and three chromogenic fenestration technologies are compared regarding to their cooling and electric lighting energy. As it is observed EC and TC windows demand the lowest cooling energy; and as previously reported ECWs require less energy for lighting than TCWs [69]. However, the necessity of wiring in EC glazing and the better ability of TCWs to maintain the visible transmission (when doped properly) [70] besides their simple structure [71] have given TCWs a cutting edge as a low-priced [72] alternative to the other counterparts.

Different technologies such as electrochromic, gasochromic, liquid crystal and suspended particle devices have been widely reviewed neglecting the TCWs [11,37,46,73]. As a result, there was a need to review thermochromic windows comprehensively. The following sections address the application of TC thin films in fenestration. Part 3 introduces thermochromism, TCWs, their structure, and their performance. Part 4 discusses TC materials, the dopants and the nanoparticles. In part 5 thin film deposition and fabrication techniques are addressed. In part 6 the energy modelings done on TCW are discussed. And lastly, in part 7 there will be a conclusion covering the whole papers reviewed.

3. Thermochromic windows (TCW)

In the first glance, the word “Thermochromic” might seem strange. But the word originates from the Greek roots: “Thermos” meaning warm or hot; and “Chroma” which means color. Generally,



Fig. 2. Sequential color switching of a thermochromic laminated glass [4].

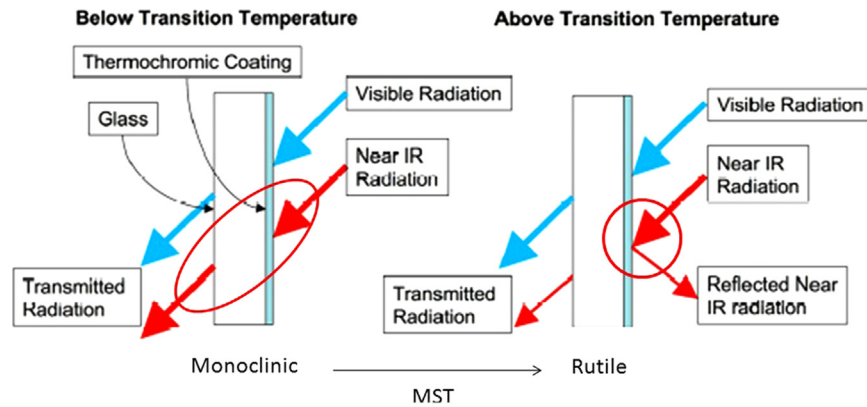


Fig.3. Schematic representation of thermochromic materials applied as an intelligent windows coating [77].

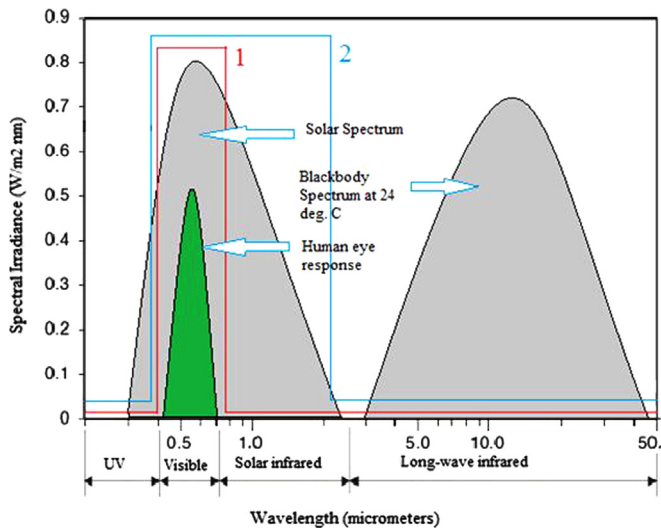


Fig. 4. The spectral transmittance of perfect TCW in cold and hot weather conditions (adopted from [81]).

thermochromic (TC) materials change color in response to temperature variations. Thermochromic window, an example of which is shown in Fig. 2, is a type of glass with TC materials that can reduce the energy demand of buildings by changing the device's reflectance and transmission properties and reducing the unwanted solar energy gain [74,75].

The TC thin film is initially in its monoclinic state (cold state) at lower temperatures (usually room temperature). Monoclinic materials behave as semiconductors, less reflective especially in near-IR (NIR) radiation. As the temperature becomes higher than a certain point, the TC material changes its nature from monoclinic to rutile state. This phenomenon is called metal to semiconductor

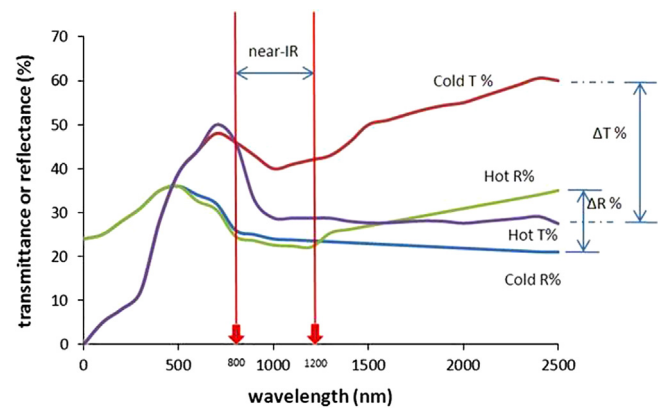


Fig. 5. Typical transmittance change (ΔT) and reflectance change (ΔR) of thermochromic films through MST (adapted from [91]).

Table 3

The Ideal optical performance of thermochromic windows (adapted from [91]).

State	Monoclinic/cold ($T < T_t$)		Rutile/hot ($T > T_t$)	
Wavelength	Visible (%)	NIR (%)	Visible (%)	NIR (%)
Transmittance (T)	60–65	80	60–65	15
Reflectance (R)	17	12	17	77

transition (MST). In rutile state (hot state) the material acts like a semi-metal, reflecting a wide range of solar radiation [76] (Fig. 3).

As illustrated in Fig. 4, the majority of heat gain in solar spectrum takes place at NIR range (800–1200 nm) [78–80].

The red line (line 1) indicates the transmittance of perfect TCW in cold state. Visible light should be transmitted and near infrared radiation should be reflected. Long-wave radiation is also reflected back to indoor. This transmittance approach leads to reduction of

Table 5
Thermochromic materials.

TC material	T_t (°C)	Cold state color	Hot state color	Reason of transition
Cuprous mercury iodide (Cu ₂ HgI ₄) [112]	55	Bright red	Dark brown	Cu(I)–Hg(II) charge transfer
Silver mercury iodide (Ag ₂ HgI ₄) [112]	47– 51	Yellow	Orange	Cu(I)–Hg(II) charge transfer
Mercury(II) iodide	126	Red	Pale yellow	Reversible change transition
Bis (dimethylammonium) tetrachloronickelate [113]	110	Raspberry-red	Blue	
Bis (diethylammonium) tetrachlorocuprate [112]	52– 53	Bright green	Yellow	Relaxation of the hydrogen bond and change of arrangement of the copper atom's d-orbitals d
Nickel sulfate	155	Green	Yellow	
Chromium(III) oxide: aluminium(III) oxide (1:9) [114]	400	Red	Grey	Changes in its crystal field
chromium-rich pyropes normally reddish-purplish	80		Green	
titanium dioxide	80		Green	
zinc oxide		White	Yellow	
indium(III) oxide		White	Yellow	
Lead(II) oxide		Dark yellow	Yellow brown	

Table 6
Effect of some materials on the thermochromic coatings.

