# **Study of the Durability of OPC versus GGBS Concrete on Exposure to Silage Effluent**

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Abstract: Ordinary portland cement (OPC) has been traditionally used in the construction of concrete silos in Ireland. However, the aggressive nature of the effluent produced by silage leads to severe degradation of the concrete. GGBS is a common addition to PC composites. It has been demonstrated that GGBS improves the general performance of PC concrete, decreasing chloride diffusion and chloride ion permeability, reducing creep and drying shrinkage, increasing sulfate resistance, enhancing ultimate compressive strength, and reducing heat of hydration and bleeding. It has also been suggested that GGBS may increase concrete durability in the aggressive environment of silos. In order to investigate this theory, a simulation study was carried out by immersing samples of mortars incorporating increasing amounts of GGBS in a silage effluent solution and a magnesium sulfate solution. Over the course of an experiment consisting of three, 28-day cycles of immersion in the silage effluent, the sample performance was evaluated by testing permeability, porosity, water absorption, capillary suction, compressive strength, and mass loss. According to the results obtained, the OPC samples suffered the highest rise in permeability and porosity, and the greatest loss in both mass and compressive strength. In addition, the durability of the mortars, when subjected to both salt crystallization and silage effluent cycles, increased with increasing amounts of GGBS. The significant rise in capillary suction, water absorption, and permeability over the course of the experiment indicates that the damage induced by the effluent is not as superficial as previously reported. Loss in mass and increase in permeability were found to be the most reliable indicators of corrosion, as they gave the most dramatic and uniform results.

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#### **Introduction**

Grass conserved as silage for winter feed is vital for livestock production. One of the storage systems most commonly used consists of unwilted silage in horizontal concrete silos. Last year, in the Republic of Ireland, 25 million tonnes of silage were produced. In turn, this silage produces about 140 l of effluent per tonne (Irish Department of Agriculture, Food, and Forestry 1985). Proper management and containment facilities for this effluent are vital, as it contains lactic and acetic acids, having an acidic pH ranging from 2.5 to 4.5 and a biochemical oxygen demand (pollution potential) of up to 200 times that of raw domestic sewage (Sangarapillai and Dumelow 1993). Pollution caused by silage effluent not only results in acute environmental damage and killing of aquatic life, but can also cause a loss of revenue for the fishing and tourism industries.

As well as having a high pollution potential, silage effluent is highly corrosive to PC concrete, the construction material of silos. Corrosion to concrete, along with high pressured cleaning, can

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adversely affect the durability of a silo, allowing uncontrolled escape of effluent. Furthermore, repairing silos costs the agricultural industry vast sums of money. In Ireland, silage silos are grant aided under the farm waste management scheme. In 2005, €5 million was spent on repairs to silage facilities in response to severe corrosion (Irish Dept. of Agriculture and Food Statistics Office 2005). Therefore, the development of a more durable concrete for use in silage facilities is of paramount importance.

Blastfurnace slag (BS) is a by-product of the steel industry. It results from the combination of iron ore with limestone flux and is obtained from the manufacture of pig iron in a blastfurnace. When BS is quenched by water, it forms a glassy material known as granulated blastfurnace slag (GBS). In contact with water, GBS possesses hydraulic properties. However, the rate of reaction is slow and needs alkalis and sulphates to activate. When mixed with portland cement (PC), as PC hydrates, it releases alkalis and sulfates, which serve as activators for the GGBS.

The final properties of GGBS concrete are determined by the pozzolanic efficiency and hydraulic activity of the GGBS, which control the amount of cement produced and the reactivity of the slag. These depend on the reactive glass content, the chemical and mineralogical composition, type of activator, and fineness of GGBS (Ganesh Babu and Sree Rama Kumar 2000). GGBS has the same constituents as portland cement, but in different amounts (Table 1). In general, the more basic slags are, the greater their hydraulic activity in the presence of alkalis. Swamy (1986) has suggested that at constant basicity, strength of concrete increases with the  $Al_2O_3$  content. Additionally, it has been observed by Frearson (1986) that hydraulic activity is enhanced with an increase in  $Al_2O_3$ , CaO, and MgO, while an increase in SiO<sub>2</sub> diminishes hydraulic activity. To ensure high alkalinity, without

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Table 1. Chemical Composition of GGBS and OPC [Adapted from Bakherev et al. (2001); Jianyong and Yan (2001); Higgins (2003)]

Chemical analysis %	Portland cement	<b>GGBS</b>
SiO <sub>2</sub>	20.10	34.35–35.04
$Al_2O_3$	4.15	11.80-15.26
Fe <sub>2</sub> O <sub>3</sub>	2.50	$0.29 - 1.40$
CaO	61.30	36.80 - 41.40
MgO	3.13	$6.13 - 9.10$
$K_2O$	0.39	0.39
Na <sub>2</sub> O	0.24	0.34
TiO <sub>2</sub>	0.24	0.42
$P_2O_5$	< 0.90	< 0.10
MnO		0.43
SO <sub>3</sub>	4.04	$0.05 - 2.43$

which slag would be hydraulically inactive, European Standard ENV 197-1:1992, recommend that the ratio of CaO+MgO to  $SiO<sub>2</sub>$  exceeds 1.

It has been demonstrated that GGBS improves the general performance of PC composites decreasing chloride diffusion and chloride ion permeability Luo et al. 2003, Yun Yeau and Kyum Kim 2005), reducing creep and drying shrinkage (Jianyong and Yan 2001), increasing sulfate resistance (Higgins 2003; Binici and Aksogan 2006), enhancing the ultimate compressive strength (Barnett et al. 2006), and reducing the heat of hydration and bleeding (Wainwright and Rey 2000). GGBS also improves concrete workability due to its high specific surface, marketed at  $375-435$  m<sup>3</sup>/kg with a fineness of approximately 460 Blaine  $(m^2/kg \text{ min})$ . This makes GGBS finer than PC (typically, PC is approximately  $300 \text{ m}^3/\text{kg}$ ). According to Swamy (1986), this leads to increased workability and a better performance in bleeding, setting times, and heat evolution.

The purpose of this paper is to determine whether PC composites incorporating GGBS are more durable than those made with PC alone under the action of the chemicals produced by silage and, hence, provide information to allow the specification of a more durable concrete mix for agricultural use in silos.

# **Background**

As aforementioned, a number of authors have studied the effect of GGBS on PC concrete concluding that GGBS enhances the general performance of PC composites improving workability, reducing creep and drying shrinkage, raising the ultimate compressive strength, and reducing bleeding and heat of hydration. In relation to durability, Kumar et al. (1987) proved that GGBS concrete is better at both resisting chloride ingress and alkali-silica reactions than PC concrete, while other authors proved that GGBS increases sulfate resistance Higgins 2003; Binici and Aksoğan 2006) and decreases the chloride diffusion coefficient and chloride ion permeability Luo et al. 2003; Yun Yeau and Kyum Kim 2005). However, silage produces lactic and acetic acids and, in relation to the durability of GGBS concrete on exposure to these, results in literature were contradictory. For example, De Belie et al. (1996) conclude that GGBS enhanced concrete possesses a higher resistance to acid attack than PC concrete, while Kleinlogel (1960) agrees with the above and states that acetic acid attacks PC concrete twice as fast as it does BS concrete. In addition, McCloskey et al. (1997) conclude that the inclusion of blastfurnace slag at 60% reduced corrosion of the concrete by silage



effluent by 18–37%. However, on the other hand, Hussey and Robson (1950) stated that the inclusion of blastfurnace slag to concrete subjected to lactic acid solutions, provides no appreciable improvement.

O'Donnell et al. (1995) studied surface effects of silage effluent on concrete using a system in which effluent constantly flowed over the samples. As measurements of deterioration, the thickness of each specimen was recorded and the variation of surface profile over time was monitored by measurements of mass, volume, direct path ultrasonic pulse velocity, and texture depth. According to these authors, the depth of the corroded layer was similar in the laboratory trials than on silo floors. However, based on their sorptivity test, the authors concluded that only the surface layers of the concrete were affected and, hence, the remaining concrete was capable of containing solutions and would not let effluent escape through permeation.