Material	Effect on MST	T_t	Film color
Pure VO ₂ crystals	–	68 °C [117]	Brown/yellow [122]
Un-doped VO ₂ films	–	50–66 °C [89]	Brown/yellow
Tungsten doping	23% ΔT at 2500 nm [108]	20 °C [76] 1.56 at% 25 °C [133] 2.7 mol%	Blue [76]
Gold nanoparticle	35–40% ΔT and 10% ΔR at 2500 nm [134]	15–20 °C [134]	Green/blue [124,125]
Fluorine doping	15 % ΔT and 5 % ΔR [127]	60 °C [127] 25 °C [28]	Varies in different temperatures Brown/yellow [127]
TiO ₂	21.2% IR- ΔT and 84% Vis- ΔT [130] 5–10% ΔT at 2500 nm [128]	50–60 °C [128]	Brown [92,128]
CeO ₂	5–10% ΔT at 2500 nm [128]	50–60 °C [128]	Brown/yellow [128]

Fluorine and tungsten-doped VO₂ are good choices for energy-saving windows [108,121]. However, tungsten (W) by reducing the T_t of VO₂ to the ideal temperature of 25 °C at 2 at% (lowering the T_t by 25 °C per dopant atomic percent incorporated) is the most effective dopant ion for VO₂ [76,122]. In another experiment, it is shown that a 2.7 mol% content of WO₃ has also made the film's T_t drop to room temperature [123].

Gold nanoparticles have been used as dopant in recent studies. It gives a pleasant green/blue color to the films and affects the MST and T_t . However, because of its high price, gold doping cannot be prevalent [124,125].

Fluorine doping of VO₂ films have been done by PVD techniques [28,121,126]. Incorporation of fluorine in the films using AACVD gave the films a lighter color but still the unpleasant yellow/brown color of un-doped film did not change [127]. The changes in transmission (15%) and reflectance (5%) are less than those of tungsten-doped films. The key effect of fluorine is the better transmission in the visible range. In contrast to previous studies [28], it is reported that the transition temperature (~60 °C) did not change noticeably from the un-doped case [127].

Performed titanium dioxide (TiO₂) and cerium dioxide (CeO₂) nanoparticles have been also used in producing VO₂ films. By utilizing these photocatalytic nanoparticles, films with both photocatalytic and TC characteristics can be afforded. Through MST, the transmittance changes insignificantly (5–10%) at

2500 nm, while the reflectance changes about 30%. The nanoparticles used do not change the color of the film [128]. Moreover, CeO₂ can be used as a protective coating over VO₂ films to plunge the chemical deteriorations [129].

TiO₂ is used to fabricate double- (or multi-) layered VO₂ thin films. The anti-reflection feature of TiO₂ can enhance the visible transmittance up to 84% [130]. Deposition of platinum (Pt) on VO₂ films can improve the IR reflection while maintaining the TC characteristic. T_t has been also lowered by 9.3 °C to around 58 °C. Pt has depressed the visible transmittance which can be fortified by adding SiO₂ antireflection coating [131]. Earlier, SiO₂ was used to improve the luminous transmittance to 75% (in 700 nm) compared with 33% of plain VO₂ case. SiO₂ has also lowered T_t to 61 °C [132].

There are some materials which can be used as deposition precursor. The most common precursors are vanadium alkoxides [89,135–137] and oxovanadium reagents [138]. Vanadium chloride should react with a source of oxygen (such as methanol or ethanol) the product of which is V₂O₅; heating the product in a reducing atmosphere results in deposition of VO₂ [139]. Addition of tetraoctyl ammonium boromide (TOAB) as surfactant to the vanadium dioxide matrix can control the distribution, shape and size of gold nanoparticles. It is reported that surfactant TOAB can reduce T_t from 50–54 °C to 42–45 °C [140] and from 52 °C to 34 °C [134].

Surfactants are molecules, which can alter the structure of films due to their ability to affect the surface tension of liquids and deposition mechanism in hybrid CVD processes. Using surfactants can decrease T_t of VO₂ films [125,134,140].

Thermochromic coatings can also be used on surfaces other than window. For example, Karlessi et al. introduced TC coatings as buildings external surfaces to reduce the heating and cooling loads in winter and summer [109].

5. Thermochromic coating technologies

Deposition of TC thin films on different substrates by physical vapor deposition (PVD), sputtering and sol-gel [28,74,121,141–145] techniques was executed by the early 2000s. In particular, VO₂ as the most appropriate TC material had been deposited using sol-gel [135,146–148], sputtering [71,80,149], pulsed laser techniques and chemical vapor deposition methods (CVD). These methods have been widely reviewed in previous studies [87,94,150,151].

As it is mentioned before, doping VO₂ films with tungsten can enhance their TC property. The doping and production methods include sputtering and PVD [28,141,152] sol-gel [74,135,144,146–148], APCVD [76,79,89,122,153,154] and AACVD [155,156]. Despite the technological developments in surface engineering, until now, low visible light transmission, high transition temperatures, low durability, poor transition performance, high deposition costs and unpleasant visible colors did not let these techniques be marketable.

For instance, in an attempt to decrease T_t by doping Sn on VO₂ films using PVD, it was reported that the un-doped films showed much lower T_t [141]. Even the recent production of w-doped VO₂ and V₂O₅ films prepared by sputtering has not shown good switching performance [110]. There have been some efforts to optimize the TC switching of films by controlling the deposition conditions. The best results are 76% for ΔT and 75% for ΔR ; though, T_t was still high [80].

Reactive sputtering is the most widely PVD method used to produce thermochromic thin films by means of ion beam sputtering [157], DC magnetron sputtering [152,158,159] and RF magnetron sputtering [78,160,161].

On the other hand, chemical vapor deposition (CVD) shows potential in manufacturing glasses in commercial scale. Specifically, atmospheric pressure chemical vapor deposition (APCVD) is well matched with float-glass mass production lines. The deposition rates of VO₂ films are noticeably fast in this method [76]. By the same token, the resulting physical properties such as durability and adhesion make this method more suitable [91]. There have been many researches investigating various CVD techniques and doping effects [76,79,122,162].

In a study conducted by Binions et al. [89] APCVD method was used to produce tungsten-doped VO₂ films by reaction of vanadyl acetylacetonate and tungsten hexachloride with oxygen. It is reported that tungsten doping causes a considerable drop in T_t as well as a great change in near infra-red optical properties [75]. The film thickness was also found to influence the transmission [89] and extent of TC switching [89,136]. The film thickness can be adjusted by changing the deposition time [134].