De Belie et al. (1996) demonstrated that concretes/mortars made using 100% PC degrade at a faster rate than those including GGBS on exposure to liquids containing lactic and acetic acid with a pH ranging between 2 and 5.5. The authors carried out resistance and permeability tests on a number of samples made with OPC, OPC without  $C_3A$ , fly ash cement, and BS cement. According to these authors, the cement type has an important influence on the corrosion of concrete by feed acids. In descending order, the groups that performed best in the solutions were: BS cement, fly ash cement, OPC, and OPC without  $C_3A$ . This study also noted that, although GGBS improved performance, the amount of GGBS used was insignificant once the percentage was over 35%.

Finally, McCloskey et al. (1997) analyzed several variables that affect the performance of concrete in silage silos including water/cement ratio, cement content, aggregate type and size, free lime content, curing regime, and aging of the concrete. They concluded that replacing PC with GGBS considerably improved the resistance of concrete to effluent and observed that weight loss was the most important factor in determining concrete corrosion.

# **Methodology**

In order to compare the durability OPC and GGBS mortars, samples of different mix proportions (see Table 2) were exposed to silage effluent. Three 28-day cycles of effluent exposure were repeated over a 4-month period. In order to determine the rate of deterioration following each cycle, the mortars were tested for permeability, capillary suction, water absorption, and mass loss. Durability test by salt crystallization and compressive tests were also carried out.

# *Mixing and Curing*

According to the Irish Dept. of Agriculture (1990), the resistance of concrete to acid attack by silage effluent improves with a high cement content and the use of the least amount of water to pro-

**Table 3.** Chemical Composition of the Synthetic Silage Effluent

Constituent	Amount $(g/kg)$
Lactic acid	15
Acetic Acid	5
Formalin (0.38% w/w formaldehyde)	3
<b>KOH</b>	3.67
NaOH	0.78
$Ca(OH)$ <sub>2</sub>	1.39
Mg(OH)	0.44

duce a stiff but workable mix, which is thoroughly compacted and subsequently cured as specified. In order to test the effect of increasing GGBS amounts, samples containing 100% OPC, 70% OPC/30% GGBS, and 50% OPC/50% GGBS were mixed for testing. Water was added to each mix according to the moisture content of the sand measured using a speedy test. Thirteen 100 mm cubes were made from each mix.

Sample preparation was carried out in accordance with BS (2000). The molds were filled and compacted using a vibrating table and covered with Hessian sacking to be left for 24 h to set. They were cured for 28 days according to the aforementioned standard and then released from the molds, labeled, and placed in a curing tank.

#### *Preparation of Silage Effluent*

A synthetic solution containing the key components of farmyard silage effluent was prepared according to O'Donnell et al. (1995). The composition of this solution is included in Table 3. The pH of the solution was adjusted to 4.0 by adding the hydroxides of cations  $(K^+, Na^+, Ca^{2+}, Mg^{2+})$  in the relative proportions specified by the aforementioned authors.

#### *Durability Test by Exposure to Silage Effluent*

As aforementioned, a 28-day cycle was repeated three times over a 4-month period. It was decided for the cycle to last for 28 days as, according to De Belie et al. (1996, 2000), silage effluent typically flows for a period of 28 days after ensiling crops and, after this time, the effluent flow reduces to a negligible amount. Four cubes of each mix were immersed in the synthetic effluent at the start of each cycle. In order to avoid blockages of salt crystals building up in the mortar pores, the mortars were put through three cycles of immersion in water for 12– 16 h during the cycling. However, as the acids reacted with the basic samples, the pH of the effluent rose and acetic and lactic acids were regularly added to the solution in order to keep the pH at 4.

# *Durability Test by Salt Crystallization Cycling*

In order to measure the resistance of ordinary portland cement (OPC) and GGBS mortar to weathering by salt crystallization, three cubes of each mix were subjected to sulfate attack in a magnesium sulfate solution. The test procedure was carried out in accordance with I.S. 5: Part 3.1 (IS 1990). The standards suggest five cycles; however, noticeable degradation was not observed in this time and ten cycles were deemed to be more suitable after which degradation was more obvious and significant results were observed. At the end of the ten cycles, the samples were immersed in water for 24 h in order to remove salt and their final weight measured.