Using APCVD, TiO₂ on VO₂ on SiO₂-coated glass and VO₂ on TiO₂ on SiO₂-coated glass have been grown and compared. In both cases, the T_t did not change significantly. The $\Delta T\%$ in the first multilayer film was 92% noticeably more than that of the case of VO₂ over TiO₂ (71%). The two films showed $\Delta R\%$ of 55% and -55%, respectively [92]. In another study, two deposition methods were used: low pressure metal-organic CVD (MOCVD), and argon

annealing of VO₂ (B) films by MOCVD. It was found that the film microstructure regulates the MST markedly [111].

Manning et al. [79,122] used water, vanadium and tungsten sources to prepare VO₂ thin films by CVD. The resulting films had the disappointing transmission of 11% to 40%. As previously suggested, the acceptable transmission level should be above 60% so that the window's optical properties can live up to the building's aesthetical and lighting standards [152].

Vernardou et al. [162] have studied the aftereffect of doping monoclinic VO₂ by atmospheric-pressure direct liquid injection metal-organic CVD (DLI-MOCVD). It was shown that T_t of tungsten-doped VO₂ drops from 60 °C to 35 °C. Aerosol assisted CVD (AACVD) is also another deposition process reducing V₂O₃ to V₂O₅ [155]. However, the AACVD techniques do not feature good mechanical properties and can be easily removed from the surfaces [163].

Hybrid atmospheric pressure and aerosol assisted (AA/APCVD) method can be used to prepare VO₂ thin films with reduced transition temperatures [124,134]. In this technique, the multifunctional feature of AACVD is mixed with the mechanical sturdiness of APCVD [124]. This method has been also used to prepare gold doped VO₂ films. Due to the surface plasmon resonance (SPR) bands, the gold nanoparticles change the color of films from unpleasant yellow/brown color to more appealing green/blue colors. In addition, it reduces the T_t and causes a surge in reflectance. The hybrid method results in more adherent films compared with AACVD films. As MST takes place, a 10% change in reflectance and 30–40% change in transmittance is observed at 2500 nm. The reason for this change may be the metallic nature of gold nanoparticles [140]. Hybrid AA/APCVD is also used to prepare VO₂ films by using a suspension of TiO₂ and CeO₂ [128].

Blackman et al. [76] used Water, vanadium chloride, and tungsten chloride for synthesizing tungsten-doped VO₂ thin films by APCVD. Among the different afforded un-doped films, the optical properties of which having been measured between 300 nm to 2500 nm wavelengths, the 80 nm-thick films showed the best performance with 60% transmission at 570 nm (visible range) and 35% change in transmission at 2500 nm (far infra-red range) through transition. Testing the tungsten doped films, the effect of different tungsten atomic percentages on reducing T_t has been evaluated. As it was previously reported [157], by increasing the tungsten content T_t changes (linearly [108]) approximately by -20 °C/at% until 3 at% after which the reduction will be pseudo linearly. In addition, the powdery yellow/brown color of films can change to more adhering, more transparent and bluer/greener films by adjusting the vanadium precursor to water ratio and increasing the tungsten content of films to more than 2.5 at%.

In choosing the deposition technique, several factors should be taken into mind including the limitations of the material to be deposited, substrate material, deposition rate, the cost of required equipments, scalability, environmental considerations and the desired film features such as thickness, micro structure, mechanical strength, and optical, thermal and thermochromic performance.

In the first glance, comparing the different VO₂ coating methods results in baffling outcomes. Compared to other techniques, PVD methods require expensive equipments and high vacuum pressure. In addition, they are time and energy consumptive and suffer from low film growth rates. To the contrary, PVD methods are more suitable for synthesizing ultra thin films, require lower temperatures, are more environmentally-friendly, compatible to a wider range of substrates and more convenient for developing multi-layer thin films [150,164]. On the other hand, CVD is believed to have the potential for producing glass in commercial scale, due to its compatibility to float-glass production line [165] and fast deposition rates [165,166]. However, PVD and CVD are considered as energy consuming techniques and are not

cost-efficient. Gao et al. reviewed solution processes for VO₂ films preparation and concluded that these methods are cheap, suitable for scaling up and easy to be utilized in practical applications. In contrast, the film characteristics such as thickness and microstructure are not precisely controllable by solution based methods [94].

To sum up, both PVD and CVD methods are suitable for experimental scale due to their scalability and superiority in terms of controlling process parameters and film features. However, they are restricted by the expensive machines and processes they require, while solution processes do not suffer from these drawbacks in industrial scale. There are two suggested approaches to these shortcomings: (1) to modify the PVD and CVD methods and to design more inexpensive equipments and coating processes (2) to improve the solution based methods so that the film features including visibility, micro structure, color, and film thickness can be adjusted by controlling the process parameters.

6. Energy modeling of thermochromic windows

There are various software tools which have been used in previous researches for window simulation including energy-10 [12], the window simulation tool Winsel [167], simulation tool SOLENE [168], self developed simulation software (SDS) based on the ASHRAE tables [13], DOE building energy simulation program [169], TRNSYS building energy simulation program [66] and simulation package Integrated ENergy Use Simulation (IENUS) [27]. Different smart windows configurations have been also simulated and modeled [167,170–175] for evaluating their energy saving effect and optical performance.

Based on theory, TCWs are capable of curtailing the buildings energy consumption by allowing visible day light, limiting undesired solar gain in the hot seasons and allowing favorable solar heat gain during the cold seasons. There a few number of studies which modeled and calculated the energy performance of TCWs. Saeli et al. modeled TCWs by energy plus software and demonstrated the technology's energy saving effectiveness both in lighting and ventilation. The total energy consumption reduced more for commercial windows (100% of wall area glazing) than the residential case (25% of wall area glazing). Since high temperature lets the window be longer in its rutile state, it was also discovered that the technology works best in cities with warmer climates [91]. They also compared TCWs with conventional glazing and showed the effectiveness of TC coating in saving energy [176].

In another study Xu et al. compared the cooling energy consumption of white glass and four Low-E glasses by using the software TRNSYS 16 and showed the best performance was attained by the double glass when the VO₂ films were coated on the inside surface of the outer pane with 85% energy saving compared to white glass. However, the heating energy consumption was the highest for TCWs [177]. The result of this study also shows that TCWs are more suitable for cooling demand climates.

To the contrary, Ye et al. evaluated the energy consumption of different windows by an energy analysis program "BuildingEnergy" and showed that VO₂ glazing has no apparent energy saving benefit and solar control advantage over conventional glazing. They concluded that, controlling the emissivity of the window is more beneficial than regulating the solar transmissivity. It was also concluded that, the energy saving effect of TCWs in summer is due to low transmittance of solar radiation and the higher absorptivity in the metal state results in higher energy consumptions which consequently makes the phase transition useless to the energy saving performance [151]. The inconsistency and scarcity of the studies in modeling the energy performance of TCWs emphasizes the necessity of more research in this field.