# *Permeability Test*

Permeability determines the durability of concrete as the less permeable the concrete, the fewer destructive substances will be able to penetrate it. This property was measured with the autoclam test. A base ring was fixed onto the surface of the sample isolating a test area with a diameter of 50 or 75 mm. A constant pressure of 0.5 bars was applied to contribute to the rate of flow across this area. The flow of water into the specimen was recorded automatically, with a data collector attached to the autoclam, every minute for 15 min. The total volume of water penetrating the concrete was recorded in m<sup>3</sup> and plotted versus the square root of time, informing on the material's permeability. In most instances this should yield a straight line graph, the slope of which may be reported as a water permeability index (WPI) with units  $m^3/$  / min.

# *Capillary Test*

The capillary tests were performed in accordance with BS (1999). The purpose of this test is to determine how much fluid will enter into the mortar through suction forces created by the water molecules and their microconnections with pore walls. The more fluid able to enter the mortar through capillary action, the more susceptible it will be to attack by silage effluent. The area of the base of each cube was calculated. The samples were then placed on thin supports and submerged to a constant depth of  $3\pm 1$  mm. At time intervals, for a total of 45 min, the cubes were removed, blotted dry, and weighed. The water suction was calculated using the formula

$$
S = (W_s - W_d/A) \cdot 100
$$

where  $S =$  suction ( $g/cm<sup>2</sup> min$ );  $W<sub>s</sub> =$  final weight after submersion after 1 min (g);  $W_d$ =initial dry weight of cube (g); and  $A$ =area  $(cm<sup>2</sup>).$ 

#### *Water Absorption Test*

The water absorption was calculated in order to assess whether there was an increase in the mortar's pore space as a result of weathering by exposure to silage effluent. This test was carried out in accordance with the Spanish Standards UNE 67-027-84 (UNE 1984). The mortars were immersed in water, at atmospheric pressure until saturation. Water absorption was measured as a percentage of the saturated mass of the specimen using the following equation:

$$
W_a = (W_s - W_d/W_s) \cdot 100
$$

where  $W_a$ =water absorption;  $W_s$ =final weight after absorption; and  $W_d$ =initial dry weight.

# **Results**

# *Visual Inspection Following Exposure to Silage Effluent*

Samples incorporating GGBS visually fared better on exposure to silage effluent over the course of the experiment. Visual degradation was obvious on mortars made from 100% OPC, with corners and edges invariably disintegrating when pressurized. In contrast, samples containing 30% as well as those including 50% GGBS remained largely unaltered, except for slight rounding of corners



**Fig. 1.** Average WPI for GGBS and PC samples at each stage of the experiment. Stage 1—prior to immersion.

of the 30% GGBS samples. The visual results indicate that the higher the percentage of GGBS incorporated into the mix, the better the durability on exposure to silage effluent. See section: Mass Loss on Exposure to Silage Effluent below.

#### *Permeability*

All samples showed an increase in permeability over the course of the experiment (Fig. 1). Although the permeability of the GGBS samples increased, their values were much lower than those reached by the PC samples. It is also clear from Fig. 1 that the WPI decreases with increasing amounts of GGBS. It can also be noted that, at the second stage of the experiment (i.e., following completion of the first 28-day cycle), there is a dip in the WPI. This was probably due to the presence of salt crystals, formed by immersion in the effluent, blocking the pathway for water ingress, thus, providing a lower WPI. However, once the samples were washed and the salt cleared from the pores, the increase in the WPI for each preceding stage was significant and relatively linear. Clearly, the permeability increase rate is greater for the PC samples than for those containing GGBS. As expected, the reference samples (unexposed to silage effluent) remained at a nearly constant average WPIs of  $1.38 \times 10^{-6}$  m<sup>3</sup> (100% OPC),  $7.43 \times 10^{-7}$  m<sup>3</sup> (30% GGBS), and  $6.48 \times 10^{-7}$  m<sup>3</sup> (50% GGBS).

#### *Water Absorption*

The PC mortars initially absorbed more water than the GGBS samples. However, all samples showed a progressive increase in water absorption over the course of the experiment (Fig. 2). The water absorption of the reference samples not exposed to silage



**Fig. 2.** Water absorption of PC and GGBS samples. Stage 1—prior to immersion.

**Table 4.** Total Increase in Water Absorption as a Result of Exposure to Silage Effluent

Sample	Average increase in water absorbed $(\%)$
OPC only	18.65
30% GGBS	12.65
50% GGBS	8.26

effluent is reported as Stage 1 in Fig. 2. An increase of water absorption is directly related to an enhancement in pore space in the mortars, thus, the greater the increase of water absorption, the greater the weathering as a result of exposure to silage effluent. The water absorption results in Fig. 2 suggest that the amount of water absorbed by the GGBS samples is lower than that absorbed by the PC samples, and that the higher the GGBS content, the lower the amount of water absorbed. This is also evidenced in Table 4, which shows the total increase in water absorption between the first and last stage of the experiment as a percentage.

#### *Capillary Test*

The variation of capillary suction as a result of exposure to silage effluent is included in Fig. 3. The capillary suction of the reference samples not exposed to silage effluent is reported as Stage 1 in Fig. 3. In general, it can be assumed that the higher the amount of water suction exerted by the pores of a particular sample, the more susceptible it would be to degradation and, hence, the less durable it would be deemed. As can be seen from Fig. 3, the 100% PC specimens showed the greatest increase in water suction; however, there is a sharp increase in water suction for all samples over the duration of the experiment. At both Stages 3 and 4, it is clear that the higher the GGBS content, the lower the water suction. However, at Stage 1, the samples with higher suction are those with the greatest GGBS content. This may be due to the microstructure of GGBS paste, with smaller pores creating a higher suction force in a denser interface (Gao et al. 2005). This trend was not observed in the measurement of water absorption (on the contrary, the amount of water absorbed by the GGBS samples was lower than that absorbed by the PC samples), since water absorption is more dependent on total volume of pores rather than on the pore size distribution. Alternatively, the higher suction of the samples with the greatest GGBS content can also suggest that, due to a lower rate of hydration, discontinuity of pores has not yet occurred.



**Fig. 3.** Average capillary suction of PC and GGBS samples at each stage (28-day effluent immersion cycle). Stage 1-prior to immersion.



Fig. 4. Mass loss (g) as a result of exposure to silage effluent. Stage 1—prior to immersion.

# *Mass Loss on Exposure to Silage Effluent*

Mass loss gave a very clear illustration of the amount of degradation that occurred in the samples as a result of effluent exposure. There was a relatively steady decline in the masses of all samples. However, the 100% OPC mortars lost mass at a faster rate than those containing GGBS, and the 50% GGBS samples performed the best showing the lowest mass loss (Fig. 4). Table 5 includes the average percentage mass loss of each sample type. When compared to the previous tests, where the 50 and 30% GGBS samples behave similarly, the percentage mass lost by the 30% GGBS samples is disproportionately large when compared to that of the 50% GGBS samples. Figs. 5–8 illustrate material loss following exposure to silage effluent.

#### *Compressive Strength*

Mechanical strength is a good indicator of long-term concrete durability, and it is generally accepted that stronger concretes will have a longer life. As expected, at the end of the silage immersion experiment, mortar mixes containing GGBS performed better, showing a greater strength than those including 100% PC Table 6). The initial compressive strength of the samples prior to effluent exposure was 13.83; 19.76, and 24.79 N/mm2 for the OPC, 30% GGBS and 50% GGBS samples, respectively. This is probably due to the cements produced by GGBS. These cements have a greater proportion of strength-enhancing compounds and less lime, which contributes little to concrete strength, than those produced in the hydration of PC. The percentage loss in strength of the 100% PC samples was nearly twice as much as that of the 50% GGBS samples.