7. Conclusions

It's been decades since the time Vanadium dioxide thermo-chromic coating was reported, however, TCWs have not been commercialized due to shortcomings such as low luminance visibility, unattractive colors, low energy-saving efficiency and high coating costs. Nano technology, suitable dopants and adding efficient anti reflecting coating can reduce transition temperature (near room temperature) and improve visible transmittance (more than 60%). Appropriate and cost-efficient coating technologies provide optimum thickness (40–80 nm), sufficient thermo-chromic transition (more than 50%) and reduces the coating costs.

The coatings can be doped with different nanoparticles. Each dopant induces a special effect on the coating. Tungsten lowers the transition temperature, gold nanoparticles bring more pleasant film colors, fluorine increases the visible transmittance and titanium dioxide adds self-cleaning and mechanical strength to the films. The most common preparation methods are PVD, sol-gel techniques, and CVD. CVD is fast and suitable for mass production. AACVD and APCVD are the two most up-to-date deposition routes having multifunctional characteristics and mechanical strength, respectively. A prudent manner is to combine the qualities of both methods by employing hybrid AA/APCVD. To recapitulate, both PVD and CVD methods are suitable for experimental scale due to their scalability and superiority in terms of controlling process parameters and film features. However, they are restricted by the expensive machines and processes they require, while solution processes do not suffer from these drawbacks in industrial scale. There are two suggested approaches to these shortcomings: (1) to modify the PVD and CVD methods and to design more inexpensive equipments and coating processes (2) to improve the solution based methods so that the film features including visibility, micro structure, color, and film thickness can be adjusted by controlling the process parameters.

All in all, the major aims that must be reckoned with thermo-chromic glazing are to maximize the change in infra-red (predominantly 800–1200 nm) Reflectivity and transmission, tapering the transition temperature to near room temperature and maintaining the proper visible transmission to conserve the lighting energy. By the same token, the emissivity of the films should be modulated. Since the emissivity of the coatings are high in both monoclinic and rutile states, this technology does not work well in cooler climates currently. In addition, the energy simulation of thermo-chromic windows also highlights the fact that this type of glazing is more efficient in warmer climates.

Finally, it should be mentioned that there are a few works contributed to the energy modeling and heat transfer analysis of thermo-chromic thin films. This can be emphasized in the future attempts.

Acknowledgment

The authors would like to acknowledge the financial support from the High Impact Research Grant (HIRG) scheme (UM-MoHE) project no UM.C/HIR/MOHE/ENG/24 (D000024-16001) to carry out this research.

References

- [1] DoE U. Buildings energy databook. Energy Efficiency & Renewable Energy Department 2011.
- [2] Fridley, DG., Estimating total energy consumption and emissions of China's commercial and office buildings. 2008.
- [3] Al-Rabghi OM, Hittle DC. Energy simulation in buildings: overview and BLAST example. Energy Conversion and Management 2001;42(13):1623–35.