#### *Salt Crystallization Cycles*

Over the course of the experiment, surface degradation was slow and minute, thus, it was difficult to visually judge performance. However, at the end of the experiment, when all excess salt was washed away, it became clear that the OPC samples had degraded significantly more than their GGBS enhanced counterparts as the aggregate could be clearly seen protruding through the corroded cement paste. The increase of surface roughness of the 100%







**Fig. 5.** Typical mortar sample before immersion in silage effluent



**Fig. 6.** Typical 100% portland cement mortar sample after final stage of immersion



**Fig. 7.** Typical 30% GGBS mortar sample after final stage of immersion



**Fig. 8.** Typical 50% GGBS mortar sample after final stage of immersion

**Table 6.** Average Loss and Percentage Loss of Compressive Strength as a Result of Exposure to Silage Effluent



**Table 7.** Average Mass Loss as a Result of Ten Cycles of Immersion in MgSO4 Solution

Sample type	% average mass loss
OPC.	$3.5 - 2.8$
30% GGBS	$1.5 - 1.4$
50% GGBS	$1.2 - 0.8$

OPC samples was far greater than that of the 50 and 30% GGBS mortars. Very little surface degradation took place to the 50% GGBS samples.

The average mass loss following immersion in  $MgSO<sub>4</sub>$  solution is included in Table 7. In addition, the mass loss after each cycle of immersion in the  $MgSO<sub>4</sub>$  solution is reported in Fig. 9. The 100% OPC samples lost mass at a faster rate than those containing a percentage of GGBS, showing the greatest total mass loss. The 100% OPC samples showed an initial increase in mass, which was then followed by a steady decrease in mass. This was probably due to cystallization of salt from the solution during the drying phase. However, once these salt crystals dissolved, the decrease in mass accelerated, and the 100% OPC mortar degraded quite rapidly in comparison to the GGBS enhanced mixes. The mass of the 50% GGBS mortar stayed relatively unchanged until the seventh cycle and even then mass loss was very slow. Although the decrease in mass of the OPC mortar over the duration of the experiment was quite small  $(2.8-3.5\%)$ , it was, however, relatively large when compared to the percentage mass lost by the 50% GGBS cubes  $(0.8-1.2\%)$ . The results suggest that increasing the percentage of GGBS in the mix enhances performance on exposure to salt crystallization.

# **Discussion**

The low values of the properties related to the presence and movement of moisture (permeability, capillary suction, and water absorption) of the GGBS mixes tested are partially responsible for the high resistance of GGBS mixes to attack by chloride and sulfate. These low values agree with the lower chloride diffusion and chloride ion permeability of GGBS concrete reported by Luo et al. (2003) and Yun Yeau and Kyum Kim (2005) and the increased sulfate resistance reported by Higgins (2003) and Binici and Aksoğan (2006); and the enhanced resistance to chloride ingress and alkali-silica reactions reported by Kumar et al. (1987).

The ultimate reason for this reduced moisture transport ability probably lies in the strong microstructure of the interfacial aggregate/binder transition zone of GGBS, which is often an area of weakness in other composites. GGBS significantly decreases both the content and the size of  $Ca(OH)_2$  crystals in the aggregate-paste interface, which makes the microstructure of the transition zone aggregate/binder dense and strong Gao et al. 2005).



**Fig. 9.** Mass loss after each cycle of immersion in the  $MgSO<sub>4</sub>$  solution

In addition, as reported by the aforementioned authors, the mineral composition of GGBS is also partially responsible for the high durability of GGBS concrete. GGBS includes more  $SiO<sub>2</sub>$  and less CaO than PC forming different hydrates in the cement paste. Specifically, hydrated PC paste contains more calcium hydroxide  $[Ca(OH)<sub>2</sub>]$  and more aluminates than GGBS, and these are the cement components most vulnerable to acid and sulfate attack. For example, when exposed to sulfates and weak acids such as those found in silage effluent,  $[Ca(OH)_2]$  forms soluble calcium salt crystals, which are then washed away by any leachate present (De Belie et al. 1996). However, in GGBS composites, GGBS reacts with the  $Ca(OH)_2$  during the hydration process, thus, less  $Ca(OH)_2$  is available to react with any acids or sulfates. In addition, when tricalcium aluminate  $[C_3A]$  comes into contact with weak acids or sulfates present in silage effluent, it forms an expansive salt known as ettringite (Ecocem. 2005), which causes microcracks to appear in the concrete mass allowing the ingress of more harmful chemicals and thus accelerating the deterioration process. Therefore, with the introduction of GGBS, salt and acid attack should decrease.