- [4] Wilde PD, Voorden MVD. Providing computational support for the selection of energy saving building components. *Energy and Buildings* 2004;36(8):749–58.
- [5] Kwak SY, Yoo SH, Kwak SJ. Valuing energy-saving measures in residential buildings: a choice experiment study. *Energy Policy* 2010;38(1):673–7.
- [6] Hong T. A close look at the China design standard for energy efficiency of public buildings. *Energy and Buildings* 2009;41(4):426–35.
- [7] Bojic M, Yik F, Sat P. Influence of thermal insulation position in building envelope on the space cooling of high-rise residential buildings in Hong Kong. *Energy and Buildings* 2001;33(6):569–81.
- [8] Cheung CK, Fuller R, Luther M. Energy-efficient envelope design for high-rise apartments. *Energy and Buildings* 2005;37(1):37–48.
- [9] Synnefa A, Santamouris M, Akbari H. Estimating the effect of using cool coatings on energy loads and thermal comfort in residential buildings in various climatic conditions. *Energy and Buildings* 2007;39(11):1167–74.
- [10] Sadineni SB, Madala S, Boehm RF. Passive building energy savings: a review of building envelope components. *Renewable and Sustainable Energy Reviews* 2011;15(8):3617–31.
- [11] Baetens R, Jelle BP, Gustavsen A. Properties, requirements and possibilities of smart windows for dynamic daylight and solar energy control in buildings: a state-of-the-art review. *Solar Energy Materials and Solar Cells* 2010;94(2):87–105.
- [12] Correa G, Almanza R. Copper based thin films to improve glazing for energy-savings in buildings. *Solar Energy* 2004;76(1-3):111–5.
- [13] Hassouneh K, Alshboul A, Al-Salaymeh A. Influence of windows on the energy balance of apartment buildings in Amman. *Energy Conversion and Management* 2010;51(8):1583–91.
- [14] Tarantini M, Loprieno AD, Porta PL. A life cycle approach to Green Public Procurement of building materials and elements: a case study on windows. *Energy* 2011;36(5):2473–82.
- [15] ISO, ISO 10291,2,3—Glass in building—Determination of steady-state U values (thermal transmittance) of multiple glazing 1994, International Standard Organization.
- [16] ISO, ISO 12567-1—Thermal performance of windows and doors—determination of thermal transmittance by hot box method—roof windows and other projecting windows, 2005, International Standard Organization.
- [17] ISO, ISO 12567-2—Thermal performance of windows and doors—determination of thermal transmittance by hot box method—roof windows and other projecting windows, 2005, International Standard Organization.
- [18] ISO, ISO 9050—determination of light transmittance, solar direct transmittance, total solar energy transmittance, ultraviolet transmittance and related glazing factors, 2003, International Standard Organization.
- [19] ISO, ISO 14438—Glass in building—determination of energy balance value—Calculation method 2002.
- [20] Karlsson J, Roos A. Modelling the angular behaviour of the total solar energy transmittance of windows. *Solar Energy* 2000;69(4):321–9.
- [21] Hutchins MG, et al. Measurement and prediction of angle-dependent optical properties of coated glass products: results of an inter-laboratory comparison of spectral transmittance and reflectance. *Thin Solid Films* 2001;392(2):269–75.
- [22] Alvarez G, et al. Spectrally selective laminated glazing consisting of solar control and heat mirror coated glass: preparation, characterization and modelling of heat transfer. *Solar Energy* 2005;78(1):113–24.
- [23] IEA. Energy efficiency requirements in building codes, energy efficiency politics for new building. International Energy Agency 2008.
- [24] Kamalisarvestani M, Mekhilef S, Saidur R. Analyzing the optical performance of intelligent thin films applied to architectural glazing and solar collectors. *Sustainability in Energy and Buildings*. Springer; 813–26.
- [25] Asdrubali F, Baldinelli G. Theoretical modelling and experimental evaluation of the optical properties of glazing systems with selective films. *Building Simulation* 2009;2(2):75–84.
- [26] Dussault J-M, Gosselin L, Galstian T. Integration of smart windows into building design for reduction of yearly overall energy consumption and peak loads. *Solar Energy* 2012;86(11):3405–16.
- [27] Gugliemetti F, Bisegna F. Visual and energy management of electrochromic windows in Mediterranean climate. *Building and Environment* 2003;38(3):479–92.
- [28] Burkhardt W, et al. W- and F-doped VO_2 films studied by photoelectron spectrometry. *Thin Solid Films* 1999;345(2):229–35.
- [29] Hammarberg E, Roos A. Antireflection treatment of low-emitting glazings for energy efficient windows with high visible transmittance. *Thin Solid Films* 2003;442(1-2):222–6.
- [30] Rosencrantz T, et al. Increased solar energy and daylight utilisation using anti-reflective coatings in energy-efficient windows. *Solar Energy Materials and Solar Cells* 2005;89(2-3):249–60.
- [31] Cheng CH, Ting JM. Transparent conducting GZO, Pt/GZO, and GZO/Pt/GZO thin films. *Thin Solid Films* 2007;516(2-4):203–7.
- [32] Durrani S, et al. Dielectric/Ag/dielectric coated energy-efficient glass windows for warm climates. *Energy and Buildings* 2004;36(9):891–8.
- [33] Zinzi M. Office worker preferences of electrochromic windows: a pilot study. *Building and Environment* 2006;41(9):1262–73.
- [34] Granqvist CG. Handbook of inorganic electrochromic materials. Elsevier Science Ltd.; 1995.
- [35] Lee ES, DiBartolomeo D. Application issues for large-area electrochromic windows in commercial buildings. *Solar Energy Materials and Solar Cells* 2002;71(4):465–91.
- [36] Wittwer V, et al. Gasochromic windows. *Solar Energy Materials and Solar Cells* 2004;84(1-4):305–14.
- [37] Lampert CM. Chromogenic smart materials. *Materials Today* 2004;7(3):28–35.
- [38] Huang S, et al. Determination of optical constants of functional layer of online Low-E glass based on the Drude theory. *Thin Solid Films* 2008;516(10):3179–83.
- [39] Ren Z, et al. Electrical and corrosion properties of the Ti5Si3 thin films coated on glass substrate by APCVD method. *Journal of Non-Crystalline Solids* 2011;357(15):2802–9.
- [40] Schaefer C, Bräuer G, Szczyrbowski J. Low emissivity coatings on architectural glass. *Surface and Coatings Technology* 1997;93(1):37–45.
- [41] Leftheriotis G, Yianoulis P, Patrikios D. Deposition and optical properties of optimised ZnS/Ag/ZnS thin films for energy saving applications. *Thin Solid Films* 1997;306(1):92–9.
- [42] Al-Shukri A. Thin film coated energy-efficient glass windows for warm climates. *Desalination* 2007;209(1-3):290–7.
- [43] Dima I, et al. Influence of the silver layer on the optical properties of the $\text{TiO}_2/\text{Ag}/\text{TiO}_2$ multilayer. *Thin Solid Films* 1991;200(1):11–8.
- [44] Lee CC, Chen SH, Jaing C. Optical monitoring of silver-based transparent heat mirrors. *Applied Optics* 1996;35(28):5698–703.
- [45] Leftheriotis G, Yianoulis P. Characterisation and stability of low-emittance multiple coatings for glazing applications. *Solar Energy Materials and Solar Cells* 1999;58(2):185–97.
- [46] Lampert CM. Smart switchable glazing for solar energy and daylight control. *Solar Energy Materials and Solar Cells* 1998;52(3-4):207–21.
- [47] Bahaj ABS, James PAB, Jentsch MF. Potential of emerging glazing technologies for highly glazed buildings in hot arid climates. *Energy and Buildings* 2008;40(5):720–31.
- [48] Sullivan R, et al. Energy performance of evacuated glazings in residential buildings. CA (United States): Lawrence Berkeley National Lab.; 1996.
- [49] Syrrakou E, Papaefthimiou S, Yianoulis P. Eco-efficiency evaluation of a smart window prototype. *Science of The Total Environment* 2006;359(1-3):267–82.
- [50] Papaefthimiou S, Leftheriotis G, Yianoulis P. Study of electrochromic cells incorporating WO_3 , MoO_3 , $\text{WO}_3\text{-MoO}_3$ and V_2O_5 coatings. *Thin Solid Films* 1999;343:183–6.
- [51] Papaefthimiou S, Leftheriotis G, Yianoulis P. Advanced electrochromic devices based on WO_3 thin films. *Electrochimica Acta* 2001;46(13-14):2145–50.
- [52] Papaefthimiou S, Leftheriotis G, Yianoulis P. Study of WO_3 films with textured surfaces for improved electrochromic performance. *Solid State Ionics* 2001;139(1-2):135–44.
- [53] Karlsson, J, et al. Control strategies and energy saving potentials for variable transmittance windows versus static windows. In: Proceedings of Eurosun, Copenhagen, Denmark, 2000.
- [54] Sullivan, R, et al. Effect of switching control strategies on the energy performance of electrochromic windows. 1994.
- [55] Sullivan, R, et al. The energy performance of electrochromic windows in heating-dominated geographic locations. 1996.
- [56] Sullivan R, Rubin M, Selkowitz S. Energy performance analysis of prototype electrochromic windows. CA (United States): Lawrence Berkeley National Lab.; 1996.
- [57] Bechinger C, Gregg B. Development of a new self-powered electrochromic device for light modulation without external power supply. *Solar Energy Materials and Solar Cells* 1998;54(1-4):405–10.
- [58] Benson, DK, et al. Stand-alone photovoltaic (PV) powered electrochromic window, 1995, Google Patents.
- [59] Huang LM, et al. Photovoltaic electrochromic device for solar cell module and self-powered smart glass applications. *Solar Energy Materials and Solar Cells* 2011.
- [60] Deb SK, et al. Stand-alone photovoltaic-powered electrochromic smart window. *Electrochimica Acta* 2001;46(13-14):2125–30.
- [61] Deb SK. Opportunities and challenges in science and technology of WO_3 for electrochromic and related applications. *Solar Energy Materials and Solar Cells* 2008;92(2):245–58.
- [62] Gao W, et al. First a-SiC \dot{a} : \dot{a} H photovoltaic-powered monolithic tandem electrochromic smart window device. *Solar Energy Materials and Solar Cells* 1999;59(3):243–54.
- [63] Gao W, et al. Approaches for large-area a-SiC:H photovoltaic-powered electrochromic window coatings. *Journal of Non-Crystalline Solids* 2000;266:1140–4.
- [64] Hauch A, et al. New photoelectrochromic device. *Electrochimica Acta* 2001;46(13-14):2131–6.
- [65] Pichot F, et al. Flexible solid-state photoelectrochromic windows. *Journal of the Electrochemical Society* 1999;146:4324.
- [66] Georg A, et al. Switchable glazing with a large dynamic range in total solar energy transmittance (TSET). *Solar Energy* 1998;62(3):215–28.
- [67] Doane, JW, G Chidichimo, NAP Vaz. Light modulating material comprising a liquid crystal dispersion in a plastic matrix, 1987, Google Patents.
- [68] Ferguson, JL. Encapsulated liquid crystal material, apparatus and method, 1992, Google Patents.
- [69] Niklasson GA, Granqvist CG. Electrochromics for smart windows: thin films of tungsten oxide and nickel oxide, and devices based on these. *Journal of Materials Chemistry* 2006;17(2):127–56.
- [70] Kanu SS, Binions R. Thin films for solar control applications. *Proceedings of the Royal Society of London Series A* 2010;466(2113):19.

- [71] Granqvist CG, et al. Progress in chromogenics: new results for electrochromic and thermochromic materials and devices. *Solar Energy Materials and Solar Cells* 2009;93(12):2032–9.
- [72] Mlyuka N, Niklasson G, Granqvist CG. Thermochromic multilayer films of VO₂ and TiO₂ with enhanced transmittance. *Solar Energy Materials and Solar Cells* 2009;93(9):1685–7.
- [73] L CM. Large-area smart glass and integrated photovoltaics. *Solar Energy Materials and Solar Cells* 2003;76(4):489–99.
- [74] Greenberg CB. Undoped and doped VO₂ films grown from VO (OC₂H₇)₃. *Thin Solid Films* 1983;110(1):73–82.
- [75] Parkin IP, Manning TD. Intelligent thermochromic windows. *Journal of Chemical Education* 2006;83(3):393.
- [76] Blackman CS, et al. Atmospheric pressure chemical vapour deposition of thermochromic tungsten doped vanadium dioxide thin films for use in architectural glazing. *Thin Solid Films* 2009;517(16):4565–70.
- [77] Kiria P, Hyettb G, Binions R. Solid state thermochromic materials. *Advances Material Letters* 2010;1(2):20.
- [78] Jin P, Tanemura S. Formation and thermochromism of VO₂ films deposited by RF magnetron sputtering at low substrate temperature. *Japanese Journal of Applied Physics Part 1* 1994;33:1478–1478.
- [79] Manning TD, Parkin IP. Atmospheric pressure chemical vapour deposition of tungsten doped vanadium (IV) oxide from VOCl₃, water and WC₁₆. *Journal of Materials Chemistry* 2004;14(16):2554–9.
- [80] Guinneton F, et al. Optimized infrared switching properties in thermochromic vanadium dioxide thin films: role of deposition process and microstructure. *Thin Solid Films* 2004;446(2):287–95.
- [81] McCluney R, Center FSE. Fenestration solar gain analysis. Citeseer 1996.
- [82] Lee MH, Cho JS. Better thermochromic glazing of windows with anti-reflection coating. *Thin Solid Films* 2000;365(1):5–6.
- [83] Morin F. Oxides which show a metal-to-insulator transition at the Neel temperature. *Physical Review Letters* 1959;3(1):34–6.
- [84] Phillips TE, Murphy RA, Poehler TO. Electrical studies of reactively sputtered Fe-doped VO₂ thin films. *Materials Research Bulletin* 1987;22(8):1113–23.
- [85] Bêteille F, et al. Switching properties of V_{1-x}Ti_xO₂ thin films deposited from alkoxides. *Materials Research Bulletin* 1997;32(8):1109–17.
- [86] MacChesney J, Guggenheim H. Growth and electrical properties of vanadium dioxide single crystals containing selected impurity ions. *Journal of Physics and Chemistry of Solids* 1969;30(2):225–34.
- [87] Nag J, Haglund Jr R. Synthesis of vanadium dioxide thin films and nanoparticles. *Journal of Physics: Condensed Matter* 2008;20:264016.
- [88] Xu G, et al. Thickness dependence of optical properties of VO₂ thin films epitaxially grown on sapphire (0 0 0 1). *Applied Surface Science* 2005;244(1–4):449–52.
- [89] Binions R, et al. Doped and un-doped vanadium dioxide thin films prepared by atmospheric pressure chemical vapour deposition from vanadyl acetylacetonate and tungsten hexachloride: the effects of thickness and crystallographic orientation on thermochromic properties. *Journal of Materials Chemistry* 2007;17(44):4652–60.
- [90] Pierce J, Goodenough J. Structure of orthorhombic V_{0.95}Cr_{0.05}O₂. *Physical Review B: Condensed Matter* 1972;5(10):4104.
- [91] Saeli M, et al. Energy modelling studies of thermochromic glazing. *Energy and Buildings* 2010;42(10):1666–73.
- [92] Evans P, et al. Multi-functional self-cleaning thermochromic films by atmospheric pressure chemical vapour deposition. *Journal of Photochemistry and Photobiology A: Chemistry* 2007;189(2–3):387–97.
- [93] Li S-Y, Niklasson GA, Granqvist C-G. Thermochromism of VO₂ nanoparticles: calculated optical properties and applications to energy efficient windows. *MRS proceedings*. Cambridge Univ Press; 2011.
- [94] Gao Y, et al. Nanoceramic VO₂ thermochromic smart glass: a review on progress in solution processing. *Nano Energy* 2012;1(2):221–46.
- [95] Zhou S, et al. Microstructures and thermochromic characteristics of low-cost vanadium–tungsten co-sputtered thin films. *Surface and Coatings Technology* 2012;206(11):2922–6.
- [96] Kivaisi R, Samiji M. Optical and electrical properties of vanadium dioxide films prepared under optimized RF sputtering conditions. *Solar Energy Materials and Solar Cells* 1999;57(2):141–52.
- [97] Binions, R, IP Parkin. Novel chemical vapour deposition routes to nanocomposite thin films. 2011.
- [98] Sobhan M, et al. Thermochromism of sputter deposited W_xO_{2-x} films. *Solar Energy Materials and Solar Cells* 1996;44(4):451–5.
- [99] Saitzek S, et al. New thermochromic bilayers for optical or electronic switching systems. *Thin Solid Films* 2004;449(1):166–72.
- [100] Saitzek S, et al. Thermochromic CeO_{2-x}VO_{2-x} bilayers: role of ceria coating in optical switching properties. *Optical Materials* 2007;30(3):407–15.
- [101] Speck K, et al. Vanadium dioxide films grown from vanadium tetra-isopropoxide by the sol–gel process. *Thin Solid Films* 1988;165(1):317–22.
- [102] Babulanan S, et al. Thermochromic VO_{2-x} films for energy-efficient windows. *Solar Energy Materials* 1987;16(5):347–63.
- [103] Evans P, et al. Multi-functional self-cleaning thermochromic films by atmospheric pressure chemical vapour deposition. *Journal of Photochemistry and Photobiology A: Chemistry* 2007;189(2):387–97.
- [104] Kang L, et al. Thermochromic properties and low emissivity of ZnO: Al/VO_{2-x} double-layered films with a lowered phase transition temperature. *Solar Energy Materials and Solar Cells* 2011;95(12):3189–94.