This research agrees with conclusions by Barnett et al. (2006) stating that GGBS enhances the ultimate compressive strength of OPC. The results of this research also agree with those by Kleinlogel (1960), McCloskey et al. (1997), and De Belie et al. (1996 and 2000), concluding that GGBS enhanced concrete possesses a higher resistance to acid attack. However, O'Donnell et al. (1995) concluded that only the surface layers of the concrete were affected, and that the remaining concrete was capable of containing the effluent escape through permeation. The present study disagrees with this, as from the capillary, permeability, and water absorption tests, it was clear that fluid could easily find its way further than the first few millimeters through the mortar, especially in those samples made from 100% PC. This would suggest that pathways for fluid exist deep within the samples. However, the immersion method used in this study was far more aggressive than the flow method used by O'Donnell et al. (1995) as, by immersion, obvious effects were noticeable at the end of the second cycle whereas it took five cycles to see obvious effects using the flow method.

De Belie et al. (1996) evidenced that the rate of deterioration of the OPC samples was greater than that of the samples containing GGBS. However, these authors noted that the amount of GGBS did not determine durability once the percentage was over 35%. In contrast, the present work demonstrates that there were significant differences between the results provided by the 30 and 50% mixes in mass loss and water absorption. Nevertheless, in other properties such as capillarity and permeability, the difference in the results was not important, thus, indicating that the amount of GGBS does not significantly impact the transport properties of the mortars near the surface. McCloskey et al. (1997) agrees with the present study, as they observed a noticeable gap in mass lost between the two different percentages of GGBS. These authors deemed weight loss to be the most reliable indication of concrete corrosion. However, this research suggests that both mass loss and increase in permeability are the most reliable indications of corrosion, as they gave the most dramatic and uniform results.

# **Conclusion**

Concrete used to build silage silos undergoes severe chemical attack over its lifetime. This damage invariably costs a lot of money to repair and, hence, research is needed in order to find a more durable concrete in this environment.

This paper has evidenced that PC composites incorporating GGBS are more durable than those made with PC alone in aggressive environments under the action of acids and salts such as those produced by silage, and that durability increased with increasing amounts of GGBS. GGBS mixes showed the smallest progressive rise in permeability, water absorption, and capillary suction as a result of silage immersion and salt cycling, as well as the smallest mass and compressive strength loss. Therefore, a more durable concrete mix for agricultural use in silos can be specified by incorporating GGBS as a partial substitute for OPC.

The addition of GGBS will increase lifespan and decrease maintenance of PC concrete silos; however, other issues such as expansion of shuttering time due to slower hydraulic development and low heat of hydration of GGBS need to be considered before a new specification is made.

This research concludes that the low values of the properties related to the presence and movement of moisture of GGBS mixes, probably due to the dense and strong microstructure of the interfacial aggregate/binder transition zone are probably responsible for the high resistance of GGBS mixes to attack in aggressive environments such as silage pits. The mineral composition of GGBS cement paste (with less aluminates and portlandite than PC) probably contributes to this resistance. The deterioration of performance of all tested mixes (based on the measurements of water absorption and mass loss) suggests that attack by silage effluent is not as superficial as previously reported, and that silage effluent corrodes PC concrete deep, significantly increasing capillary suction as well as absorption and permeability. Finally, this paper concludes that mass loss and increase in permeability (as measured with the autoclam test) are the most reliable indicators of concrete corrosion as they gave the most dramatic and uniform results.