- [105] Kana J, et al. Thermochromic VO_{2-x} thin films synthesized by rf-inverted cylindrical magnetron sputtering. *Applied Surface Science* 2008;254(13):3959–63.
- [106] Lee M-H. Thermochromic glazing of windows with better luminous solar transmittance. *Solar Energy Materials and Solar Cells* 2002;71(4):537–40.
- [107] Chen B, et al. Al³⁺-doped vanadium dioxide thin films deposited by PLD. *Solar Energy Materials and Solar Cells* 2009;93(9):1550–4.
- [108] Shi J, et al. Preparation and thermochromic property of tungsten-doped vanadium dioxide particles. *Solar Energy Materials and Solar Cells* 2007;91(19):1856–62.
- [109] Karlessi T, et al. Development and testing of thermochromic coatings for buildings and urban structures. *Solar Energy* 2009;83(4):538–51.
- [110] Luo Z, et al. Microstructures and thermochromic properties of tungsten doped vanadium oxide film prepared by using VOX-W-VOX sandwich structure. *Materials Science and Engineering: B* 2011;176(9):762–6.
- [111] Sahana M, Dharmaparakash M, Shivashankar S. Microstructure and properties of VO₂ thin films deposited by MOCVD from vanadyl acetylacetonate. *Journal of Materials Chemistry* 2002;12(2):333–8.
- [112] Thought of Amherst. 2010 [cited 2013 2013-05-04]; Available from: (<http://www3.amherst.edu/~thoughts/contents/amberger-thermochromism.html>).
- [113] Bukleski M, Petruševski VM. Preparation and properties of a spectacular thermochromic solid. *Journal of Chemical Education* 2009;86(1):30.
- [114] Bamfield P, Hutchings MG. Chromic phenomena: technological applications of colour chemistry. Royal Society of Chemistry; 2010.
- [115] Ryabova L, Serbinov I, Darevsky A. Preparation and properties of pyrolysis of vanadium oxide films. *Journal of the Electrochemical Society* 1972;119:427.
- [116] MacChesney J, Potter J, Guggenheim H. Preparation and properties of vanadium dioxide films. *Journal of the Electrochemical Society* 1968;115:52.
- [117] Leroux C, Nihoul G, Van Tendeloo G. From VO_{2(B)} to VO_{2(R)}: theoretical structures of VO₂ polymorphs and in situ electron microscopy. *Physical Review B: Condensed Matter* 1998;57(9):5111.
- [118] Gutarra A, et al. Electrochromism of sputtered fluorinated titanium oxide thin films. *Applied Physics Letters* 1994;64(13):1604–6.
- [119] Babulanan S, et al. Thermochromic VO₂ films for energy-efficient windows. *Solar Energy Materials* 1987;16(5):347–63.
- [120] Xu G, et al. Optimization of antireflection coating for VO₂-based energy efficient window. *Solar Energy Materials and Solar Cells* 2004;83(1):29–37.
- [121] Burkhardt W, et al. Tungsten and fluorine co-doping of VO₂ films. *Thin Solid Films* 2002;402(1–2):226–31.
- [122] Manning TD, et al. Intelligent window coatings: atmospheric pressure chemical vapor deposition of tungsten-doped vanadium dioxide. *Chemistry of Materials* 2004;16(4):744–9.
- [123] Cho J-H, et al. Thermochromic characteristics of WO_{3-x} doped vanadium dioxide thin films prepared by sol–gel method. *Ceramics International* 2012;38:S589–93.
- [124] Binions R, et al. Hybrid aerosol assisted and atmospheric pressure CVD of gold doped vanadium dioxide. *Chemical Vapor Deposition* 2008;14(1 2):33–9.
- [125] Saeli M, et al. Nano-composite thermochromic thin films and their application in energy-efficient glazing. *Solar Energy Materials and Solar Cells* 2010;94(2):141–51.
- [126] Khan K, Niklasson G, Granqvist C. Optical properties at the metal insulator transition in thermochromic VO_{2-x}F_x thin films. *Journal of Applied Physics* 1988;64(6):3327–9.
- [127] Kiri P, et al. Fluorine doped vanadium dioxide thin films for smart windows. *Thin Solid Films* 2011;520(4):1363–6.
- [128] Warwick MEA, et al. Hybrid chemical vapour and nanoceramic aerosol assisted deposition for multifunctional nanocomposite thin films. *Thin Solid Films* 2011.
- [129] Saitzek S, et al. Thermochromic CeO₂–VO₂ bilayers: role of ceria coating in optical switching properties. *Optical Materials* 2007;30(3):407–15.
- [130] CHEN Z, et al. VO₂-based double-layered films for smart windows: optical design, all-solution preparation and improved properties. *Solar Energy Materials and Solar Cells* 2011;95(9):2677–84.
- [131] Kang L, et al. Pt/VO₂ double-layered films combining thermochromic properties with low emissivity. *Solar Energy Materials and Solar Cells* 2010;94(12):2078–84.
- [132] Chen HK, et al. The preparation and characterization of transparent nanosized thermochromic VO₂–SiO₂ films from the sol–gel process. *Journal of Non-Crystalline Solids* 2004;347(1–3):138–43.
- [133] Cho, J-H, et al. Thermochromic characteristics of WO₃-doped vanadium dioxide thin films prepared by sol–gel method. *Ceramics International*, (0).
- [134] Saeli M, et al. Templated growth of smart coatings: hybrid chemical vapour deposition of vanadyl acetylacetonate with tetraoctyl ammonium bromide. *Applied Surface Science* 2009;255(16):7291–5.
- [135] Livage J. Optical and electrical properties of vanadium oxides synthesized from alkoxides. *Coordination Chemistry Reviews* 1999;391–403190 1999:391–403.
- [136] Maruyama T, Ikuta Y. Vanadium dioxide thin films prepared by chemical vapour deposition from vanadium (III) acetylacetonate. *Journal of Materials Science* 1993;28(18):5073–8.
- [137] Parkin IP, et al. Thermochromic coatings for intelligent architectural glazing. *Journal of Nano Research* 2008;2:1–20.
- [138] Barrea D, et al. Vanadyl precursors used to modify the properties of vanadium oxide thin films obtained by chemical vapor deposition. *Journal of the Electrochemical Society* 1999;551146 1999:551.

- [139] Bramwell S, et al. Bulk magnetization of the heavy rare earth titanate pyrochlores—a series of model frustrated magnets. *Journal of Physics: Condensed Matter* 2000;12:483.
- [140] Saeli M, et al. Templated growth of smart nanocomposite thin films: hybrid aerosol assisted and atmospheric pressure chemical vapour deposition of vanadyl acetylacetonate, auric acid and tetraoctyl ammonium bromide. *Polyhedron* 2009;28(11):2233–9.
- [141] Lee MH, Kim MG, Song HK. Thermochromism of rapid thermal annealed VO₂ and Sn-doped VO₂ thin films. *Thin Solid Films* 1996;290:30–3.
- [142] Nygren M, Israelsson M. A DTA study of the semiconductor-metallic transition temperature in. *Materials Research Bulletin* 1969;4(12):881–6.
- [143] Jin P, Tanemura S. Relationship between transition temperature and x in V_{1-x}W_xO₂ films deposited by dual-target magnetron sputtering. *Japanese Journal of Applied Physics* 1995;34(5A):2459–60.
- [144] Takahashi I, Hibino M, Kudo T. Thermochromic properties of double-doped VO₂ thin films prepared by a wet coating method using polyvanadate-based sols containing W and Mo or W and Ti. *Japanese Journal of Applied Physics* 2001;139140 2001:1391.
- [145] Takahashi I, Hibino M, Kudo T. Thermochromic V_{1-x}W_xO₂ thin films prepared by a wet coating method using polyvanadate solutions. *Japanese Journal of Applied Physics* 1996;35:L438–40.
- [146] Beteille F, Livage J. Optical switching in VO₂ thin films. *Journal of Sol–Gel Science and Technology* 1998;13(1):915–21.
- [147] Livage J, et al. Sol–gel synthesis of oxide materials. *Acta Materialia* 1998;46(3):743–50.
- [148] Livage J, et al. Optical properties of sol–gel derived vanadium oxide films. *Journal of Sol–Gel Science and Technology* 1997;8(1):857–65.
- [149] Cui HN, et al. Thermochromic properties of vanadium oxide films prepared by dc reactive magnetron sputtering. *Thin Solid Films* 2008;516(7):1484–8.
- [150] Kiri P, Hyett G, Binions R. Solid state thermochromic materials. *Advanced Materials Letters* 2010;1(2):86–105.
- [151] Ye H, Meng X, Xu B. Theoretical discussions of perfect window, ideal near infrared solar spectrum regulating window and current thermochromic window. *Energy and Buildings* 2012:164–7249 2012:164–72.
- [152] Sobhan M, et al. Thermochromism of sputter deposited W_xV_{1-x}O₂ films. *Solar Energy Materials and Solar Cells* 1996;44(4):451–5.
- [153] Binions R, Piccirillo C, Parkin IP. Tungsten doped vanadium dioxide thin films prepared by atmospheric pressure chemical vapour deposition from vanadyl acetylacetonate and tungsten hexachloride. *Surface and Coatings Technology* 2007;201(22–23):9369–72.
- [154] Qureshi U, Manning TD, Parkin IP. Atmospheric pressure chemical vapour deposition of VO₂ and VO₂/TiO₂ films from the reaction of VOCl₃, TiCl₄ and water. *Journal of Materials Chemistry* 2004;14(7):1190–4.
- [155] Piccirillo C, Binions R, Parkin IP. Synthesis and functional properties of vanadium oxides: V₂O₃, VO₂, and V₂O₅ deposited on glass by aerosol assisted CVD. *Chemical Vapor Deposition* 2007;13(4):145–51.
- [156] Piccirillo C, Binions R, Parkin IP. Synthesis and characterisation of W-doped VO₂ by aerosol assisted chemical vapour deposition. *Thin Solid Films* 2008;516(8):1992–7.
- [157] Jorgenson GV, Lee JC. Doped vanadium oxide for optical switching films. *Solar Energy Materials* 1986;14(3–5):205–14.
- [158] Ghanashyam Krishna M, Debaugé Y, Bhattacharya A. X-ray photoelectron spectroscopy and spectral transmittance study of stoichiometry in sputtered vanadium oxide films. *Thin Solid Films* 1998;312(1–2):116–22.
- [159] Talledo A, Granqvist C. Infrared absorption in lithium-intercalated vanadium pentoxide films. *Journal of Physics D: Applied Physics* 1994;244527 1994:2445.
- [160] Xue-Jin W, et al. Surface oxidation of vanadium dioxide films prepared by radio frequency magnetron sputtering. *Chinese Physics B* 2008;17:3512.
- [161] Kana JBK, et al. Thermochromic VO₂ thin films synthesized by rf-inverted cylindrical magnetron sputtering. *Applied Surface Science* 2008;254(13):3959–63.
- [162] Vernardou D, Pemble ME, Sheel DW. Tungsten doped vanadium oxides prepared by direct liquid injection MOCVD. *Chemical Vapor Deposition* 2007;13(4):158–62.
- [163] Binions R, Carmalt CJ, Parkin IP. Aerosol-assisted chemical vapour deposition of sodium fluoride thin films. *Thin Solid Films* 2004;469:416–9.
- [164] Livage J. Vanadium pentoxide gels. *Chemistry of Materials* 1991;3(4):578–93.
- [165] Guzman G, et al. Electrical switching in VO₂ sol–gel films. *Journal of Materials Chemistry* 1996;6(3):505–6.
- [166] Galy J. A proposal for (B) VO₂ phase transition: a simple crystallographic slip. *Journal of Solid State Chemistry* 1999;148(2):224–8.
- [167] Jonsson A, Roos A. Evaluation of control strategies for different smart window combinations using computer simulations. *Solar Energy* 2010;84(1):1–9.
- [168] Miguet F, Groleau D. A daylight simulation tool for urban and architectural spaces—application to transmitted direct and diffuse light through glazing. *Building and Environment* 2002;37(8):833–43.
- [169] Reilly MS, et al. Modeling windows in DOE-2.1E. *Energy and Buildings* 1995;22(1):59–66.
- [170] Clarke J, Janak M, Ruysevelt P. Assessing the overall performance of advanced glazing systems. *Solar Energy* 1998;63(4):231–41.
- [171] Syrrakou E, Papaefthimiou S, Yianoulis P. Eco-efficiency evaluation of a smart window prototype. *Science of the Total Environment* 2006;359(1):267–82.
- [172] Dussault J-M, Gosselin L, Galstian T. Integration of smart windows into building design for reduction of yearly overall energy consumption and peak loads. *Solar Energy* 2012.
- [173] Gelin K, et al. Thermal emissivity of coated glazing—simulation versus measurements. *Optical Materials* 2005;27(4):705–12.
- [174] Jonsson A, Roos A, Jonson EK. The effect on transparency and light scattering of dip coated antireflection coatings on window glass and electrochromic foil. *Solar Energy Materials and Solar Cells* 2010;94(6):992–7.
- [175] Jonsson A, Roos A. Visual and energy performance of switchable windows with antireflection coatings. *Solar Energy* 2010;84(8):1370–5.
- [176] Saeli M, et al. Optimisation of thermochromic thin films on glass; design of intelligent windows. *Advances in Science and Technology* 2011;75:79–90.
- [177] Xu X, et al. Simulation and improvement of energy consumption on intelligent glasses in typical cities of China. *Science China Technological Sciences* 2012;55(7):1999–2005.