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## **References**

- Bakherev, T., Sanjayan, J. G., and Cheng, Y. (2001). "Resistance of alkali-activiated slag concrete to carbonation." *Cem. Concr. Res.*, 31(9), 1277-1283(7).
- Barnett, S. J., Soutsos, M. N., Millard, S. G., and Bungey, J. H. (2006). "Strength development of mortars containing ground granulated blastfurnace slag: Effect of curing temperature and determination of apparent activation energies." Cem. Concr. Res., 36(3), 434-440.
- Binici, H., and Aksoğan, O. (2006). "Sulfate resistance of plain and blended cement." Cem. Concr. Compos., 28(1), 39-46.
- BS. (1999). "Determination of water absorption by capillarity." *BS No*. *EN 1925*.
- BS. (2000). "Methods of testing mortars." *BS No. 4551*.
- De Belie, N., Lenehan, J. J., Braam, C. R., Svennerstedt, B., Richardson, M., and Sonck, B. (2000). "Durability of building materials and components in the agricultural environment. Part III: Concrete structures." *J. Agric. Eng. Res.*, 76, 3–16.
- De Belie, N., Verselder, H. J., De Blaere, B., Van Nieuwenburg, D., and Verschoore, R. (1996). "Influence of the cement type on the resistance of concrete to feed acids." *Cem. Concr. Res.*, 26(11), 1717-1725. Ecocem. (2005). (www.ecocem.ie).
- Ganesh Babu, K., and Sree Rama Kumar, V. (2000). "Efficiency of GGBS in concrete." *Cem. Concr. Res.*, 30(7), 1031-1036.
- Gao, J. M., Qian, C. X., Liu, B., and Wang, L. L. (2005). "ITZ microstructure of concrete containing GGBS." Cem. Concr. Res., 35(7), 1299–1304.
- Higgins, D. D. (2003). "Increased sulfate resistance of ggbs concrete in the presence of carbonate." Cem. Concr. Compos., 25(8), 913-919.
- Hussey, A. V., and Robson, T. D. (1950). "High alumina cement as a constructional material in the chemical industry." *Proc. Symp. Materials of Construction in the Chemical Industry*, Birmingham, 217– 222.
- Irish Dept. of Agriculture, Food, and Forestry. (1985). "Guidelines and recommendations on control of pollution from farmyard wastes."
- Irish Dept. of Agriculture, Food, and Forestry. (1990). "Minimum specification for concrete silage bases and concrete walled silos." *S.128*.
- Irish Dept. of Agriculture and Food Statistics Office. (2005). "Farm waste management scheme."
- Irish Standards (IS). (1990). "Method of test for the determination of the soundness of coarse aggregate." *Irish Standards 5, Part 3.1*.
- Jianyong, L., and Yan, Y. (2001). "A study on creep and drying shrinkage of high performance concrete." Cem. Concr. Res., 31(8), 1203-1206.
- Kleinlogel, A. (1960). The influence of different physical chemical ele*ments on concrete*, 1st Ed., Dunod, Paris.
- Kumar, A., Roy, D. M., and Higgins, D. D. (1987). "Diffusion through concrete." *Concrete*, 31–32.
- Luo, R., Cai, Y., Wang, C., and Huang, X. (2003). "Study of chloride binding and diffusion in GGBS concrete." Cem. Concr. Res., 33(1),  $1 - 7$
- McCloskey, A., Cleland, D. J., Frost, J. P., Rankin, G. I. B., and Wilson, J. W. (1997). "Improving the durability of concrete in silos for grass silage." *Proc., Concrete for a Sustainable Agriculture*, Stavenger, Norway, 154–162.
- O'Donnell, C., Dodd, V., O'Kiely, P., and Richardson, M. (1995). "A study of the effects of silage effluent on concrete: Part 1. Significance of concrete characteristics." *J. Agric. Eng. Res.*, 60, 83–97.
- Sangarapillai, V. G., and Dumelow, J. (1993). "Rapid assessment of the durability of concrete exposed to silage effluent." *Farm Build. Prog.*, 114, 15–17.
- Swamy, R. N. (1986). Cement replacement materials, 1st Ed., Surrey Univ. Press, Surrey, U.K.
- UNE. (1984). "Normas ladrillos: Determinación de la absorción de agua y ensayo de heladicidad." *No. 67-027-84*, Instituto Español de Normalización (AENOR), ed.
- Wainwright, P. J., and Rey, N. (2000). "The influence of ground granulated blastfurnace slag (GGBS) additions and time delay on the bleeding of concrete." Cem. Concr. Compos., 22(4), 253-257.
- Yun Yeau, K., and Kyum Kim, E. (2005). "Corrosion resistance of concrete with ground granulate blast-furnace slag." *Cem. Concr. Res.*, 35(7), 1391–1399.